सीनेट की 64^{वीं} बैठक का कार्यवृत्त MINUTES OF THE 64TH MEETING OF THE SENATE

19 अप्रैल 2016 19[™] APRIL 2016



भारतीय प्रौद्योगिकी संस्थान रूड़की रूड़की — 247 667 (भारत) INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ROORKEE – 247 667 (INDIA)

INDIAN INSTITUTE OF TECHNOLOGY, ROORKEE ROORKEE-247 667 (INDIA)



INDEX

Item No.	Particulars	Page(s)
64.1	11 जनवरी 2016 को आयोजित हुई सीनेट की 63वीं बैठक के कार्यवृत्त की पुष्टि किया जाना। To confirm the minutes of the 63 rd meeting of the Senate held on 11 th January 2016.	3
64.2	सीनेट द्वारा 11 जनवरी 2016 को आयोजित अपनी 63वीं बैठक में लिए गए निर्णयों के कियान्वयन हेतु की गई कार्रवाई की रिपोर्ट प्राप्त किया जाना। To receive a report on the actions taken to implement the decisions taken by the Senate in its 63 rd meeting held on 11 th January 2016.	4
64.3	पी.एच.डी. प्रोग्राम में कसल्टेन्सी प्रोजक्ट फैलो को प्रवेश दिए जाने के प्रस्ताव पर विचार करना। To consider the proposal of admitting consultancy project fellows in Ph.D. programme.	4
64.4	टाईडस केन्द्र में स्टार्ट—अप कंपनियों के इक्यूबेशन रखने वउले छात्रों हेतु शैक्षणिक नियमों व विनियमों के अनुमोदन एवं उन्हें सम्मिलित किए जाने के लिए अध्ययन अंतराल स्टडी ब्रेक के प्रावधान के संबंध में डा० संजीव मन्हास, संकाय प्रभारी, टाईडस के अनुरोध पर विचार करना। To consider the request of Mr. Sanjeev Manhas, Faculty-in-Charge, TIEDS regarding provisions of study break for approval and incorporation in Academic rules and regulations for students' having start-ups companies incubation in TIEDS Centre	4
64.5	नैनो टैक्नोलोजी केन्द्र में पी.एचडी. में प्रवेश के लिए न्यूनतम शैक्षिक योग्यता पर विचार करना। To consider the Minimum Educational Qualifications for admission to Ph.D. Programme of the Centre of Nanotechnology.	5

64.6	अधिसंख्य आधार पर सभी आईआईटी मे अंतर्राष्टीय छात्रों के लिए स्नातकोत्तर छात्रों की स्वीकृत संख्या का 20 प्रतिशत प्रवेश लेने की मानव संसधान विकास मंत्रालय द्वारा अग्रेषित प्रो0 देवांग वी0 खख्खर, निदेशक, आईआईटी बाम्बे की रपोर्ट पर विचार करना। To consider the report of Prof. Devang V. Khakhar, Director, IIT Bombay forwarded by MHRD to take admission upto 20% of sanctioned strength of postgraduate students for International students in all the IITs on supernumerary basis.	5
64.7	आगामी सत्र 2016—17 से मानविकी और सामाजिक विज्ञान विभाग द्वारा चलाए जाने वाले प्रथम वर्ष एम0एससी0 अर्थशास्त्र के पाठयक्रम और पाठयक्रम सामग्री पर विचार करना। To consider the syllabus and course content of 1st year M.Sc. Economics programme to be run by the Department of Humanities & Social Sciences from upcoming session 2016-17	б
64.8	शरद और स्प्रिगं सेमेस्टर 2016–17 सत्र के लिए शैक्षणिक कैलेंडर पर विचार करना। To consider the Academic Calendar for the Autumn and Spring Semester 2016-17 session.	6
64.9	विभागाध्यक्ष, भूकम्प अभियांत्रिकी विभाग द्वारा संस्तुत एक वैकल्पिक कोर्स के रूप मे सीईएन—642 (विश्लेषण और पुलों के डिजाईन) एम0टैक0 के स्टक्चरर में (स्टक्चरल और गतिशीलता) कोर्स को शामिल किए जाने को रिपोर्ट करना। To report the inclusion of subject CEN-642 (Analysis and Design of Bridges) in the structure of M.Tech. (Structural & Dynamics) as an Elective course as recommended by Head, Department of Earthquake Engineering.	7
64.10	विभागाध्यक्ष, वास्तुकला और योजना विभाग द्वारा संस्तुत एक वैकल्पिक कोर्स सीटीएन—611 (परिवहन प्रणाली के प्रबंधन) के रूप में शहरी एवं ग्रामीण योजना के स्टकचर में शमिल किए जाने को रिपोर्ट करना। To report the inclusion of subject CTN-611 (Management of Transport Systems) in the structure of Master of Urban and Rural Planning, as an Elective course as recommended by Head, Department of Architecture & Planning.	7
64.11	डिसरटेसन में न्यूनतम पासिगं ग्रेड में किए गए परिवर्तन को रिपोर्ट करना। To report the change in the minimum passing grade in Dissertation.	7

64.12	स्नातकोत्तर कार्यक्रम 2016 में प्रवेश के लिए किए गए परिवर्तनों को रिपोर्ट करना।	7-9
	To report the changes for admission to PG programmes 2016	
64.13	शैक्षिक दौरे के लिए छात्रों को वित्तिय सहायता प्रदान करने के प्रस्ताव को रिपोर्ट करना। To report the proposal for providing the financial help to students for Educational Tour.	9
64.14	दिनांक 10.03.2016 को आयोजित हुई सीनेट की कार्यकारी समिति के 52वी बैठक के कार्यवृत्त को रिपोर्ट करना। To report the of the minutes of the 52nd Meeting of the Executive Committee of the Senate held on 10.03.2016.	9
64.15	उन छात्रों को पी0 एच0डी0 उपाधि प्रदान करने पर विचार किया जाना, जिन्होंने विभिन्न पाठ्यक्रमों में 1 जनवरी 2016 से अब तक उपाधि प्राप्त किए जाने की अर्हता प्राप्त की है। To consider award of the Ph.D. Degrees to the students who have completed the requirements for the award of the Ph.D. Degree in various disciplines w.e.f. 1st January 2016 to till date.	9
64.16	भू विज्ञान विभाग द्वारा प्रस्तावित निम्नलिखित दो विभागीय एंच्छिक पाठयकम पर विचार करना : 1. ईएसन—456: फील्ड जियोलोजी फार आईएमटी (भूवैज्ञानिक प्रौद्योगिकी) 2. ईएसन—628: फील्ड जियोलोजी फार एमएससी (एप्लाईड जियोलोजी)	10
	To consider the syllabi of following two Departmental Electives proposed by the Department of Earth Sciences: (i) ESN-456: Field Geology for IMT (Geological Technology) (ii) ESN-628: Field Geology for M.Sc. (Applied Geology)	
64.17	विभागीय वैकल्पिक पाठयकम ईएसएन—581: एडवांसड सिसमीक प्रोस्पेक्टिग को 9वे समेस्टर से 7वे समेस्टर में स्थानानंतरण और ईएसएन— पेट्रोफिजिक्स और सिसमिक रोक करकटराईजेशन के नए पाठयकम के लिए भूविज्ञान विभाग की डीएफसी की सिफारिश पर विचार करना । To consider the recommendation of the DFC of Department of Earth Sciences for shifting Departmental Elective Course ESN-581: Advanced Seismic Prospecting from 9th semester to 7th semester and the syllabi of new course ESN-xxx: Petrophysics and Seismic Rock Characterization.	10

64.18	मानविकी और सामाजिक विज्ञान विभाग द्वारा प्रस्तावित निम्नलिखित एमएससी अर्थशास्त्र प्रोग्राम कोर कोर्स के पाठयकम पर विचार करना।	10
	To consider the syllabus of following programme core course of M.Sc. Economics proposed by the Department of Humanities & Social Sciences.	
64.19	ईटीई उत्तर लिपियों में पृष्ठों की संख्या को कम करने के लिए कागज बचाने के प्रस्ताव पर विचार करना।	11
:	To consider the proposal of reducing the number of pages in ETE answer scripts to save paper.	
64.20	रसायन विज्ञान विभाग द्वारा प्रस्तावित एमएससी (कैमिस्ट्री) के लिए परिवर्तित पाठयकम की संरचना और कोर्स के पाठयकम पर विचार करना।	, 11
	To consider the revised curriculum structure for M.Sc. (Chemistry) and syllabi of courses as proposed by the Department of Chemistry.	
64.21	एक वैकल्पिक विषय में छात्रों की न्युनतम संख्या को कम करने के प्रस्ताव पर विचार करना।	12
	To consider the proposal of reducing the minimum number of students in an elective.	,
64.22	गणित विभाग द्वारा प्रस्तावित प्री-पीएचडी के पाठयकमों पर विचार करना। To consider the syllabi of Pre-Ph.D. courses proposed by the Department of Mathematics.	12
64.23	इलेक्ट्रोनिक्स एवं संचार अभियांत्रिकी विभाग द्वारा प्रस्तावित ईसीएन-5— डिजिटल सिस्टम डिजाईन के पाठृयकम पर विचार करना। To consider the syllabus of ECN-5xx: Digital System Design proposed by the Department of Electronics and	12
	Communication Engineering.	
64.24	आईएफएम द्वारा दिए गए सुझाव के अनुसार बाहरी परीक्षक द्वारा नोन डिस्क्लोजर एग्रीमेंट लेकर पीएचडी थिसिस के गोपनीय बौद्विक सम्पदा के ओपन और क्लोजड हिस्से को चिन्हित करने पर विचार करना। To consider identifying "Open" and "Closed" portions of Ph.D. thesis having confidential intellectual property and taking "Non-Disclosure Agreement" from the	13
	external examiners as suggested in IFM.	10
64.25	पेपर की स्वीकृति के अभाव में पीएचडी के लिए अधिकतम अविष के अंत में थीसिस जमा करने पर विचार करना।	13
	To consider the submission of thesis at the end of the maximum duration for Ph.D. in the absence of acceptance of paper.	
64.26	आयुष से पोस्ट ग्रेजुएट स्तर के पाठयकम, बायोटेक्नोलोजी में पीएचडी प्रवेश के लिए आयुर्वेद में एमडी/एमएस को भी एक योग्य मानदंड को सम्मलित करने पर विचार करना।	13
	To consider including Post Graduate level course from AYUSH leading to MD/MS in Ayurveda also an eligible criteria for Ph.D. admission in Bio –Technology.	

64.27	प्रो० एम०पी० वार्ष्णेय, तत्कालीन थेामसन इंजिनियरिंग कालेज रूडकी से प्रथम पी०एच०डी० उपाधि प्राप्त करने वाले और थोमसन कालेज के अलुमनस ने रू० 1.11 करोड़ इंडोमेंट फंड स्थापित करने के लिए दिये और निदेशक महोदय ने इस आफर को स्वीकार किया । यह इंडोमेंट फंड विद्युत अभियांत्रिकी विभाग की एक प्रयोगशाला के लिए कुछ यंत्र प्राप्त करनें, निश्चित पी०एच०डी फैलोशीप और एक आनरिक लेक्चर के लिए प्रयोग किये जाने के बारे मे रिपोर्ट करना।	14
	To report that Prof. M.P. Varshney, an alumnus of erstwhile Thomason College of Engineering Roorkee and the first alumnus of Thomason College to obtain a Ph.D. degree has offered a sum of Rs. 1.11 crore to establish an endowment fund and the Director has accepted the offer. The endowment fund will be used to acquire some	
	equipment for one of the laboratories in Electrical Engineering Department, creation of Ph.D. fellowship and an honorific lecture.	
64.28	शैक्षणिक वर्ष 2016—17 से स्नातक कार्यक्रम के लिए टयूशन फीस रू० 2.00 लाख प्रति वर्ष किये जाने के सम्बन्ध में एक ई—मेल दिनांक 11.04.2016 जिसके साथ स्कैन किया हुआ पत्र सं० 24—2/2016 टीएस दिनांक 08.04. 2016 जो कि सभी आईआईटी के निदेशकों को सम्बोधित है को रिपौर्ट करना।	14
	To report the e-mail dated 11.04.2016 alongwith a scan copy of letter F.No. 24-2/2016 TS-1 dated 08.04.2016 addressed to the Director of all IITs regarding revision of tuition fee for the undergraduate programme to Rs. 2.00 lakhs a year from the Academic Year 2016-17.	
App. 'A'	आगामी सत्र 2016—17 से मानविकी और सामाजिक विज्ञान विभाग द्वारा चलाए जाने वाले प्रथम वर्ष एम0एससी0 अर्थशास्त्र के पाठयकम और पाठयकम सामग्री। Syllabus and course content of 1st year M.Sc. Economics programme to be run by the Department of Humanities & Social Sciences from upcoming session 2016-17	15-36
App. 'B'	शरद और स्प्रिगं सेमेस्टर 2016–17 सत्र के लिए शैक्षणिक कैलेंडर। Academic Calendar for the Autumn and Spring Semester 2016-17 session.	37-45

·	े रे क्यांकारी समिति की	46-49
App. 'C'	दिनांक 10.03.2016 को आयोजित हुई सीनेट की कार्यकारी समिति की	10 15
	52वी बैठक का कार्यवृत्त ।	
	Minutes of the 52 nd Meeting of the Executive	
	Committee of the Senate held on 10.03.2016	50-55
App. 'D'	उन पी0 एच0डी0 उपाधि प्रदान करने वाले छात्रों की सूची जिन्होंने विभिन्न	50-55
_ 1,1	पाठ्यक्रमों में 1 जनवरी 2016 से अब तक उपाधि प्राप्त किए जाने की	
	अर्हता प्राप्त की है।	
	List of the Ph.D. Degrees to the students who have	
	sompleted the requirements for the award of the Ph.D.	
	Degree in various disciplines w.e.f. 1st January 2016 to	
	till date.	
App. 'E'	भू विज्ञान विभाग द्वारा प्रस्तावित निम्नलिखित दो विभागीय एंच्छिक	56
App. E	। पाठयक्मः	
	— — कि. कि.के.की कार आर्टकमटी (भवैज्ञानिक	
	प्रौद्योगिकी)	
	2. ईएसन-628: फील्ड जियोलोजी फार एमएससी (एप्लाईड जियोलोजी)	
	Syllabi of following two Departmental Electives	
	proposed by the Department of Earth Sciences:	
	(i) ESN-456: Field Geology for IMT (Geological	
•	Technology)	
-	(ii) ESN-628: Field Geology for M.Sc. (Applied	
	Geology)	
App. 'F'	भृविज्ञान विभाग के ईएसएन–583 पैट्रोफिजिक्स और सिसमिक रोक	57-58
Tipp. 1	करकटराईजेशन के नए पाठयकम	
	Syllabi of new course ESN-583: Petrophysics and	
	Seismic Rock Characterization of Department of Earth	
	Coiences	
App. 'G'	गानिकी और सामाजिक विज्ञान विभाग द्वारा प्रस्तावित एमएससी अर्थशास्त्र	59-60
App. d	प्रोग्राम कोर कोर्स एचएसएन—513: पब्लिक पालिसी थ्योरी और प्रक्टिस	
	पाटराकम् ।	
	Syllabus of HSN-513: Public Policy Theory and Practice	
	programme core course of M.Sc. Economics proposed	
	by the Department of Hilmanities & Social Sciences.	
Ann 'U'		61-63
App. 'H'	अंगोधित कवर पेज ।	
	Modified cover pages both for End Term and Mid Term	
	Examinations and cover page for supplementary sheet	
(72	Examinations and cover page 151 प्रमुख्य (क्रेमिस्टी) के लिए	64-138
App. 'I'	रसायन विज्ञान विभाग द्वारा प्रस्तावित एमएससी (कैमिस्ट्री) के लिए	
	परिवर्तित पाठयकम की संरचना और कोर्स के पाठयकम।	
	Revised curriculum structure for M.Sc. (Chemistry) and	
	syllabi of courses as proposed by the Department of	
	Chemistry.	

App. 'J'	गणित विभाग द्वारा प्रस्तावित प्री-पीएचडी के पाठयकम। Syllabi of Pre-Ph.D. courses proposed by the Department of Mathematics.	139-146
App. 'K'	इलैक्ट्रोनिक्स एवं संचार अभियांत्रिकी विभाग द्वारा प्रस्तावित ईसीएन–578 डिजिटल सिस्टम डिजाईन के पाठ्यकम। Syllabus of ECN-578: Digital System Design proposed by the Department of Electronics and Communication Engineering.	147-148
App. 'L'	शैक्षणिक वर्ष 2016—17 से स्नातक कार्यक्रम के लिए टयूशन फीस रू० 2.00 लाख प्रति वर्ष किये जाने के सम्बन्ध में सभी आईआईटी के निदेशकों को सम्बोधित पत्र सं0 24—2/2016 टीएस दिनांक 08.04.2016 Letter F.No. 24-2/2016 TS-1 dated 08.04.2016 addressed to the Director of all IITs regarding revision of tuition fee for the undergraduate programme to Rs. 2.00 lakhs a year from the Academic Year 2016-17.	149

MEETING SECTION INDIAN INSTITUTE OF TECHNOLOGY ROORKEE



Minutes of the $64^{\rm th}$ Meeting of the Senate held on 19.04.2016 in the Senate Hall of the Institute.

Following were present:

1.	Prof. Vinod Kumar, Dy. Director	In the Chair in lieu of Director (medical leave)
2.	Prof. (Mrs.) Pushplata	(Architecture & Planning)
3.	Prof. V. Devdas	(Architecture & Planning)
4.	Prof. (Mrs.) Ila Gupta	(Architecture & Planning)
5.	Prof. R.P. Singh	(Biotechnology)
6.	Prof. Ramasare Prasad	(Biotechnology)
7.	Prof. Partha Roy	(Biotechnology)
8.	Prof. Vijay Kumar Agarwal	(Chemical Engineering)
9.	Prof. B. Prasad	(Chemical Engineering)
10.	Prof. C.B. Majumdar	(Chemical Engineering)
11.	Prof. R.N. Goyal	(Chemistry)
12.	Prof. Ravi Bhushan	(Chemistry)
13.	Prof. Anil Kumar	(Chemistry)
14.	Prof. (Mrs.) Mala Nath	(Chemistry)
15.	Prof. U.P. Singh	(Chemistry)
16.	Prof. K.C. Gupta	(Chemistry)
17.	Prof. S.S. Jain	(Civil Engineering)
18.	Prof. Deepak Kashyap	(Civil Engineering)
19.	Prof. C.S.P. Ojha	(Civil Engineering)
20.	Prof. S.K. Ghosh	(Civil Engineering)
21.	Prof. M. Parida	(Civil Engineering)
22.	Prof. Praveen Kumar	(Civil Engineering)
23.	Prof. N.K. Samadhiya	(Civil Engineering)
24.	Prof. Z. Ahmad	(Civil Engineering)
25.	Prof. Kamal Jain	(Civil Engineering)
26.	Prof. B.R. Gurjar	(Civil Engineering)
27.	Prof. Manoj Mishra	(Computer Science & Engineering)
28.	Prof. M.L. Sharma	(Earthquake Engineering)
29.	Prof. (Mrs.) Amita Sinvhal	(Earthquake Engineering)
30.	Prof. Yogendra Singh	(Earthquake Engineering)
31.	Prof. B.K. Maheshwari	(Earthquake Engineering)
32.	Prof. Pankaj Agrawal	(Earthquake Engineering)
33.	Prof. Manish Shrikhande	(Earthquake Engineering)



34. Prof. D.C. Srivastava (Earth Sciences) 35. Prof. G.J. Chakrapani (Earth Sciences) 36. Prof. Sandeep Singh (Earth Sciences) Prof. Anand Joshi 37. (Earth Sciences) 38. Prof. R. Krishnamurthi (Earth Sciences) 39. Prof. Pramod Agarwal (Electrical Engineering) 40. Prof. S.P. Srivastava (Electrical Engineering) 41. Prof. N.P. Padhy (Electrical Engineering) 42. Prof. Debashish Ghosh (Electronics & Communication Engg.) 43. Prof. D.S. Arva (Hydrology) 44. Prof. D.K. Nauriyal (Humanities & Social Sciences) 45. Prof. S.P. Singh (Humanities & Social Sciences) 46. Prof. Y.S. Negi (Paper Technology) 47. Prof. S.C. Sharma (Paper Technology) 48. Prof. (Mrs.) Rama Bhargava (Mathematics) 49. Prof. R.C. Mittal (Mathematics) 50. Prof. V.K. Kativar (Mathematics) 51. Prof. Kusum Deep (Mathematics) 52. Prof. Tanuja Srivastava (Mathematics) 53. Prof. N. Sukavanam (Mathematics) 54. Prof. S.P. Yadav (Mathematics) 55. Prof. Pradeep Kumar (Mechanical & Industrial Engg.) Prof. S.C. Sharma 56. (Mechanical & Industrial Engg.) 57. Prof. Dinesh Kumar (Mechanical & Industrial Engg.) 58. Prof. B.K. Gandhi (Mechanical & Industrial Engg.) 59. Prof. Ravi Kumar (Mechanical & Industrial Engg.) 60. Prof. S.K. Nath (Metallurgical & Materials Engg.) 61. Prof. Anjan Sil (Metallurgical & Materials Engg.) 62. Prof. B.S.S. Daniel (Metallurgical & Materials Engg.) 63. Prof. Rajesh Srivastava (Physics) 64. Prof. Vir Singh (Physics) 65. Prof. G.D. Varma (Physics) 66. Prof. M.L. Kansal (WRD&M) 67. Prof. R.P. Saini (Alternate Hydro Energy Centre) 68. Prof. Arun Kumar (Alternate Hydro Energy Centre) 69. Dr. R. Balasubramanian Institute Computer Centre 70. Dr. A.K. Sharma Associate Dean, Academic Studies 71. Dr. Inderdeep Singh Associate Dean of Students' Welfare (Discipline) 72.

Dr. Barjeev Tyagi, Associate Professor, Department of Electrical Engineering

73. Dr. Manish Mishra, Associate Professor, Department of Mech. & Ind. Engg.

74. Dr. Umesh Sharma, Associate Professor, Department of Civil Engineering

Dr. K.R. Justin Thomas, Associate Professor, Department of Chemistry 75.

Dr. Rajat Agarwal, Associate Professor, Department of Management Studies 76.

77. Mr. Prashant Garg, Registrar & Secretary, Senate



The Chairperson welcomed the members to the 64th Meeting of the Senate.

Before taking up the agenda, the Chairperson thanked the undermentioned outgoing Senate members and recorded its appreciation for their valuable contributions in the meetings of the Senate:

- 1. Prof. R.G.S. Sastri, Department of Earth Sciences
- 2. Prof. V.K.Nangia, Department of Management Studies.
- 3. Dr. P. Jeevanandam, Associate Dean of Students' Welfare (Foreign Students)

The Chairperson also welcomed the under-mentioned new member to the Senate and hoped for his valuable contribution and active participation in its functioning:

Dr. Z. Rahman, Head, Department of Management Studies

The Senate noted the communications received from the following members for not attending the current meeting:

- 1. Prof. Girishwar Misra, Vice-Chancellor, Mahatma Gandhi Antarrashtriya Hindi Vishwavidyalaya, Wardha.
- 2. Prof. P.K. Ghosh, Department of Metallurgical & Materials Engg.
- 3. Prof. M.P. Sharma, Alternate Hydro Energy Centre
- 4. Prof. M.R. Maurya, Department of Chemistry
- 5. Prof. Prof. K.S. Hariprasad, Department of Civil Engineering
- 6. Prof. Mahendra Singh, Department of Civil Engineering
- 7. Prof. (Mrs.) Shashi, Department of Chemical Engineering
- 8. Prof. Pradeep Bhargava, Department of Civil Engineering

The Agenda was then taken up:

Item No. 64.1: To confirm the minutes of the 63rd meeting of the Senate held on 11.01.2016.

As no comments were received, the minutes of the 63rd meeting of the Senate held on 11.01.2016 were confirmed. While confirming the minutes, the issue of Educational Tour was raised by a member. It was informed that as the Educational Tour was made part of the curriculum structure of undergraduate programme at 3rd Year level first time, some of the departments could not arrange the tours in time during session. After discussion, it was decided that such departments may award '1' grade to the students in the Spring Semester 2015-16 session and allowed to schedule their Educational Tour in the beginning of the next semester



to complete this requirement by 30th September 2016. The 'I' grade be converted to actual grade (AP/AF) once the tour is organized.

It was also decided that this has to be completed before MTE of the next semester.

Item No.64.2: To receive a report on the actions taken to implement the decisions taken by the Senate in its 63rd meeting held on 11.01.2016.

The Senate noted the actions taken on the resolutions of the 63rd meeting held on 11.01.2016. The issue of constitution of Oral Defence Committee was again raised by a member. After due discussion it was decided that there will be no change in the constitution of Oral Defence Committee as mentioned in the Ph.D. Ordinances and Regulations.

Item No. 64.3: To consider the proposal of admitting Consultancy Project Fellows in Ph.D. programme

Considered the proposal of admitting Consultancy Project Fellows in Ph.D. programme as recommended by the IRC and decided that fellows selected in a Consultancy Project shall not be admitted in the Ph.D. programme directly. However, they can take admission in Ph.D. programme, only if selected in regular selection alongwith MHRD candidates.

Further, it was reiterated that Project Fellows selected in Sponsored Projects by the extended selection committee (having two more members from the department) can be admitted in the Ph.D. programme subject to satisfying the conditions notified earlier.

Item No. 64.4: To consider the request of Mr. Sanjeev Manhas, Faculty-in-Charge, TIEDS regarding provisions of study break for approval and incorporation in Academic rules and regulations for students' having start-ups companies incubation in TIEDS centre.

Considered the recommendations of the IRC on the proposal of Faculty-in-Charge, TIEDS regarding provisions of study break for students having start-ups companies incubation in TIEDS Centre and decided that no study break shall be given to any research scholar as there will be conflict of interests of student and supervisor.

Item No. 64.5: To consider the Minimum Educational Qualifications for admission to Ph.D. Programme of the Centre of Nanotechnology.

Considered the recommendation of the IRC regarding minimum educational qualification as given below, for admission to Ph.D. programme of Centre of Nanotechnology and decided the same be approved.

Centre of Nanotechnology: B.E./B.Tech./B.Pharm from relevant disciplines and higher.

All other standards for admission as given in Ph.D. Ordinances & Regulations will be applicable.

Item No. 64.6: To consider the report of Prof. Devang V. Khakhar, Director, IIT Bombay forwarded by MHRD to take admission upto 20% of sanctioned strength of postgraduate students for International students in all the IITs on supernumerary basis.

Considered the recommendation of the IAPC to take admission of International students in postgraduate programmes as proposed by MHRD and decided that, in principle, the proposal is accepted with the following additional points:

- 1. The number of International students shall be on supernumerary basis subject to availability of accommodation in campus.
- 2. The admission in M.Tech./Ph.D. in Engineering Departments shall be done only through GATE.
- 3. The GATE cut-off for International students for M.Tech. programme shall be decided by the Admission Committee.
- 4. There shall be annual contingency fund for the international students in addition to fees.

The candidates be admitted only from the session 2017-18 onwards.



Item No. 64.7: To consider the syllabus and course content of 1st year M.Sc. Economics programme to be run by the Department of Humanities & Social Sciences from upcoming session 2016-17

As considered and recommended by the IAPC, the Senate decided that the syllabi of the following programme core courses of M.Sc. Economics proposed by the Department of Humanities & Social Sciences as given at **Appendix 'A'** be approved with minor modification as suggested by the members:

- 1. HSN-502: Microeconomics I
- 2. HSN-503: Macroeconomics I
- 3. HSN-504: Mathematics for Economists
- 4. HSN-505: Development Economics
- 5. HSN-506: Money, Banking and Financial Markets
- 6. HSN-507: Microeconomics II
- 7. HSN-508: Macroeconomics II
- 8. HSN-509: Basic Econometrics
- 9. HSN-510: International Economics
- 10. HSN-511: Environmental Economics
- 11. HSN-512: Advanced Econometrics

Item No. 64.8: To consider the Academic Calendar for the Autumn and Spring Semester 2016-17 session.

Considered the Academic Calendar for the Autumn and Spring Semesters of 2016-17 session as suggested by the committee and decided that the same be approved. The approved Academic Calendar is given in **Appendix 'B'**. After discussion, the following were also decided:

- (a) Only the following students' events shall be organized in an academic session:
- 1. Cultural Festival THOMSO
- 2. Technical Festival COGNIZANCE
- 3. Hobbies Club Annual Function SRISHTI
- 4. Annual Sports Meet SANGRAM
- (b) All other students' events shall be clubbed with any one of the above events.



Item No. 64.9: Reported the inclusion of subject CEN-642 (Analysis and Design of Bridges) in the structure of M.Tech. (Structural & Dynamics) as an Elective course as recommended by Head, Department of Earthquake Engineering.

The Senate accepted and noted the recommendation of Head, Department of Earthquake Engineering for inclusion of CEN-642: Analysis and Design of Bridges in the structure of M.Tech. (Structural & Dynamics) as an Elective course.

Item No. 64.10: Reported the inclusion of subject CTN-611 (Management of Transport Systems) in the structure of Master of Urban and Rural Planning, as an Elective course as recommended by Head, Department of Architecture & Planning.

The Senate accepted and noted the recommendation of Head, Department of Architecture & Planning for inclusion of CTN-611: Management of Transport Systems in the structure of Master of Urban and Rural Planning as an Elective course.

Item No. 64.11: Reported the change in the minimum passing grade in Dissertation.

The Senate accepted and noted that the minimum passing grade in Dissertation in the new system should be 'D+' in place of 'C+'.

Item No. 64.12: Report the changes for admission to PG programmes 2016

The Senate accepted and noted the following changes for admission to PG programmes 2016

1. The changes suggested by the Centre for Transportation Systems regarding distribution seat GATE discipline wise and recommended the same as under:

Academic Deptt/ Centre	Academic Programme		Main Gate Discipline(s)			Other GATE Disciplines						
& (Code)	Name	Code	GATE Discipline Code	GEN	OBC	SC	ST	GATE Discipline Code	GEN	OBC	SC	ST
CTRANS (TSC)	M.Tech. (Infrastructure Systems)	49	CE (7)	4	2	1	0	ME/PI/CH/EE/ EC/CS/AR (8)	4	2	1	1



2. The GATE disciplines suggested by the Botechnology Department for its new M.Tech.+Ph.D. programme to be started from 2016 and recommended the same as under:

Academic Deptt/ Centre & (Code)	- I				Other GATE Disciplines							
	Name	Code	GATE Discipline Code	GEN	OBC	SC	ST	GATE Discipline Code	GEN	OBC	SC	ST
Botechnology (BTD)	M.Tech. (Bioprocess Engineering)	50	CH/BT (12)	6	3	2	1	AG/XE(3)	1	1	1	0

3. Changes suggested by the Hydrology department is not accepted and it will remain the same as existing.

Academic Deptt/ Centre & (Code)	Academi Programa		Main G	Sate Discipline(s)				Other GATE Disciplines *				
	Name	Code	GATE Discipline Code	GEN	OBC	SC	ST	GATE Discipline Code	GEN	OBC	သွ	TS
Hydrology (HYD)	M.Tech. Hydrology	33	CE/AG (12)	6	3	2	1	ME/GG/XE/ PH/EY(3)	1	1	1	0

- * To be filled from main GATE discipline in case candidates are not available from other GATE disciplens.
- 4. M.Tech. Polymer Science & Engineering in the Department of Polymer & Process Engineering (SRE Campus) which was approved by BOG to be started from the year 2016 will not start this year as the syllabi are not yet prepared by the Department.
- 5. The sponsored candidates be admitted in the Deptt of Electronics and Communication Engineering this year as recommended by the Department.
- 6. The sponsored candidates will not be admitted in the Deptt of Computer & Science Engineering this year as recommended by the Department except Army/DRDO/QIP.
- 7. The candidates working in Institute/ University awarding PG degree itself are not eligible for admission as part-time or full-time candidate, if facilities are not available except QIP candidates.
- 8. The tentative schedule of the PG Admission was discussed and decided the following :-

IMPORTANT DATES	
GATE Result will be announced	March 19, 2016 (Saturday)
Online Application Process (Open)	March 21, 2016 (Monday)
Last Date for Applying Online	April 10, 2016 (Sunday)
Last date for deposit of fee	April 12, 2016 (Tuesday)
Last date for receiving of completed Application at PG Admission Office, IIT Roorkee	April 18, 2016 (Monday)
Last date for uploading of call letters for Interview/Written Test/Counselling	May 09, 2016 (Monday)
Medical Examination for Persons with different abilities (PD) Candidates by the Medical Board	June 07, 2016 (Tuesday)
Interview/Written Test	June 08, 2016 (Wednesday)
Announcement of merit list after interview/written test	June 09, 2016 (Thursday night)
Counselling for admission	June 09-11, 2016 (Thu-Sat)
Last date for Withdrawal of Admission Offered	June 24, 2016 (Friday)
Display of 1 st upgraded list including Waitlisted Candidates (on Institute Website)	June 27, 2016 (Monday)

Display of 2 nd upgraded list including Waitlisted Candidates (on Institute Website)	July 04, 2016 (Monday)
Date of Registration	July 14, 2016 (Thursday)
Classes Begin	July 18, 2016 (Monday)
Final Counselling cum Registration of the Waitlisted	July 22, 2016 (Friday)
Candidates in Merit for the vacant seats, if any	

Item No. 64.13: Reported the proposal for providing the financial help to students for Educational Tour.

The Senate accepted and noted the following norms for Educational Tour:

- (a) Travelling Allowance by bus/ train only upto maximum of 3rd AC.
- (b) Accommodation charges as per actual or subject to maximum of @ Rs. 125/- per day.
- (c) DA or Food bill @ Rs. 100/- per day.

The above rules are applicable to pending cases also.

Item No. 64.14: Reported the of the minutes of the 52nd Meeting of the Executive Committee of the Senate held on 10.03.2016.

The Senate accepted and noted the minutes of the $52^{\rm nd}$ meeting of the Executive Committee of the Senate held on 10.03.2016 as given at **Appendix 'C'**.

Item No. 64.15: To consider award of the Ph.D. Degrees to the students who have completed the requirements for the award of the Ph.D. Degree in various disciplines w.e.f. 1st January 2016 to till date.

The Senate decided that the award of the Ph.D. Degrees to the students who have completed the requirements for the award of the Ph.D. Degree in various disciplines w.e.f. 1st January 2016 to till date of this Senate meeting be approved. The list of the Ph.D. Degrees is given at **Appendix 'D'**.

It was also decided that henceforth, only the table indicating the details (name of candidate, supervisor(s), examiners, department and title of the thesis) shall be put



before the Senate for approval and not the vive-voce examination report.

Item No. 64.16: To consider the syllabi of following two Departmental Electives proposed by the Department of Earth Sciences:

- (i) ESN-456: Field Geology for IMT (Geological Technology)
- (ii) ESN-628: Field Geology for M.Sc. (Applied Geology)

As considered and recommended by the IAPC, the Senate decided that the proposal of Department of Earth Sciences to include Departmental Electives **ESN-628 Field Geology** for the students of IMT (Geological Technology) and M.Sc.(Applied Geology) for strengthening the field geology component and imparting advance training on structural analysis in field be approved. The syllabus of the said course is given at **Appendix 'E'**.

Item No. 64.17: To consider the recommendation of the DFC of Department of Earth Sciences for shifting Departmental Elective Course ESN-581: Advanced Seismic Prospecting from 9th semester to 7th semester and the syllabus of new course ESN-xxx: Petrophysics and Seismic Rock Characterization.

As considered and recommended by the IAPC, the Senate decided that the recommendation of Department of Earth Sciences for shifting Departmental Elective Course **ESN-581: Advanced Seismic Prospecting** from 9th semester to 7th semester be approved.

As considered and recommended by the IAPC, the Senate further decided that the syllabus of new elective course ESN-583: Petrophysics and Seismic Rock Characterization as proposed by the department as given at Appendix 'F' be approved.

Item No. 64.18: To consider the syllabus of HSN-513: Public Policy-Theory and Practice programme core course of M.Sc. Economics proposed by the Department of Humanities & Social Sciences:

As considered and recommended by the IAPC, the



Senate decided that the syllabus of Programme Core Course **HSN-513: Public Policy- Theory and Practice** for M.Sc. (Economics) as proposed by the Department of Humanities & Social Sciences be approved.

The Senate further decided that the syllabus as given in **Appendix 'G'** be approved.

Item No.64.19: To consider the proposal of reducing the number of pages in ETE answer scripts to save paper.

Considered the recommendations of IAPC of reducing the number of pages in ETE answer scripts to save paper and decided that the following be approved:

- (i) Number of pages of ETE answer script shall be reduced to 24.
- (ii) Extra sheets, as supplementary answer scripts (4 pages at a time), shall be provided to students as and when required.
- (iii) Cover page of answer script shall be modified to include the signature of student.
- (iv) Number of pages in MTE shall be kept same as 16 but extra sheets shall be provided as supplementary answer scripts.

The modified cover pages both for End Term and Mid Term Examinations and cover page for supplementary sheet are given in **Appendix 'H'**.

Item No. 64.20: To consider the revised curriculum structure for M.Sc. (Chemistry) and syllabi of courses as proposed by the Department of Chemistry.

Considered the revised curriculum structure for M.Sc. (Chemistry) as recommended by the IAPC in view of the phasing out of their Integrated M.Sc. (Chemistry) programme and decided that the same be approved. The Senate also considered the syllabi of the modified courses as recommended by IAPC and decided that the same be approved. The modified structure of M.Sc. (Chemistry) and the syllabi of all the courses are given in **Appendix '1'**



Item No. 64.21: To consider the proposal of reducing the minimum number of students in an elective.

Considered the recommendations of IAPC of reducing the minimum number of students in an elective so that Departments can run more number of electives to facilitate UG students to take extra electives for Departmental Honours Degree and decided that the minimum number of students required to run any Departmental Elective or Open Elective be reduced as below:

- (a) Minimum number of students required to run a Departmental Elective for UG shall be reduced to **5**.
- (b) Minimum number of students required to run an Open Elective shall be reduced to **15**.

Item No. 64.22: To consider the syllabi of Pre-Ph.D. courses proposed by the Department of Mathematics.

As considered and recommended by the IAPC, the Senate decided that the syllabi of the following Pre-PhD courses proposed by the Department of Mathematics be approved with minor specifications:

- (i) MAN-901: Selected topics in Analysis
- (ii) MAN-902: Advanced Numerical Analysis
- (iii) MAN-903: Theory of Differential Equations
- (i) MAN-904: Selected Topics in Algebra

The syllabi of the recommended courses are given in **Appendix 'J'**

Item No. 64.23: To consider the syllabus of ECN-5xx: Digital System Design proposed by the Department of Electronics and Communication Engineering.

As considered and recommended by the IAPC the Senate approved the replacement of existing ECN-574: Semiconductor Materials, Devices and Characterization of Autumn Semester for the students of M.Tech. (Microelectronics and VLSI) with the proposed course ECN-578: Digital System Design. This shall be applicable form the session 2016-17. The syllabus of the proposed course is given in Appendix 'K'

Item No. 64.24: To consider identifying "Open" and "Closed" portions of Ph.D. thesis having confidential intellectual property and taking "Non-Disclosure Agreement" from the external examiners as suggested in IFM.

Considered the issue of identifying "Open" and "Closed" portions of Ph.D. thesis having confidential intellectual property as suggested in IFM and decided that there is no need to identify "Open" and "Closed" portions of Ph.D. thesis having confidential intellectual property" as recommended by the IRC.

The Senate also considered the recommendations of the IRC of taking "Non-Disclosure Agreement" from the external examiners and decided that the same be approved. It was also decided that this agreement proforma shall be sent to examiner while taking his/her consent to act as examiner. This will be made effective from the beginning of the next session.

Item No. 64.25: To consider the submission of thesis at the end of the maximum duration for Ph.D. in the absence of acceptance of paper.

The Senate considered the recommendations of the IRC that if the research scholar has completed 4½ years and has finished his/her work as proposed at the time of candidacy and submitted papers in journals of repute but acceptance has not come, the candidate may be allowed to present presubmission seminar for possible submission and after detailed discussion decided that the matter be referred to the departments to deliberate on this issue and give suggestions. The then revised recommendations of IRC based on the suggestions of the Departments will be considered in the next meeting of the Senate.

Item No. 64.26: To consider including Post Graduate level course from AYUSH leading to MD/MS in Ayurveda also an eligible criteria for Ph.D. admission in Bio –Technology.

Considered the recommendations of the IRC of including Post Graduate level recognized courses leading to MD/MS in Ayurveda also an eligible criteria for Ph.D. admission in Bio -Technology and decided that the same be approved.

Item No. 64.27: Repoted that Prof. M.P. Varshney, an alumnus of erstwhile Thomason College of Engineering Roorkee and the first alumnus of Thomason College to obtain a Ph.D. degree has offered a sum of Rs. 1.11 crore to establish an endowment fund and the Director has accepted the offer. The endowment fund will be used to acquire some equipment for one of the laboratories in Electrical Engineering Department, creation of Ph.D. fellowship and an honorific lecture.

The Senate accepted and noted that Prof. M.P. Varshney, an alumnus of erstwhile Thomason College of Engineering Roorkee and the first alumnus of Thomason College to obtain a Ph.D. degree has offered a sum of Rs. 1.11 crore to establish an endowment fund and the Director has accepted the offer. The endowment fund will be used to acquire some equipment for one of the laboratories in Electrical Engineering Department, creation of Ph.D. fellowship and an honorific lecture.

UNDER ANY OTHER ITEMS:

Item No. 64.28: Reported the e-mail dated 11.04.2016 alongwith a scan copy of letter F.No. 24-2/2016 TS-1 dated 08.04.2016 addressed to the Director of all IITs regarding revision of tuition fee for the undergraduate programme to Rs. 2.00 lakhs a year from the Academic Year 2016-17.

The Senate accepted and noted the e-mail dated 11.04.2016 alongwith a scan copy of letter F.No. 24-2/2016 TS-1 dated 08.04.2016 addressed to the Director of all IITs regarding revision of tuition fee for the undergraduate programme to Rs. 2.00 lakhs a year from the Academic Year 2016-17(Appendix 'L').

The meeting ended with a vote of thanks to the Chair.



Appendix-B

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 502 Course Title: Microeconomics I

2. Contact Hours: L: 3 T: 1 P: 0

3. Examination Duration (Hrs): Theory: 3 Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

5. Credits: 4 6. Semester: **Autumn** 7. Subject Area: **PCC**

8. Pre-requisite: Nil.

9. Objective: To provide an understanding of consumer and producer behavior, concept of markets and competition, and tools of comparative statics and their application to price theory.

S. No.	Contents	Contact hours
1	Consumer Behaviour:	6
	Theories of consumer behavior, preference orderings, demand, duality theory, revealed preference aggregate demand	
2	Producer Behaviour:	8
	Technology and production sets, costs and production functions (for competitive firm),cost minimization, profit maximization & duality theoryonce more, aggregate supply	
3	Competitive Market:	8
	Equilibrium in a competitive market, stability and comparative static properties, general equilibrium and Pareto optimality - fundamental theorems of welfare economics, externalities and market failures	
4	Imperfect Markets:	6
	Imperfect competition and market structure, bilateral monopoly, single product monopoly, discriminating monopoly, durable goods monopoly, recycling, leasing versus selling & the Coase conjecture; Monopsony and monopolistic market structure	
5	Game Theory with Applications to Oligopoly Theory: Game Theory: Extensive and normal form representation of games, Nash equilibrium (in pure and mixed strategies), definition and existence; subgame perfection in dynamic games. Applications: strategic behaviour of firms in a market (Bertrand, Cournot and Stackelberg models) and entry deterrence.	8

6	Markets for Inputs:	6
	Price and employment of inputs, the firm's demand curve-one	
	variable input and several variable inputs, market demand curve	
	and supply curve, equilibrium price and employment of an input,	
	rent, imperfectly competitive output markets	
	Total	42

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Hal R. Varian, <i>Microeconomic Analysis</i> , W. W. Norton & Company, 3 rd Edition	2010
2	Andreu Mas-colell, Michael D. Whinston and Jerry R. Green John, <i>Microeconomic Theory</i> , Oxford University Press, 1 st Edition	2012
3	Geoffrey Alexander Jehle and Philip J. Reny, Advanced Microeconomic Theory, Financial Times Prentice Hall, 3 rd Revised Edition	2011
4	Hal R. Varian, <i>Intermediate Microeconomics with Calculus</i> , W. W. Norton & Company, International Student Edition	2014
5	H. Gravelle and R. Rees, <i>Microeconomics</i> , Pearson India, 3 rd Edition	2007
6	E. Mansfield and G. Yohe, <i>Microeconomics</i> , Viva-Norton, 11 th Edition	2010



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 503

Course Title: Macroeconomics I

2. Contact Hours:

L: 3

T: 1

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS:0

MTE: 25

ETE: 50

PRE:0

5. Credits: 4

6. Semester: Autumn

7.Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart knowledge of working of the whole economy: the determination of income, output, employment, savings, interest rates, inflation and macroeconomic policies.

S. No.	Contents	Contact hours
1	Introduction:	2
	Macroeconomic concepts and issues	
2	National Income Accounting:	7
	National income accounting structure, key concepts,	
	measurements and difficulties, and circular flow of income- for	
	closed and open economy; GNP and welfare, Inter-temporal and	
	international comparisons of national income, Social accounting	ľ
•	and its significance, Green GDP, Beyond GDP	
3	Income and Employment Theory:	6
	Aggregate Demand and aggregate supply; Schools in	
	Macroeconomics: Classical to Neo-Keynesian, Multiplier and	
	accelerator principles; Equilibrium in the product, labour and	
	money markets	
4	IS-LM Model:	5
-	Adjustment towards equilibrium, multiplier analysis, role of	
	government in terms of monetary and fiscal policy	
5	Consumption and Saving:	7 *
	Consumption under certainty- Life-Cycle and Permanent Income	
	Hypothesis; Consumption under uncertainty-The Modern	
	Approach, Random Walk Hypothesis; Consumption and Risky	
	Asset; Beyond the Permanent Income Hypothesis	



6	Inflation and Unemployment:	8
	Inflation- theories, measurement, causes, effects and social costs; Unemployment, the demand for and the supply of labour, Wage rigidity; Inflation-unemployment trade-off, Phillips curve, inflation targeting, Seignorage and Inflation Problems, Non-accelerating inflation rate of unemployment, Inflationary expectations- Adaptive and rational expectations hypothesis	
7	Investment Theories:	7
	Keynesian, Post-Keynesian, Neo-Keynesian (Jorgenson approach), Lags in investment function	
	Total	42

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Gregory Mankiw, Macroeconomics, Macmillan, 8th Edition	2013
2	Bernanke B.S., Olekalns, N. and H. R. Frank, <i>Principles of Macroeconomics</i> , McGraw-Hill, 4 th Edition	2014
3	Errol D'Souza, <i>Macro Economics</i> , Pearson Education, Pearson, 2 nd Edition.	2012
4	Richard T. Froyen, <i>Macroeconomics: Theories and Policies</i> , Pearson Education, Pearson, 10 th Edition.	2014



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 504 Course Title: Mathematics for Economists

2. Contact Hours: L: 2 T: 1 P: 0

3. Examination Duration (Hrs): Theory: 2 Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

5. Credits: 3 6. Semester: Autumn 7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To introduce the mathematical tools and techniques necessary for understanding economic theories and their real-life applications.

S. No.	Contents	Contact hours
1.	Introduction to Mathematical Economics:	2
	Economic models; Review of fundamental concepts- Real	
	number system, sets, relations and functions	
2.	Linear Models and Matrix Algebra:	6
	Solving univariate and Multivariate Systems of Linear Equations,	
	Basic matrix operations, Determinant; Applications to analysis of	
	Markets and National Income Accounting	
3.	Use of Differentiation in Economic Analysis:	5
	Partial & Total Derivatives; Tests of continuity and monotonicity of	
	a function; Average and Marginal concepts; Elasticity	
4.	Optimization and Equilibrium:	5
	Maxima and minima; Unconstrained optimization techniques;	
	Constrained optimization techniques - Lagrange Multiplier	
	Method; Application to consumer behavior, production and costs.	
5.	Integral Calculus and Dynamic Analysis:	5
	First and higher order ordinary differential equations;	
	Simultaneous differential equations-Dynamic Input-output	
	models; Integral calculus application to welfare analysis	
6.	Mathematical Programming:	5
	Simple linear programming models; Duality Theorem; Non-Linear	
-	Programming	
	Total	28



S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1.	Alpha C Chiang & Kevin Wainwright, <i>Fundamental Methods of Mathematical Economics</i> , McGraw Hill Education, 4 th Edition	2013
2.	Carl P. Simon, Mathematics for Economists, Viva Books	2010
3.	Sydsaeter, K., Hammond, P. J. and Strom, A., <i>Essential Mathematics for Economic Analysis</i> , Pearson, 4 th Edition	2012
4.	Sydsaeter, K., Hammond, P. J. and Strom, A., Further Mathematics for Economic Analysis, Pearson, 2 nd Edition	2008



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 505 Course Title: Development Economics

2. Contact Hours: L: 3 T: 1 P: 0

3. Examination Duration (Hrs): Theory: 3 Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

5. Credits: 4 6. Semester: Autumn 7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart knowledge of various development theories, strategies and issues.

S. No.	Contents	Contact Hours
4	Introduction:	4
1	Basic concepts of growth and development, inclusive growth, change in the meaning of development over a period of time, socio-economic development indicators	4
2	Growth, Poverty and Inequality: Uni- and multi-dimensional concepts of poverty, poverty indicators, poverty and vulnerability, inequality and development; Inverted-U Hypothesis; Measurement of inequality (Lorenz curve, Gini coefficient, Theil index); Inequality axioms	6
3	Theories of Economic Growth: Classical, Keynesian, post-Keynesian, Neo-Keynesian and Neo-Classical	7
4	Strategies of Economic Development: The Big Push theory of industrialization; the Rostow's Theory of Stages of Growth; the Balanced versus Unbalanced Growth	6
	theories; Economic Dualism.	
5	Institutions and Economic Development: Types of institutions- Formal and informal, government, private and non-governmental; Property rights, legal systems, rule of law; Contribution of institutions to economic growth and efficiency; Role of civil societies in economic development, The rights-based approach to development; social capital and development	7
6	Population and Economic Development: Role of population growth in the development process, Population growth and living standards, Low Level Equilibrium trap; Demographic dividend; Migration and development	6
7	Structural Adjustment and Development:	6

	The fall of the centrally planned economies; State versus Market; Effects of Structural Adjustment on economic growth; Structural adjustment and stabilization policies of IMF and World Bank		
***	Total	42	

S. No.	Author (s)/Title/ Publisher	Year of Publication/ Reprint
1	A.P. Thirlwall, <i>Growth and Development</i> , Palgrave Macmillan, 8 th Edition	2006
2	Michael P. Todaro and Stephen C. Smith, <i>Economic Development</i> , Prentice Hall, 11 th Edition	2011
3	K. Staudt, Managing Development, Sage Publications	1991
4	Subrata Ghatak, <i>Introduction to Development Economics</i> , Routledge, 4 th Edition	2003
5	Ray, Debraj, <i>Development Economics</i> , Princeton University Press	1998
6	Word Development Reports, World Bank	Annual



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 506

Course Title: Money, Banking, and Financial

Markets

2. Contact Hours:

L: 2

T: 1

P: 0

3. Examination Duration (Hrs): Theory 1: 2

Practical:0

4. Relative Weightage:

CWS: 25

PRS:0

MTE: 25

ETE: 50

PRE:0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To provide knowledge of economic principles that underlies the role of money

in the economy: the operation of banks and other financial institutions and

the impact of credit and monetary policies on the economy.

S. No.	Contents	Contact hours
1	Introduction: Understanding 'money'- Functions, Evolution and Measures of Money; Overview of financial system and markets	2
2	Demand for and Supply of Money: Money Demand function- Classical and Keynesian approaches to money demand, Quantity theory of money, Keynes' liquidity preference theory; Neutrality of money; Friedman's modern quantity theory of money, Baumol-Tobin Model; Money in IS-LM framework	5
3	Structure of Interest Rates: Behaviour and term structure of interest rates, Changes in Equilibrium Interest rate, Yield curve and Theories on Term Structure of Interest rates- Expectations Theory, Segmented Market, Liquidity Premium	5
4	Money and Capital Markets: Instruments, targets and functions, Banking Industry- structure, role and performance; NBFIs and NBFCs; Development Financial institutions, Financial reforms and policies	, 7
5	Advanced Topics in Banking Finance: Market micro-structure, Role of regulation and incentives, International standards	3
6	Central Banking and Monetary Policy: Role and functions of Central Bank; Credit policy, Monetary policy- tools, strategy and refects, Policy rules vs. discretion,	6



Optimal monetary policy			!
	Total	28	ì

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Frederic S. Mishkin, <i>Economics of Money, Banking, and Financial Markets</i> , Prentice Hall, 10 th Edition.	2012
, 2	Carl E. Walsh, <i>Monetary Theory and Policy</i> , MIT Press, 3 rd Edition.	2010
3	Bennett T. McCallum, <i>Monetary Economics: Theory and Policy</i> , Macmillan, Facsimile Edition	1989
4	L M Bhole, and Jitendra Mahakud, <i>Financial Institutions and Markets</i> , Tata McGraw-Hill Education, 5 th Edition	2009
5	Annual Reports, Reserve Bank of India	Annual



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 507

Course Title: Microeconomics II

2. Contact Hours:

L: 3

T: 1

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS:0

MTE: 25

ETE: 50

PRE:0

5. Credits: 4

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: Elementary knowledge of mathematics and intermediate microeconomics.

9. Objective: To develop understanding on decision making under uncertainty, advance game theory, role of information in economic analysis and current topics in microeconomic theory.

S. No.	Contents	Contact hours
. 1	Decision-making under Uncertainty: Decision making under risk, decision making under uncertainty, attitudes towards risk - applications of theories to understand lotteries and gambling, stochastic dominance, expected utility theorem, critiques of expected utility theory, measurement of risk aversion.	8
2	Game Theory under Incomplete Information: Static and Dynamic Games of Incomplete information - Concepts and applications Principal agent models - Hidden action and hidden information	8
3	Information Economics: Market Failure under adverse selection, signaling models, screening models (will address price discrimination by a monopolist when buyer type is "private" information), optimal contact design in a moral hazard (hidden action model), applications	9
4	Special Topics 1: (i) Mechanism design with money and without money (ii) Applications: Spectrum auctions, coal auctions, school choice, kidney exchange, etc.	9
5	Special Topics 2: (i) Network economics (ii) Law and Economics	8
	Total	42



S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Hal R. Varian, <i>Microeconomic Analysis</i> , W. W. Norton & Company, 3 rd Edition	2010
2	James Bergin, <i>Microeconomic theory: A concise course</i> , Oxford University Press, 1 st Edition	2005
3	Martin J Osborne, An Introduction to Game Theory, Oxford University Press	2003
4	Nicholas Barr, <i>Economics of the Welfare State</i> , Oxford University Press, 5 th Edition	2012
5	Sanjeev Goyal, Connections: An Introduction to the Economics of Networks, Princeton University Press,	2009
6	Robert D. Cooter and Thomas S. Ulen, <i>Law and Economics</i> , Addison-Wesley, New York, 6 th Edition	2011



NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 508

Course Title: Macroeconomics II

2. Contact Hours:

L: 3

T: 1

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage:

CWS: 25

PRS:0

WTE: 25

ETE: 50 PRE:0

5. Credits: 4

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: Basic understanding of macroeconomics

9. Objective: To provide knowledge to analyze and develop theories and models describing macroeconomic dynamics.

S. No.	Contents	Contact
		hours
1	Dynamic General Equilibrium Model:	7
	Consumption and savings optimization-households and firms, the	ļ
	Ramsey model, the Diamond model, Intergenerational issues and	
	simple neo-classical general equilibrium model	
2	Growth in Dynamic General Equilibrium:	9
	Growth theories- assumptions, the dynamics of the model and	
	impacts, Solow model, Speed of Convergence, Role of Human	
	Capital, Empirical Issues; The golden rule of accumulation;	
	Endogenous Growth Theory: AK model; Learning by Doing,	
	Further Extensions	
3	Real Business Cycles:	9
	Concept of economic fluctuations, Real business cycle approach-	
	RBC model, household behavior and the special case of the	
	model, Evaluation and accounting for Business cycle fluctuations,	
	implications	
4	Micro Foundations in Macroeconomics:	8
	Lucas critique, Stagger price adjustments - Caplin-Spulber model,	
	Neo-Keynesian Economics, Coordination failure models, real	
	non-Walrasian theories	
5	Macroeconomics and Open Economy Aspects:	9
	Market for foreign exchange, devaluation and depreciation, real	
	and nominal exchange rate, factors affecting exchange rate,	
	Mundell-Fleming model, fixed versus floating exchange rate, price	
	adjustment, role of fiscal and monetary policies under alternative	
	exchange rate regimes, purchasing power parity concept Total	42
L	TOLAI	42

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1.	David Romer, Advanced Macroeconomics, McGraw Hill, 4 th Edition	2011
2	Ragbhendra Jha, <i>Macroeconomics for Developing Countries</i> , Routledge, 2 nd Edition	2003
3	Olivier Blanchard, and David R. Johnson, <i>Macroeconomics</i> , Pearson; 6 th Edition	2012
4	Sala-I-Martin Xavier Barro Robert J., <i>Economic Growth</i> , The MIT Press; 2 nd Edition	2003



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 509

Course Title: Basic Econometrics

2. Contact Hours:

L: 3

T: 1

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0

MTE: 25

ETE: 50

PRE:0

5. Credits: 4

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: Basic knowledge of Mathematics

9. Objective: To provide knowledge of basic econometric tools and techniques useful for economic analysis.

10. Details of Course:

S. No.	o. Contents	
1	Introduction: Nature and meaning of econometrics, structure of economic data, approaches to econometric model building	2
2	Basic Statistical Concepts: Sample and sample space, Random variables and Probability distributions and their properties; Tests of hypothesis; Estimation techniques	
3	Simple and Multiple Linear Regression Models: Formulation, Estimation and Evaluation-Model with one and <i>k</i> Independent Variables, Assumptions of CLRM, estimation methods (OLS and ML methods), marginal effects, point and interval estimates, Gauss-Markov theorem (BLUE), Test of hypotheses - individual coefficients, testing model parameters	9
4	Functional Forms and Model Specification: Regression models with quadratic effects, log-transformation and non-linear equations; Test of restrictions imposed on the relationship of more parameters-Chow test, Wald test, LR test, Ramsay's RESET, Bera-Jarque test of normality of errors, Hausman specification test	7
5	Violations in CLRM: Heteroskedasticity, Multicollinearity and Autocorrelation- consequences, detection and remedies; specification error, omitted variables, errors in measurement; Aitken theorem and GLS method	10
6	Dummy Variables: Meaning and Uses of Dummy Variables; Dummy Variable Trap;	6



11. Suggested Books:

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Paul R. Krugman and Maurice Obstfeld, <i>International Economics: Theory and Policy</i> , Pearson Education, 8 th Edition	2008
2	Dominick Salvatore, <i>International Economics: Trade and Finance</i> , Wiley, 10 th Edition	2012
3	Jagdish N. Bhagwati, Arvind Panagariya and T N. Srinivasan, Lectures on International Trade, The MIT Press	1998
4	Bo Sodersten and Geoffrey Reed, <i>International Economics</i> , Palgrave Macmillan, 3 rd Edition	1994
5	Robert C. Feenstra, Advanced International Trade: Theory and Evidence, Princeton University Press, 2 nd Edition	2015



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 510

Course Title: International Economics

T: 1

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: **CWS**: 25

PRS:0

MTE: 25 ETE: 50

PRE:0

5. Credits: 4

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To develop understanding of international trade theories, strategies and policies in local and global context.

10. Details of Course:

S. No.	Contents	Contact
		hours
1	An Overview of the World Economy	3
	Causes of Trade and Gains From Trade, Trade and Growth,	
	Trade and Environment	
2	Theories of International Trade I:	7
	The Ricardian Model and Comparative Advantage; Factor	
	Endowments and Pattern of Trade and Specialization (The	
	Heckscher-Ohlin Model), Factor Price Equalization	
3	Theories of International Trade II:	8
	New Approaches to Trade Theory, The Distribution of Income	
	and Adjustments, Empirical testing of the Inter-Industry trade and	
	Technological progress in Inter-Industry Trade	
4	International Trade Policy:	8
٠	The Partial Equilibrium Analysis of Trade Policy, The General	•
	Equilibrium Analysis of Trade Policy, Instruments of Trade Policy:	
	Tariffs, Quotas, Most Favoured Nation and Other Policy	
	Instruments	
5	Strategic Trade Policy:	.8
	Strategic tariff interaction, Dumping and anti-dumping measures,	
	The theory of Custom Unions, WTO and international trade,	
	GATT, Trade Creation and Trade Diversion	<u> </u>
6	The Balance of Payments and Determinants of Exchange	8
	Rates:	
	Balance of Payments, Determination of Exchange Rate:	
	Purchasing Power Parity, Demand for and Supply of Foreign	
	Exchange and other theories; IMF & International monetary	
	arrangements	
	Total	42



11. Suggested Books:

S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Paul R. Krugman and Maurice Obstfeld, <i>International Economics: Theory and Policy</i> , Pearson Education, 8 th Edition	2008
2	Dominick Salvatore, <i>International Economics: Trade and Finance</i> , Wiley, 10 th Edition	2012
3	Jagdish N. Bhagwati, Arvind Panagariya and T N. Srinivasan, Lectures on International Trade, The MIT Press	1998
4	Bo Sodersten and Geoffrey Reed, <i>International Economics</i> , Palgrave Macmillan, 3 rd Edition	1994
5	Robert C. Feenstra, Advanced International Trade: Theory and Evidence, Princeton University Press, 2 nd Edition	2015



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 511 Course Title: Environmental Economics

2. Contact Hours: L: 3 T: 1 P: 0

3. Examination Duration (Hrs.): Theory: 3 Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

5. Credits: 4 6.Semester: Spring 7. Subject Area: PCC

8. Pre-requisite:Nil

9. Objective: To enrich the understanding of various issues related to the economy and environment and basic theories of environmental economics with greater emphasis on their application.

10. Details of Course:

S. No.		
1.	Introduction: Basic concepts and tools; scope and relevance; difference among environmental economics, ecological economics and natural resource economics; collective choices: Harden thesis of tragedy of commons, Prisoners' dilemma game and Olson concept of collective actions	
2.	Environment and Development Poverty, environment, and economic growth; Environmental Kuznets Curve and its empirical evidences; Ecosystem services and development; Ecological footprints	
3.	Efficiency, Market and Market Failure: Concept of efficiency, Pareto optimality, efficiency and competitive markets, efficiency in exchange of goods and bads, consumer and producer surplus; Market Failure: public and private bads, rivalry and excludability, externalities, market and optimum provision of public goods and bads, methods to correct market failure; Pigovian Fees	8
4.	Property Rights: The rights of polluter and the victim, The Coase theorem and its policy significance	4
5.	Environmental Valuation and Cost Benefit Analysis: Environmental values, Consumer demand for environmental goods, welfare effects of a price change, Methods of environmental valuation: revealed and stated preference methods (Hedonic, Household Production function, travel cost, and Contingent valuation methods); Issues in estimation of costs and benefits, pollution cost, benefits from controlling pollution, efficiency iggollution control, limitations of cost-	10



	Production function, travel cost, and Contingent valuation methods); Issues in estimation of costs and benefits, pollution cost, benefits from controlling pollution, efficiency in pollution control, limitations of cost-benefit analysis	
6.	Environmental Regulations and Policy: Rationale for regulations, basic regulatory instruments, issues and effects of environmental regulations; The Kyoto Protocol, CDM and Carbon Credits; Environmental Regulatory Framework and Environmental Policy in India	8
	Total	42

11. Suggested Books:

Sr. No.	Name of Books/Authors	Year of Publication/ Reprint
1	Charles D. Kolstad, <i>Environmental Economics</i> , Oxford university Press, First Indian Edition, New Delhi.	2006
2	Tom Tietenberg, <i>Environmental and Natural Resource Economics</i> , Addison Wesley Longman, Inc., 7 th Edition	2006
3	Steven C. Hackett, <i>Environmental and Natural Resources Economics: Theory, Policy, and the Sustainable Society,</i> Routledge, 4 th Edition	2010
4	Nick Hanley, <i>Introduction to Environmental Economics</i> , Oxford University Press, 2 nd Edition	2013
5	Daniel W. Bromley, <i>Handbook of Environmental Economics</i> , Blackwell Publishing, 1 st Edition	1995
6	Report of Ministry of Environment and Forests, Government of India, New Delhi	Annual



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: HSN - 512

Course Title: Advanced Econometrics

T: 1

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative Weightage:

CWS: 25

PRS:0

ETE: 50 MTE: 25

PRE:0

5. Credits: 4

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Knowledge of Basic Econometric Methods

9. Objective: To provide knowledge of advanced econometric techniques with statistical software applications.

10. Details of Course:

S. No.	Contents	Contact
-		hours
1	Qualitative and Limited Dependent Models:	7
	Linear probability model(LPM)- estimation and limitations, Logit	
	and Probit models, Censored and truncated regression, MLE	
	estimation-Tobit model, Applications	
2	Simultaneous Equation Systems:	11
•	Structural and reduced form models, simultaneous equation bias,	
	identification problem and estimation methods- 2SLS, 3SLS and	
	LIML; SUR procedure; Problem of Endogeneity- causes and	
	solution; Instrumental Variables estimation; GMM- estimation and	
3	testing for overidentification Dynamic Econometric Models:	8
J	Autoregressive and Distributed lag (ADL) relationships-	U
	specification and problems in OLS estimation, Koyck's and	
	Almon's approach to DLM, partial adjustment and adaptive	
•	expectations method, Exogeneity tests, Wu-Hausman test	
4	Time Series Analysis:	9
-	Stationary and non-stationary time series, unit root tests; Concept of	
	cointegration, two-variable model – Engle-Granger method and	
	ECM; system of equations – vector autoregression (VAR),	
	Johansen procedure, VECM, Granger causality	
5	Analysis of Panel Data: Balanced vs. unbalanced panel, Fixed	7
	effects model, random effects model, fixed vs. random effects- Wu	
	Hausman test, dynamic panel model	
	Total	42



11. Suggested Books:

S.No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
1	Greene, W.H., <i>Econometric Analysis</i> , Pearson, 7 th Edition	2012
2	Verbeek, M., <i>Guide to Modern Econometrics</i> , Wiley, 3 rd Edition	2008
3	A. Colin Cameron and Pravin K. Trivedi, <i>Microeconometrics</i> using Stata, Stata Press, 2 nd Edition	2010
4	Badi H. Baltagi, <i>Econometric analysis of Panel Data</i> , Wiley, 5 th Edition	2013
5	Walter Enders, <i>Applied Econometric Time Series,</i> Wiley, 3 rd Edition	2013



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ACADEMIC CALENDAR FOR THE YEAR 2016-17 (Autumn Semester)

S.No.	Details	Autumn Semester		
		Date	Day	
1.	Reporting and Registration of New Ph.D. students.	08.07.2016	Friday	
2.	Institute reopens and Registration of all new PG students	14.07.2016	Thursday	
3.	Registration of all existing students in the Departments/ Centres	15.07.2016	Friday	
4.	Re-examination and Second examination on medical	14.07.2016 -	Thursday	
	grounds (for Spring Semester 2015-16)	16.07.2016	Saturday	
5.	Commencement of Classes for Autumn Semester (2016-	18.07.2016	Monday	
	17) except UG I Yr	-		
6.	Reporting and Registrations of all new UG/IMT/IMS students	21.07.2016	Thursday	
7.	Orientation programme for all new students including	22.07.2016-	Friday -	
	selections of all newly admitted UG/IMT/IMS students for	24.07.2016	Sunday	
	N.C.C./N.S.S./N.S.O. and Language Proficiency test			
8.	Registration/Counselling for vacant seats of all PG programmes	22.07.2016	Friday	
9.	Closing of admissions	22.07.2016	Friday	
10.	Last date for sending the grades of Re-examination	22.07.2016	Friday	
11.	Commencement of Classes for UG I Yr	25.07.2016	Monday	
12.	Online subject registration of all new students including	30.07.2016-	Saturday	
	proficiency registration	03.08.2016	Wednesda	
13.	Last date of Academic Registration	01.08.2016	Monday	
14.	Uploading of roll lists of registered students	02.08.2016	Tuesday	
15.	Last date of addition/deletion of courses	03.08.2016	Wednesda	
16.	Uploading of final course-wise roll lists of registered students including proficiency	06.08.2016	Saturday	
17.	Assignment of Major projects to all B.Tech. final year students	12.08.2016	Friday	
	Independence Day	15.08.2016	Monday	
	Janmashtami	25.08.2016	Thursday	
18.	Request to Departments to send list of Institute Elective/	09.09.2016	Friday	
	Open Elective/ Departmental Elective (both UG and PG)			
	courses to be offered in Spring Semester 2016-17			
19.	Notification to students regarding shortage of attendance	09.09.2016	Friday	
	by the Departments upto 08.09.2016		-	
	ld-ul-Zuha (Bakrid)*	12.09.2016	Monday	
20.	Mid Term Examination (MTE) for all students	13.09.2016-	Tuesday-	
	L	16.09.2016	Friday	

S.No.	Details	Autumn Semester		
A'140'	Details	Date	Day	
21.	Last date to receive Institute Elective/ Open Elective / Departmental Elective (UG-PG) courses to be offered in Spring Semester from the departments	16.09.2016	Friday	
22.	Notification to UG/IDD/IMT/IMS students about Institute Elective/ Open Elective/ Departmental Elective (UG-PG) courses to be offered in Spring semester 2016-17	22.09.2016	Thursday	
23.	Last date for withdrawal of courses	23.09.2016	Friday	
24.	Last date for requesting Second Examination	23.09.2016	Friday	
25.	Intimation to parents/guardians in respect of students having short attendance & upload on website	23.09.2016	Friday	
26.	Annual Convocation 2016	24.09.2016	Saturday	
27.	Submission of remaining document(s) by all new students	30.09.2016	Friday	
28.	Online subject registration for Institute Elective/ Open Elective/ Departmental Elective (UG-PG) courses by students for next semester	30.09.2016- 04.10.2016	Friday - Tuesday	
	Mahatma Gandhi's Birthday	02.10.2016	Sunday	
29.	Last date to display allotted list of Institute Electives/ Open Electives/ Departmental Electives to students for Spring Semester 2016-17	08.10.2016	Saturday	
•	Dusshera (Vijaya Dashmi)	11.10.2016	Tuesday	
	Muharram*	12.10.2016	Wednesday	
30.	Last date for Finalization and Display of Time Tables for Spring Semester 2016-17 by all Departments and sending to Academic Section	21.10.2016	Friday	
	THOMSO 2016	21.10.2016	Friday-	
		23.10.2016	Sunday	
31.	Notification of End Term Examination schedule including seating plan (Institute Core and Elective Courses)	28.10.2016	Friday	
	Diwali (Deepawali)	30.10.2016	Sunday	
· .	Govardhan Puja	31.10.2016	Monday	
	Bhai Duj	01.11.2016	Tuesday	
32.	Ph.D. Interview	02.11.2016-	Wednesday-	
22	Online filling of Response Forms and Subject Registration	03.11.2016	Thursday	
33.	Online filling of Response Forms and Subject Registration for next semester by all students	02.11.2016- 07.11.2016	Wednesday - Monday	
34.	Evaluation of Final Year M.Tech./M.Arch./M.U.R.P.	02.11.2016-	Wednesday-	
J . 7.	M.Tech.(ES) /IDD/IMT Dissertation	21.11.2016	Monday	
35.	Communication from Chairman, DAPC to Course Coordinators requesting to submit the final list of students having short attendance	03.11.2016	Thursday	
36.	Online Application for Change of Branch during 2016-17 session by 1 st year B. Tech./IMT/IMS students	03.11.2016- 10.11.2016	Thursday - Thursday	



S.No.	Details	Autumn	Autumn Semester		
O.110.		Date	Day		
37.	Display of list of students having short attendance upto 07.11.2016 by the Departments/Centres and to send the Final list to Academic Section	08.11.2016	Tuesday		
38.	Last date of Teaching	09.11.2016	Wednesday		
39.	Notification of detained students due to shortage of attendance in End Term Exam by the Academic Section	09.11.2016	Wednesday		
	Guru Nanak Birthday	14.11.2016	Monday		
40.	End Term Examination (excluding Sunday) Practical examinations, if any, may be held during last few laboratory days).	11.11.2016- 22.11.2016	Friday- Tuesday		
41.	Notification to UG students regarding filling of Departmental Honour and Minor Specialization Courses	15.11.2016	Tuesday		
42.	B.Tech. Project Evaluation Stage-1	23.11.2016 25.11.2016	Wednesday -Friday		
43.	On-line filling of choices for DHC and MSC	23.11.2016- 26.11.2016	Tuesday- Friday		
44.	Last date of showing End Term Examination Answer Scripts to students	28.11.2015	Monday		
45.	Finalization of grades by the Grade Moderation Committee	30.11.2016	Wednesday		
46.	Display of grades for all courses by the Departments	30.11.2016	Wednesday		
47.	Last date of sending grades to Academic Section	30.11.2016	Wednesday		
48.	Last date to accept requests for Summer Internship at IITR	30.11.2016	Wednesday		
49.	Last date to contact departments/centres for grade modification, if any, by students	02.12.2016	Friday		
50.	Last date for sending modified grades to academic section	05.12.2016	Monday		
51.	Winter vacation for students (except for M.Tech/ IDD final year and Ph.D. students)	03.12.2016- 02.01.2017	Saturday - Monday		
52.	Winter vacation for Teaching Faculty (Faculty members can avail total 65 days of vacation during the winter & summer breaks)	05.12.2016- 02.01.2017	Monday - Monday		
53.	Last date for applying for Re-Examination	09.12.2016	Friday		
54.	Last date of preparation of Grade sheets	12.12.2016	Monday		
	ld-e-Milad*	13.12.2016	Tuesday		
55.	Notification regarding allotment DHC and MSC	17.12.2016	Friday		
56.	Finalization of Change of Branch of 1 st year B.Tech./ IMT/ IMS students for the session 2016-17	21.12.2016	Wednesday		
	Christmas Day	25.12.2016	Sunday		
57.	Reporting and Registration of new Ph.D. students	26.12.2016	Monday		
58.	Last date to send the list of recommended candidates for Summer Internship at IITR	30.12.2016	Friday		
59.	Institute Reopens for Spring Semester	03.01.2017	Tuesday		
60.	Registration of all existing stude of in the Departments/ Centres	03.01.2017	Tuesday		



<u> </u>		Autumn	Semester
S.No.	Details	Date	Day
61.	Re-examination and Second examination on medical ground (for Autumn Semester 2016-17)	04.01.2017- 06.01.2017	Wednesday -Friday
62.	Commencement of Classes for all students for Spring	04.01.2017	Wednesday
	Semester 2016-17		<u></u>

^{*}Subject to visibility of Moon.

Teaching days for Autumn Semester 2016-17 (w.e.f. 14.7.2016 to 09.11.2016)

Day			IV	lonths			
	July	August	September	October	Nove mber	Less for MTE/ THOMSO	Total days
Monday	18,25	1,8,22,29	5,19,26	3,10,17,24	7	-	14
Tuesday	19,26	2,9,16,23,30	6, 13 ,20,27	4,18,25	8	1.	14
Wednesday	20,27	3,10,17,24,31	7, 14 ,21,28	5,19,26	2,9	1	15
Thursday	14,21,28	4,11,18	1,8, 15 ,22,29	6,13,20,27	3	1	15
Friday	15,22,29	5,12,19,26	2,9,16,23,30	7,14, 21 ,28	4	2	15
Total days	12	21	21	18	6	5	78-5 = 73

Details of Saturday and Sunday used in MTE/THOMSO

MTE (September 13-16, 2016) - 13.09.2016 - Tuesday

14.09.2016 - Wednesday

15.09.2016 - Thursday

16.09.2016 - Friday

THOMSO (October 21-23, 2016) - 21.10.2016 – Friday

22.10.2016 - Saturday

23.10.2016 - Sunday

October 21, 2016 (Friday) - - 40 his will be a Non-Teaching Working Day.



INDIAN INSTITUTE OF TECHNOLOGY ROORKEE ACADEMIC CALENDAR FOR THE YEAR 2016-17 (Spring Semester)

S.No.	Details	Autumn	Semester
3.NU.	Details	Date	Day
1.	Reporting and Registration of New Ph.D. students.	26.12.2016	Monday
2.	Institute reopens and Registration of all new PG students	03.01.2017	Tuesday
3.	Registration of all existing students in the Departments/ Centres	03.01.2017	Tuesday
4.	Re-examination and Second examination on medical	04.01.2017 -	Wednesday-
	grounds (for Autumn Semester 2016-17)	06.01.2017	Friday
5.	Commencement of Classes for Spring Semester	04.01.2017	Wednesday
6.	Online subject registration of all new Ph.D. students	09.01.2017- 10.01.2017	Monday- Tuesday
7.	Last date for sending the grades of Re-examination	11.01.2017	Wednesday
8.	Submission of progress reports of the Ph.D. students to Academic Section by the Departments/Centres	16.01.2017	Monday
9.	Last date of Academic Registration	16.01.2017	Monday
10.	Uploading of roll lists of registered students	17.01.2017	Tuesday
11.	Last date of addition/deletion of courses	17.01.2017	Tuesday
12.	Uploading of final course-wise roll lists of registered students	20.01.2017	Friday
13.	Finalization of Seminar and Dissertation topics of 1st year	20.01.2017	Friday
-	M.Tech./M.U.R.P./M. Arch. and IDD/IMT 4 th year students		
	Republic Day	26.01.2017	Thursday
14.	Last date to finalize the list of candidates for Summer Internship at IITR	31.01.2017	Tuesday
15.	Notification of MTE Schedule	06.02.2017	Monday
16.	Notification to students regarding shortage of attendance by the Departments upto 16.02.2017	17.02.2017	Friday
17.	Mid Term Examination (MTE) for all students	20.02.2017-	Monday-
		23.02.2017	Thursday
18.	Request to Departments to send list of Institute Elective/ Open Elective/ Departmental Elective (both UG and PG) courses to be offered in Autumn Semester 2017-18	27.02.2017	Monday
19.	Submission of remaining document(s) by all new students	27.02.2017	Monday
20.	SCIENCE DAY	28.02.2017	Tuesday
21.	Last date for withdrawal of courses	03.03.2017	Friday
22.	Last date for requesting Second Examination on medical ground	03.03.2017	Friday
23.	Intimation to parents/guardians in respect of students having short attendance & upload on website	03.03.2017	Friday
24.	Last date to receive Institute Elective/ Open Elective/ Departmental Elective (UG-PG) courses to be offered in Autumn Semester from the departments	06.03.2017	Monday

		Spring S	emester
S.No.	Details	Date	Oay
25.	Notification to UG/IDD/IMT/IMS students about Institute Elective/ Open Elective/ Departmental Elective (UG-PG) courses to be offered in Autumn Semester 2017-18	10.03.2017	Friday
	Holi	13.03.2017	Monday
26.	Online subject registration for Institute Elective/ Open	14.03.2017-	Tuesday-
i	Elective/ Departmental Elective (UG-PG) courses by students for next semester	19.03.2017	Sunday
27.	Annual Hobbies Club Exhibition	18.03.2017-	: Saturday –
	SRISHTI - 2017	19.03.2017	Sunday
28.	Last date to display allotted list of Institute Electives/ Open Electives/ Departmental Electives to students for Autumn Semester 2017-18	22.03.2017	W⊭dnesday
29.	COGNIZANCE – 2017	24.03.2017-	Friday -
25.	OGGINEANOE 2011	26.03.2017	Sunday
30.	Ph.D. Interview	30.03.2017-	Thursday-
50.	TILD. INCOVICE	31.03.2017	Friday
31.	Notification of End Term Examination Schedule including Seating Plan (Institute Core and Elective Courses)	31.03.2017	Friday
32.	Annual Sports Meet	01.04.2017-	Saturday-
UL .	SANGRAM - 2017	02.04.2017	Sunday
33.	Notification to students regarding switching over from B.Tech. to IDD programme	03.04.2017	Monday
34.	Last date for Finalization of Time Tables by all Departments	04.04.2017	Tuesday
	Rama Navami	05.04.2017	Wednesday
	Mahavir Jayanti	09.04.2017	Sunday
35.	Seminar presentation of M.Tech. 1 st Year	10.04.2017-	Monday –
		13.04.2017	Thursday
36.	Online filling of Response Forms and Subject Registration	10.04.2017-	Monday -
	for next semester by all students	15.04.2017	Saturday
37.	Communication from Chairman, DAPC to Course Coordinators to submit the final list of students having short attendance	13.04.2017	Thursday
\	Good Friday	14.04.2017	Friday
38.	Viva-Voce Examination for Major Project (Only for B.	17.04.2017-	Monday -
00.	Tech. 4 th year students)	18.04.2017	Tuesday
39.	Display of list of students having short attendance upto 17.04.2017 by Departments/Centres and to send the Final list to Academic Section	18.04.2017	Tuesday
40.	Notification of detained students for shortage of attendance in End Term Examination by the Academic Section - 42 -	19.04.2017	Wednesday



S.No.	Details	Spring 9	Semester
J.14U.	Details	Date	Day
41.	Submission of proficiency grades by all concerned Officers-in-Charges to Academic Section	21.04.2017	Friday
42.	Last date of Teaching	21.04.2017	Friday
43.	End Term Examination (excluding Sunday)	22.04.2017-	Saturday -
	Practical examinations, if any, may be held during last few laboratory days	01.05.2017	Monday
44.	Notification to UG students regarding filling of Departmental Honour and Minor Specialization Courses	24.04.2017	Monday
45.	Submission of Final Year M.Tech./ M. Arch./ M.U.R.P./	01.05.2017	Monday -
	M.Tech. (ES)/ IDD/IMT Dissertation	-05.05.2017	Friday
46.	On-line filling of choices for DHC and MSC	02.05.2017- 06.05.2017	Monday- Friday
4 7.	Notification to PG students to switch over from M.Tech. to Ph.D. programme	02.05.2017	Monday
48.	Last date of showing of End Term Examination Answer Scripts to students	05.05.2017	Friday
49.	Finalization of grades by the Grade Moderation Committees	08.05.2017	Monday
50.	Display of Grades for all courses	08.05.2017	Monday
51.	Last date for sending Grades to Academic Section	08.05.2017	Monday
52.	Summer Vacation for Students (except for M.Tech./	08.05.2017-	Monday -
	IDD Final Year and Ph.D. students)	13.07.2017	Thursday
53.	Evaluation of Final Year M.Tech./ M. Arch./ M.U.R.P./ M.Tech.(ES)/ IDD/IMT Dissertation and sending g ades	08.05.2017- 12.05.2017	Monday - Friday
54.	Buddha Purnima	10.05.2017	Wednesday
55.	Last date to contact Departments/Centres by students for grade modification, if any	11.05.2017	Thursday
56.	!.ast date for sending modified grades to Academic Section	12.05.2017	Friday
57.	Last date for applying for the re-examination of Spring Semester 2017-18	15.05.2017	Monday
58.	Summer Vacation for Teaching Faculty	15.05.2017-	Monday -
		13.07.2017	Thursday
59.	Last date for preparation of Grade sheets (except for M.Tech./ M. Arch./ IDD Final Year students)	19.05.2017	Friday
60.	Notification regarding allotment of IDD programme and DHC and MSC	26.05.2017	Friday
61.	Declaration of result of switchover from M.Tech. to Ph.D.	31.05.2017	Thursday
62.	Last date for preparation of Grade Theets of Final Year M.Tech./ M. Arch/ IDD/IMT	05.06.2017	Monday

,



	Pakalla	Spring S	emester
S.No.	Details	Date	Day
63.	Id-ul-Fiter*	26.06.2017	Monday
64.	Reporting and Registration of New Ph.D. students	07.07.2017	Friday
65.	Institute reopens and Registration of all New PG students	13.07.2017	Thursday
66.	Reporting and Registrations of all New UG/IMT/IMS students	13.07.2017	Thursday
67.	Registration of all existing students in the Departments/ Centres	14.07.2017	Friday
68.	Re-examination and Second examination on medical grounds (for Spring Semester 2016-17)	14.07.2017- 15.07.2017	Friday - Saturday
69.	Commencement of Classes for Autumn Semester (2017-18)	17.07.2017	Monday

^{*}Subject to change on visibility of moon.



Teaching days for Spring Semester 2016-17 (w.e.f. 04.1.2017 to 21.4.2017)

			Month	าร		
Day	January	February	March	April	Less for MTE/Cogn.	Total days
Monday	9,16,23,30	6,13, 20 ,27	6,20,27	3,10,17	1	13
Tuesday	10,17,24,31	7,14, 21 ,28	7,14,21,28	4,11, 18	1	14
Wednesday	4,11,18,25	1,8,15, 22	1,8,15,22,29	12,19	1	14
Thursday	5,12,19	2,9,16, 23	2,9,16,23,30	6,13 ,20	1	14
Friday	6,13,20,27	3,10,17,24	3,10,17, 24 ,31	7,21	1	14
Total days	19	20	22	13	4	74-5 = 69

Details of days used in MTE and COGNIZANCE

MTE (February 20- 23, 2017) - 20.02.2017 - Monday

21.02.2017 -Tuesday 22.02.2017 - Wednesday 23.02.2017 - Thursday

COGNIZANCE (March 24-26, 2017) - 24.03.2017 – Friday

25.03.2017 - Saturday 26.03.2017 - Sunday

COGNIZANCE (March 24, Friday) - This will be a Non-Teaching working day

NOTE:

1. The committee also discussed the various events being organized throughout the year and recommended that only the following students' events be organized in an academic session:

1. Cultural Festival - THOMSO

2. Technical Festival - COGNIZANCE

3. Hobbies Club Annual Function - SRISHTI

4. Annual Sports Meet - SANGRAM

2. It was also decided that any poster related to above activities be put/ displayed only one week prior to the start of the activity with due approval from DOSW.



बैठक अनुभाग

भारतीय प्रौद्योगिकी संस्थान रूड़की

MEETING SECTION INDIAN INSTITUTE OF TECHNOLOGY ROORKEE



Minutes of the 52nd Meeting of the Executive Committee of the Senate (ECS) held on 10.03.2016 at 12.00 Noon in the Committee Room of the Institute.

The following were present:

	10 10110 H 21 1	
	Director Deputy Director Deans	Prof. Pradipta Banerji Prof. Vinod Kumar
	 (a) Academics (b) Alumni Affairs & International R (c) Faculty Affairs (d) SRIC (e) Students' Welfare (f) Associate Dean Academic Studies 	Prof. Deepak Kashyap Prof. M. Parida Prof. D.K. Nauriyal
4.	Heads of the Departments	
	 (a) Architecture & Planning (b) Biotechnology (c) Chemistry (d) Civil Engineering (e) Computer Science & Engg. (f) Earthquake Engineering (g) Earth Sciences (h) Electrical Engineering (i) Electronics & Communication (j) Humanities & Social Sciences (k) Hydrology 	Prof. Ila Gupta Dr. Partha Roy Prof. M.R. Maurya Prof. C.S.P. Ojha Prof. Manoj Misra Prof. Y. Singh Prof. Anand Joshi Prof. S.P. Srivastava Prof. Debashis Ghosh Prof. Rashmi Gaur Prof. D.S. Arya
	(1) Mathematics	Prof. V. K. Katiyar
	 (m) Mechanical & Industrial Engine (n) Metallurgical & Materials Engine (o) Management (p) Paper Technology (q) Physics (r) WRD&M 	ering Prof. Dinesh Kumar eering Prof. Anjan Sil Dr. Vinay Sharma Prof. Y.S. Negi Prof. Rajesh Srivastava Prof. S.K. Mishra

- 5. Heads of the Academic/Service Centres/ Centres of Excellence
 - (a) Alternate Hydro Energy Centre(b) Continuing Education Centre

(c) Nano Technology

(d) Disaster Mitigation & Management

(e) Institute Computer Centre

Prof. M.P. Sharma Prof. B.K. Gandhi Prof. S.K. Nath

Prof. B.K. Maheshwari Dr. R. Balasubramanian

6. Chairman, Library Advisory Committee

7. Chairman, JEE 8. Coordinator, QIP

9. Registrar (Secretary, ECS)

Prof. Praveen Kumar Dr. Rajat Rastogi Prof. B.K. Gandhi Mr. Prashant Garg

The issues at hand were then taken up:

Item No. 52.1: To consider the letter of Chairman, JAB-2016 to

1. Explore the possibilities of filling the vacant seat of an academic year in the next academic year

2. Reducing the seats from the academic programmes that are not high in demand.

The Committee considered the letter of Chairman JAB-2016 and the recommendations of the 47th meeting of the IAPC held on 11th February 2016 and decided that the number of seats in each UG programme be modified as per table given at **Appendix 'A'**. The modified seats will be applicable with effect from 2016-17 session.

Item No. 52.2: To consider the following options for those PG programmes where the percentage of placement offers is low:

(i) Reducing the number of intakes

(ii) Revising the structure completely for better employability

(iii) Merger of few PG Programmes

(iv) Closer of PG Programmes

The Committee considered the options proposed for PG programmes where the percentage of placement is low and observed that some of the PG programmes are too specialized and, therefore, the employment opportunities for the graduates of these programmes some time became less. After discussion, it was decided that the departments may look into their PG programmes and propose the modifications in light of discussion.

It was also decided that the proposed modification will be considered in the Senate meeting likely to be held in July 2016. It was also emphasized that while framing the structure and syllabi of the modified courses, the level of course be kept such that the graduates are directly absorbed in industries. Departments may also indicate such industries while sending their proposal. The modified PG structure programme will be implemented from 2017-18 session.

Item No. 52.3: To identify the key users from each of the Departments/Centres/Sections/Units etc. for the purpose of training related to ERP.

The Director briefed the members about the ERP Project named Advaita which was launched on 21st September 2015. Its 'As-Is' and 'To-be' processes have been completed and Wave-I of the project will start in May 2016. For the purpose of imparting training, nominations be invited of 'Key Users' to look into core-issue related to ERP. The nominations should be of people who would be directly related to processes in the Departments/Centres/Units. They should be from the faculty, technical and non-technical staff who have at least 2 – 3 years remaining service.

Item No. 52.4: To consider follow up action for preparing the structure of 2-year M.Tech. (Polymer Science) courses as decided by the Senate.

The Director conveyed all Head of the Departments about displeasure for not following the decision of the Senate. He stated that all faculty members/ officers and employees of the Institute must deliver and fulfill the requirements of the Institute besides the routine jobs and responsibilities.

If, anyone fails or refuses the Institute's work, it will be considered violation of 'Code of Conduct' and invites disciplinary action against him/her.

The meeting ended with a vote of thanks to the Chair.



TABLE

~ - - 1 - <i>i</i>			Appro	oved Se	ats fo	r 2016-17	
Code [*]	Programme	GE	OBC	sc	ST	PD	Total
1	2	3	4	5	6	7	8
R4-4104	ВІОТЕСН	17	10	5	3	OB-1	35
R4-4107	CHEMICAL	56	30	16	8	G-1 OB-1 SC-1	110
R4-4109	CIVIL	61	32	18	9	G-2 OB-1 ST-1	120
R4-4110	CSE	38	20	11	6	G-1 SC-1	75
R4-4111	ELECTRICAL	61	32	18	9	G-2 OB-1 ST-1	120
R4-4114	E&CE	40	22	12	6	G-1 SC-1	80
R5-4117	ENG. PHYSICS	15	8	5	2	G-1	30
R4-4125	MECHANICAL	50	27	15	8	G-2 SC-1	100
R4-4127	MMED	50	27	15	8	G-2 SC-1	100
R4-4135	PSE	18	9	5	3	OB-1	35
R4-4136	P&1	20	11	6	3	OB-1	40
R5-5101	ARCH.	18	9	5	3	G-1	35
R5-5302	IMT-GT	15	8	5	2	G-1	30
R5-5303	IMT-GPT	15	8	5	2	OB-1	30
R5-5502	IMS-APM	15	8	5	2	G-1	30
	TOTA	L 489	261	146	74	29	970

A

Appendix 'D' Item No. Senate/64.15

With Anurag Chauhan Deptit Topic Supervisor Wr. Anurag Chauhan AHEC EVOLVING OPTIMUL INTEGRATED Dr. R. P. Saini Mr. Mukesh Kumar Singhal AHEC OPTIMUL PLANNING OF MEDIUM HEAD Dr. Arun Kumar PRENEWAS SYSTEM MODEL Wis. Aanchal Sharma ARE RENEWAS PROVIDER PPLICATIONS Dr. P. S. Charin in Port Encir Model Mr. Pawan Rektia ASE FRONGONS PROVIDER Dr. P. S. Charin in Port Encir Model Mr. Rajeev Kumar Arya ASE RENIGONS PRASTING MATERIALS Dr. P. S. Charin in Provide Street Call MATE Mr. Rajeev Kumar Arya ASE REVICOPHOSPHAZENE BASED Dr. P. S. Charin in Provide Materials Mr. Brejan Lal BM STRESS NOW TREALS PROVING MATERIALS AND		·····						· 	. I		tem	No. S	enate	·/ UT.
With Annuag Chauhan AHEC EVOLVING OPTIMAL WITHGRANTED FROME AND ALLONG STAND-ALONG AND	PDC Date	18.03.16	04.04.16	22.03.16	24.02.16	22.03.16	09.02.16	22.02.16	17.02.16	23.02.16	07.03.16			09.02.16
Mane Depth	Examiner (For./Ind.)	Prof. M.H. Nehrir / USA Dr. S. Bandyopadhyay / IIT Bombay Dr. Ashok S. / NIT Calicut	Prof. Robert Boes / Switzerland Dr. O. P. Rahi / NIT Hamirpur	Prof. K. Orehouning / Switzerland Prof. J. Mathur / MNIT Jaipur		Prof. Maaruf Ali / UK Dr. B. Suman / DRDO Dehradun Prof. Majid Jamil / JMI, Delhi	Prof. S. Bruce Thomson / Canada Dr. P. Purang / IIT Bombay Dr. Sameer Pingle / Ahmedabad	Prof. Namjae Cho / Korea Prof. Geetika / MNIT Allahabad	Prof. Birendra K. Mishra / USA Prof. A. K. Jain / IITD	Dr. PDD Dominic / Malaysia Prof. Sasmita Palo / TISS Mumbai Dr. V.J.S. kumar / NIT Tiruchinapalli	Prof. Edyta Rudawska / Poland Dr. V.J.S. kumar / NIT Tiruchinapalli Prof. Sasmita Palo / TISS Mumbai	Prof. Jie Shen / China Dr. R. Baral / IIT Madras Prof. Meenakshi Gupta / IIT Bombay	Prof. Richard A. Posthuma / Texas Dr. Madhurima Deb / IIM Kashipur Dr. K. Das, XIM Bhubaneshwar	Prof. Arun K. Bhunia / USA Dr. A. K. Mohanty / NDRI Karnal Prof. B.S. Chadha / GNDU Amritsar
NameDeptt.Mr. Anurag ChauhanAHECMr. Mukesh Kumar SinghalARMs. Aanchal SharmaASEMr. Rajeev Kumar AryaASEMs. Akansha TyagiBMMs. Akansha TyagiBMMr. Bhajan LalBMMs. Bindu SinghBMMs. Pratibha VermaBMMs. Pooja KushwahaBMMs. Rashmi SinghBMMs. Rashmi SinghBMMs. Rashmi SinghBM	Supervisor	Dr. R. P. Saini	Dr. Arun Kumar	Dr. S. Y. Kulkarni Dr. P. S. Chani	Dr. Paritosh Mohanty	Dr. S. C. Sharma	Dr. R. L. Dhar	Dr. S. Rangnekar	Dr. A. K. Sharma	Dr. M. K. Rao	Dr. M. K. Rao	Dr. M. K. Rao	Dr. J. K. Nayak	Dr. N. K. Navani
Mr. Bhajan Lal Mr. Bhajan Lal Mr. Bratibha Verma Ms. Paooja Kushwaha Ms. Rashmi Singh Ms. Rashmi Singh Ms. Rashmi Singh Ms. Reshmi Singh			-	ENERGY EFFICIENT RETROFIT MODEL FOR ENGINEERING CAMPUSES IN INDIAN COMPOSITE CLIMATE	CYCLOPHOSPHAZENE BASED NANOPOROUS HYBRID MATERIALS FOR ENVIRONMENTAL APPLICATIONS	IMPLEMENTATION OF ROUTING PROTOCOL WITH SOFT COMPUTING METHODS IN WIRELESS SENSOR NFTWORK	STRESS AND ITS EFFECT ON PSYCHOLOGICAL HEALTH MODERATING ROLE OF EMOTIONAL INTELLIGENCE	JOB SATISAFCTION AND ORGANISATIONAL COMMITMENT AS PREDICTORS OF HUMAN CAPITAL CREATION	WORKING CAPITAL MANAGEMENT EFFICIENCY IN INDIA: AN EMPIRICAL ANALYSIS	EFFECTS OF INTELLECTUAL CAPITAL ON PERFORMANCE: EXAMININ THE ROLE OF MEDIATORS	PREDICTORS AND OUTCOMES OF CREATIVITY COMPONENTS: A STUDY IN INDIAN PSUS	EXAMINING THE INFLUENCE OF KNOWLEDGE ELEMENTS ON INDIVIDUAL COMPETENCE	COMPULSIVE BUYING & POST PURCHASE REGRET: FAMILIAL & NON- FAMILIAL STRESSORS	MOLECULAR AND BIOCHEMICAL STUDIES OF PROBIOTIC PROPERTIES OF Lactobacillus fermentum NKN51
Mr. Bhajan Lal Mr. Bhajan Lal Mr. Bratibha Verma Ms. Paoja Kushwaha Ms. Rashmi Singh Ms. Rashmi Singh Ms. Rashmi Singh Ms. Rashmi Singh	Deptt.	AHEC	AHEC	AR	ASE	ASE	BM	BM	BM	BM	BM	BM	BM	E B
13 12 14 10 0 0 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1		Mr. Anurag Chauhan	Mr. Mukesh Kumar Singhal	Ms. Aanchal Sharma	Mr. Pawan Rekha	Mr. Rajeev Kumar Arya	Ms. Akansha Tyagi -	Mr. Bhajan Lal	Mr. Utkarsh Goel	Ms. Bindu Singh	10 Ms. Pratibha Verma	Ms. Pooja Kushwaha	Ms. Rashmi Singh	Ms. Rekha Sharma

1	11 Me Caroto Trinothi	F	CONTRACTOR OF THE CONTRACTOR			į
		<u> </u>	MITOXANTRONE WITH G-QUADRUPLEX	Or. Kitu Barthwal	Prof. M. W. Germann / Georgia Dr. G. S. Kumar / CSIR Kolkata	01.02.16
15	15 Ms. Priya Vashisth	ВТ	DESIGNING OF GELLAN BASED ELECTROSPUN NANOFIBERS FOR BIOMEDICAL APPLICATIONS	Dr. Vikas Pruthi Dr. R. P. Singh	Prof. Pietro Matricardi / Roma Dr. B. Mishra / BHU Varanasi	27.01.16
16	Ms. Manju Narwal	BT	FUNCTIONAL CHARACTERIZATION OF ALPHAVIRAL PROTEINS INVOLVED IN REPLICATION	Dr. Shailly Tomar	Prof. R. Padmanabhan / USA Prof. Savita Yadav / AIIMS Delhi	07.03.16
17	Mr. Gulab Khushalrao Pathe	<u>გ</u>	DESIGN, SYNTHESIS AND ESTROGEN RECEPTOR BINDING STUDY OF FLAVONE AND INDANONE BASED LIGANDS	Dr. Naseem Ahmed	Prof. Saeed R. Khan / USA Prof. M. L. N. Rao / IITK Prof. Faiz Ahmed Khan / IITH	27.01.16
198		Cζ	DETERMINATION, DEGRADATION AND REMOVAL OF POLYCYCLIC AROMATIC HYDROCARBONS	Dr. Bina Gupta	Prof. Eric Lichtfouse / France Prof. M. Muneer / AMU Aligarh Prof. Alok Mittal / MNIT Bhonal	25.01.16
19	Mr. Vinod	ζ	ENANTIOSEPARATION OF CERTAIN PHARMACEUTICALS BY LIQUID CHROMATOGRAPHY	Dr. Ravi Bhushan	Prof. Danica Agbaba / Serbia Prof. Ajai Kumar Singh / IITD Prof. K. N. Singh / BHU Varanasi	19.02.16
5 1 -	Ivir. Sudnir Kumar Shoora	ბ	STUDIES ON SOME CHEMICAL SENSORS FOR BIOMOLECULES AND TOXIC METALS		Prof. B. J. Sanghvi / USA Prof. R.K. Mahajan /GNDU Amritsar Dr. K. P. Singh / CSIR Lucknow	08.03.16
7 8	21 Ms. Jyoti Singh Tomar	ბ	UNCONVENTIONAL PROTEIN TARGETS Dr. R. K. Peddinti OF MULTI DRUG RESISTANT ACENATOBACTER	Dr. R. K. Peddinti	Prof. David B. Neau / USA Dr. Savita Yadav / AllMS Delhi	21.03.16
7.7	Mr. Sanotsh Kumar Reddy P	⋩	OXIDATIVE C-C, C-S AND S-N COUPLINGS: SYNTHESIS OF ARYL SULFIDES, BIARYLS AND SULFONAMIDES	Dr. R. K. Peddinti	Prof. Janine Cossy / Farance Prof. I.N.N. Namboothiri / IIT Bombay Prof. G. Sekar / IIT Madras	31.03.16
3 3		CE	PERFORMANCE EVALUATION AND MICROBIAL COMMUNITY DYNAMICS OF VERMIFILTRATION	Dr. A. A. Kazmi	Prof. Fyyaz Ali Memon / UK Prof. M. M. Ghangrekar / IITKh	11.01.16
4 C	Ms. Nieena Kumari Sharma	H C	ONSITE FEMS		Prof. Masafumi Fujita / Japan Prof. A. K. Gupta / IITKh Prof. Mohammad Jawed / IITG	02.02.16
3 6	Ms. Catavateral Nupali Catifay	<u>Н</u>	J	ıt	Prof. Krishna R. Reddy / USA Prof. G. R. Dodagoudar / IITM	29.01.16
		<u>.</u>		Dr. Kamal Jain Dr. Ajay Gairola I	Prof. Xuan Zhu / Australia Prof. U. C. Mohanty, ITTBhu Dr. Bhoop Singh / NRDMS Dethi	11.02.16
27	Mr. Medalson Ronghang	CE	EFFICACY OF RIVERBANK FILTRATION IN HILLY AREA	Dr. Indu Mehrotra Dr. Pradeep Kumar Dr. Thomas Grischek F	Dr. Indu Mehrotra Prof. Chittaranjan Ray / USA Dr. Pradeep Kumar Prof. M. M. Ghangrekar / IITKh Dr. Thomas Grischek Prof. Ligy Philip / IITM	14.01.16

[
7	28 Mr. John Robert Prince M.	Ö		Dr. Bhupendra Singh	Dr. Bhupendra Singh Prof. B. Vijaya Rangan / Austria Prof. Ravindra Gettu / IITM	16.02.16
56	29 Mr. Asif Hussain Shah	핑	AN EXPERIMENTAL INVESTIGATION OF FIRE PERFORMANCE OF EARTHQUAKE DAMAGED STRUCTURES	Dr. U. K. Sharma	Prof. Faris Ali / UK Dr. S. Ghosh / IIT Bombay Prof. Sudhirkumar Barai / IITV	15.02.16
30	Mr. Ritu Raj	S	EFFECTS OF CROSS-SECTIONAL SHAPES ON RESPONSE OF TALL BUILDINGS UNDER WIND LOADS	Dr. A. K. Ahuja	Prof. R. Panneer Selvam / USA Dr. Rajeev Gupta / IITK Dr. Rajeev K. Gard / CRRI Delbi	29.02.16
33		핑	CONSOLIDATION INDUCED SOLUTE TRANSPORT THROUGH CLAY DEPOSITS	Dr. Mahendra Singh Dr. C. S. P. Ojha	Prof. Ronny Berndtsson / Sweden Prof. Rajesh Srivastava / IIT Kanpur	28.03.16
32		H)	ELECTROCHEMICAL TREATMENT OF SYNTHETIC PURIFIED TEREPHTHALIC ACID WASTEWATER	Dr. B. Prasad	Prof. Rueyan Doong / Taiwan Dr. Anurag Garg / IIT Bombay Dr. S. V. Mohan / CSIR Hyderabad	11.04.16
33	Mr. Mohammad Mardani Nokandeh	9	-Ġ	Dr. Indrajít Ghosh	Prof. Peter T. Savolainen / USA Dr. Sudip K. Roy / IIEST Shibpur	01.04.16
გ. 52	Mr. Anurag Aeron	DIMC	WEB BASED SPATIAL DECISION SUPPORT SYSTEM FOR FLOOD DISASTER MITIGATION	Dr. R. D. Garg Dr. D. S. Arya Dr. S. P. Addanwal	Dr. K. Prasad Vadrevu / USA Dr. B. Sahoo / IIT Kharagpur Prof. R. D. Gunta / MNIT Allahahad	14.03.16
35	Ms. Niyati Baliyan	CSE	QUALITY ASSESSMENT OF SEMANTIC I		Prof. Lorna Uden / UK Dr. D. Samanta / IITKh Dr. Lalit K. Singh / AED Mumbai	27.01.16
36	36 Mr. Shitala Prasad	CSE		Dr. P. Sateesh K. Dr. D. Ghosh	Prof. Izzet Kale / UK Prof. I. Chakrabarti / IIT Kharagpur	02.03.16
37		ITI ITI		Dr. M. K. Pathak Dr. S. P. Srivastava	Prof. Ned Mohan/USA Prof. D. Thukaram/ IISc Bangalore	15.01.16
ထ္ထ	Mr. Venkata Ramana Naik	Ш	Ш	Dr. S. P. Singh	Prof. A.K.S. Bhat / Canada Prof. Bhim Singh / IITD	04.02.16
ලිදු <u>.</u>	Mr. Poddar Monappa Gundappa	Ш	SIS	Dr. Vinod Kumar Dr. Y. Paul Sharma	Prof. Peter Macfarlane / UK Prof. S. Dandapat / IITG	18.02.16
6			ENTIFICATION AND FAULTS IN SYSTEM	Jr. R. P. Maheshwari	Dr. R. P. Maheshwari Prof. O.P. Malik / Canada Dr. Nilanjan Senroy / IITD	16.02.16
4	Mr. Nandkishor Wasudeorao Kinhekar		NG	Dr. N. P. Padhy Dr. H. O. Gupta	Prof. J.W. Maramgon Lima / Brazil Dr. B. K. Panigrahi / IITD Dr. Sukumar Mishra / IITD	11.02.16
42	42 Mr. Yadav Arvindkumar Ramrekha	Ш	CLASSIFICATION OF HARDWOOD SPECIES USING MULTIRESOLUTION DEATURE EXTRACTION TECHNIQUES	Dr. R. S. Anand Dr. M. L. Dewal Dr. Sangeeta Gupta	Prof. R.M. Rangayyan / Canada Prof. K. K. Biswas / IIT Delhi	04.03.16

-	Tall A					
4	45 ivii. Ajay Kriunteta	E&CE	APPLICATION OF FUZZY LOGIC AND PSO IN SOME IMAGE PROCESSING AND ANALYSIS METHODS	Dr. D. Ghosh	Prof. Shahram Latifi / USA Prof. M. Hanmandlu / IITD	17.02.16
4	44 Mr. Pankaj Kumar Pal	E&CE	DUAL-K SPACER ENGINEERD DEVICES FOR HIGH PERFORMANCE DIGITAL CIRCUIT/SRAM APPLICATIONS	Dr. S. Dasgupta Dr. B. K. Kaushik	Prof. Niraj K. Jha / USA Dr. M. Gupta / DU, Delhi	29.02.16
45	5 Mr. Ranjay Hazra	E&CE	PERFORMANCE ANALYSIS OF NON- COHERENT IR-UWB COOPERATIVE COMMUNICATION SYSTEM	Dr. Anshul Tyagi	Prof. K. S. Kwak / Korea Dr. A. K. Chaturvedi / IIT Kharagpur Dr. S. N. Marchant / IIT Bombav	07.03.16
4		E&CE	MODELING OF CROSSTALK EFFECTS IN CMOS GATE DRIVEN ON-CHIP INTERCONNECTS USING FDTD TECHNIQUE	Dr. B. K. Kaushik Dr. A. Patnik	Prof. EbyG. Friedman / New York Prof. Roy P. Paily / IIT Guwahati Prof. S. Jit / BHU Varansi	02.03.16
47		ES	FIRE	Dr. Pitamber Pati	Prof. Wolfgang Wagner / Austria Dr. R. S. Chatterjee / IIRS Dehradun Dr. Arindam Guha / NRS Hyderabad	09.03.16
8 4		È		Dr. D. K. Srivastava	Prof. Ramesh Teegavarapu / USA Dr. Arup Kumar Sarma / IITG	15.02.16
စ္53	9 Mr. Mahendra Kumar Choudhary 2	НΥ	1	Dr. D. K. Srivastava Dr. Sunita Devi	Prof. Vijay P. Singh / USA Dr. Arup Kumar Sarma / IITG	15.02.16
20	Mr. Kamlesh Kumar	MA	MARKOVIAN QUEUEING MODELS FOR I REPAIRABLE REDUNDANT SYSTEM WITH UNRELIABLE SERVERS	Dr. Madhu Jain	Prof. Kuo-Hsiung Wang / Taiwan Prof. U. C. Gupta / IITG	14.01.16
57	Ms. Renu Saini	MA	TRANSVERSE VIBRATION OF SOME NON-HOMOGENEOUS RECTANGULAR PLATES	Dr. Roshan Lal	Prof. Liew Kim Meow / Hong Kong Dr. S. C. Pradhan / IITKh	25.01.16
25		MA	FINITE ELEMENT STUDY OF TRANSPORT PHENOMENA IN COMPLEX FLUIDS	Dr. Rama Bhargava	Prof. Jinho Lee / South Korea Dr. Amiya Kumar Pani /IITB Prof. S. K. Tomar / PU Chandiaarh	24.02.16
53	Mr. Binod Kumar Singh	MA	COMPARATIVE STUDY OF SOME ALGORITHMS FOR IMAGE RECONSTRUCTION	Dr. Tanuja Srivastava	Prof. Nipon Theera-Umpon / Thiland Dr. R. K. S. Rathore / IIT Kanpur	28.03.16
5. 1.	54 Mr. Subhas Khajanchi	MΑ	T11	Dr. Sandip Banerjee	Dr. Sandip Banerjee Prof. Hien Tran / Releigh Dr. Ram Rup Sarkar / NCL Pune Dr. S. P. Chakrabarty / IIT Guwahati	04.03.16
22	Mr. Divyesh Patel	MA	<u>z</u>	Dr. Tanuja Srivastava	 	24.02.16
26	56 Ms. Manisha Verma	MA	NEW FEATURE DESCRIPTORS FOR IMAGE RETRIEVAL, OBJECT TRACKING AND SHOT DETECTION	Dr. R. Balasubramanian	Prof. D. Hatzinakos / Canada Prof. S. Agarwal / MNNIT Allahabad	11.04.16

A

27	57 Mr. Temesgen Berhanu Yallew	MIE	CHARACTERIZATION AND MACHINABILITY STUDY OF NATURAL	Dr. Pradeep Kumar Dr. Inderdeep Singh	Dr. Pradeep Kumar Prof. J. Paulo Davim / Portugal Dr. Inderdeep Singh Prof. Kamal K. Kar / IITK	22.02.16
58	Mr. Bhupendra Singh	MIE	CHARACTERIZATION OF WAKE AND SUPPRESSION OF FLUID FORCES ACTING ON A ROW OF SQUARE	Dr. S. Dutta Dr. B. K. Gandhi	Dr. Somashekhar S. Hiremath / IITM Prof. Muammer Ozgoren / Turkey Prof. P. K. Panigrahi / IIT. Kanpur Dr. Atul Sharma / IIT. Bombay	16.03.16
29	Mr. Saurabh Kumar Yadav	MIE	CYLINDERS A STUDY ON THE LUBRICATION OF HYDROSTATIC THRUST PAD BEARINGS	Dr. S. C. Sharma	Prof. Michel Fillon / France Dr. K. P. Nair / NIT Calicut	08 03 16
[8		ļ			Dr. R. K. Pandey / IIT Delhi	21.00.00
09	Mr. Pankaj Kumar Gupta	MIE	INVESTIGATIONS ON ECDM FOR SUBTRACTIVE MICRO-FABRICATION ON GLASS	Dr. Akshay Dvivedi Dr. Pradeep Kumar	Prof. P. Yarlagadda / Australia Dr. V. Yadava / MNNIT Allahabad	02.03.16
61	Mr. Rajesh Choudhary	MIE	STUDY OF NATURAL CONVECTION HEAT TRANSFER IN AL ₂ O ₃ /WATER NANOFLUIDS	Dr. S. Subudhi	Prof. Sushanta K. Mitra / Canada Dr. A. Dalal / IIT Guwahati Dr. K. C. Sahu / IIT Hyderabad	25.02.16
- 62	Mr. Ashutosh	MIE	MODELING OF RECONFIGURABLE MANUFACTURING SYSTEM WITH AVAILABILITY CONSIDERATION	Dr. Dinesh Kumar Dr. P. K. Jain	Prof. S. Nahavandi / Australia Dr. S. Sahu / IIT Bhubneshwar Dr. Ravi Shankar / IIT Delhi	21.03.16
34	Mr. Harak Sachin Sudhakar	MIE	DYNAMICS OF DRAFT GEAR FOR FREIGHT STOCK VEHICLE	Dr. S. C. Sharma Dr. S. P. Harsha	Prof. C. Nataraj / USA Dr. N. S. Vyas / IIT Kanpur	05.04.16
<u> </u>	64 Ms. Preeti Joshi	MIE	MULTIWALLED CARBON NANOTUBE COMPOSITES AND THEIR MECHANICAL BEHAVIOR	Dr. S. H. Upadhyay	Prof. Sondipon Adhikari / UK Dr. B. B. Khatua / IIT Kharagpur Dr. Satish Chandra Jain / IIT Mandi	03.03.16
65	Mr. Gagandeep Bhardwaj	MIE	FATIGUE CRACK GROWTH SIMULATIONS USING EXTENDED ISOGEOMETRIC ANALYSIS	Dr. I. V. Singh	Prof. Timon Rabczuk / Germany Prof. B. N. Rao / IIT Madras Dr. D. K. Sehgal / IIT Delhi	11.04.16
99	Mr. Sanjay Singh Rathore	MME	PROCESSING AND PROPERTIES OF SINTER-FORGED Fe-2Cu-0.7C-xMo ALLOYS	Dr. V. V. Dabhade	Prof. Wojiech Misiolek / Bethlehem Dr. G. Appa Rao / MD Hyderabad Dr. M. Karanjal / CNM Hyderabad	21.01.16
29	Mr. Tilak Chandra Joshi	MME	PROCESSING AND PROPERTIES OF Y ₂ O ₃ REINFORCED AL-7075 COMPOSITES BY POWDER FORGING	Dr. V. V. Dabhade Dr. Ujjwal Prakash	Prof. K. K. Shankar / Australia Dr. Rajesh Prasad / IITD	17.02.16
89	Mr. Bharat Bhushan	FN	DEVELOPMENT OF ALBUMIN BASED NANOPARTICLES FOR BIOMEDICAL APPLICATIONS	Dr. P. Gopinath	Prof. Yuzuru Takamura / Japan Dr. B. Bose / IIT Guwahati Prof. R. J. Kumar / AIMSRCC Kerla	11.03.16
69	69 Mr. Umesh Kumar Gaur	L Z	SYNTHESIS OF PURE AND DOPED CUO NANOSTRUCTURES AND THEIR MULTIFUNCTIONAL PROPERTIES	Dr. G. D. Verma Dr. Anil Kumar	Prof. M. Willander / Sweden Prof. Hitendra K. Malik / IIT Delhi Prof. K. Sethupathi / IIT Madras	08.04.16
2	70 Mr. Girish Semwal	표	DESIGN OF LONG PERIOD WAVEGUIDE GRATING DEVICES USING NATURAL OPTIMIZATION ALGORITHMS	Dr. Vipul Rastogi	Prof. Pascal Baldi / France Prof. Arun Kumar / IITD	25.02.16

ن	71	71 Mr K Thininothaiah	2	THE THOUSE CONTRACTOR OF THE PARTY OF THE PA			
1			Ē	CIRCUITS USING PLASMONIC MIM WAVEGUIDE	Dr. Vipul Rastogi Dr. N. P. Pathak	Prof. Ke Wu / Canada Prof. Vishnu Priye / ISM Dhanbad Prof. R. K. Sinha / Cella Chandianal	29.02.16
<u> </u>	72	Ms. Stuti Rani	Ŧ	STURUCTURAL OPTICAL AND MAGNETIC PROPERTIES OF SPINEL OXIDE NANOPARTICLES	Dr. G. D. Verma	Prof. M. S. Seehra / USA Prof. Sujeet Chaudhary / IIT Delhi	02.03.16
<u>-</u>	23	73 Mr. Harish Sharma Akkera	표	SYNTHESIS AND CHARACTERIZATION OF NI-MN-X (X: IN, SB) FSMA THIN FILMS	Dr. Davinder Kaur Dr. Inderdeep Singh	Prof. Jean Pivin / France Prof. Chiranjib Mitra / IISER Kolkata	11.03.16
<u> </u>	4	74 Ms. Lata Thakur	표	ANISOTROPIC QUARK GLUON PLASMA. DISSOCIATION OF QUARKONIUM STATES	Dr. B. K. Patra	Prof. Sangyong Jeon / Canada Dr. H. Mishra / PRL Ahmedabad	18.03.16
<u>'!</u>	c)	Ms. Navjot Kaur	Æ	TERNARY NITI BASED SHAPE MEMORY ALLOY THIN FILMS	Dr. Davinder Kaur	Prof. Claus Robholz / Cyprus Prof. Sujeet Chaildhary / IIT Delbi	02.03.16
- 1	9/	76 Ms. Dipti	H H	STUDY OF ATOMIC COLLISION PROCESSES AND PLASMA MODELING	Dr. Rajesh Srivastava	Dr. Yuri Ralchenko / USA Dr. Bobby Antony / ISM Dhanbad Dr. R.K. Sahoo / PRI. Ahmedahad	17.03.16
<u> </u>		// Mir. Jogi Ganesh Dattatreya Tadimeti 	PP T	ION TRANSPORT AND ITS FACILITATION IN ELECTRODIALYSIS	Dr. Pushplata	Prof. Tongwen Xu / China Dr. S. Sridhar / CSIR Hyderabad	12.02.16
· 3 -		Ms. Shilpa Kulkarni	PT	STUDIES ON DEINKING PROCESS AND UTILIZATION OF DEINKING SLUDGE	Dr. Vivek Kumar Dr. M. C. Bansal	Prof. M. Bin Ismail / Malaysia Prof. R. Subramanian / Div. Bonzolan	29.03.16
	262		VRDM	WRDM AN ANALYTICAL STUDY ON RIVERBED AND RIVERBANK FILTRATION	Dr. M. L. Kansal	Prof. Thomas Grischek / Germany Prof. M. Sekhar / IISc Bandalore	25.01.16
<u></u> [0	08	ar Chandniha	VRDM	WRDM WATERSHED SUSTAINABILITY INDEX FRAMEWORK AND ITS ESTIMATION FOR A WATERSHED	Dr. M. L. Kansal Dr. Aditya Tyagi	Prof. V. Sridhar / USA Prof. C. Chatterjee / IIT Kharagpur	11.03.16
ν α	α		VRDM	WRDM IMPACT OF CLIMATE AND LANDUSE CHANGES ON GROUNDWATER IN A CANAL COMMAND	Dr. Deepak Khare	Dr. Qin Xiaosheng / Singapore Prof. K. N. Tiwari / IIT Kharagpur Prof. Rohit Goyal / MNIT Jaipur	02.04.16
ر ام	4 6	Mr Nordio Decite	ا ک	S JDIES AS	Dr. A. K. Singh Dr. A. K. Jain	Prof. J.A.O.S. Pedreno / Spain Prof. Sushil K. Singh / BHU Varansi	11.04.16
<u> </u>	2 -	oo ivii. Navailo Daratha	11		Dr. B. Das	Prof. Mesut Baran / Raleigh Prof. P. R. Bijwe / IIT Delhi	11.04.16
	¥ =	64 Mr. Man Mohan Garg	Щ	MODELING AND CONTROL OF DC-DC CONVERTERS	Dr. Y. Vijay Hote Dr. M. K. Pathak	Prof. Bimal K. Bose / USA Dr. K. R. Rajagopal / IIT Delhi	11.04.16

Appendix-A

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF THE DEPARTMENT:

EARTH SCIENCES

Subject Code: ESN- 628 1. :

Course Title: Field Geology

Contact Hours 2.

L: 0

T:0

P:6

Examination Duration (Hrs): 3.

Theory: 0

Practical: 6

Relative Weightage: CWS:0 4.

PRS: 50

MTE:0

ETE:0

PRE: 50

Credits: 3 5.

6. Semester: Spring

7. Subject Area: PEC

Pre-requisite: Nil 8.

Objective: To impart advance training on mapping analysis of geological structures in 9.

Details of Course: Geological and Structural mapping of structures in field. 10.

S No. List of field experiments						
S. No.	Identification of mappable rock units and common structural and tectonic settings.					
1.	Identification of mappable rock units and common structural compass, in field					
	Hands-on training on GPS, Sunto and Bruton geological compass in field.					
2.	Observations on planar and linear tabric elements.					
3.	o: if we and mothods for collection of oriented samples ill lielu.					
-	Mapping of mesoscopic brittle structures: Fractures, joints, veins and different					
4.	times of faulte					
5.	Manning of ductile structures: Folds, ductile shear zones and boudins.					
	Compatric and kinematic analysis of observations on online structures.					
6.	Geometric and kinematic analysis of checonations on ductile structures.					
7.	Geometric and kinematic analysis of observations on ductile structures.					

11. Suggested Books:

S.No.	Name of Authors/ Books/ Publishers	Year
1.	Lisle, R. J., 2004. Geological Structures and maps: A Practical Guide,	2004
2	3rd Edn, , Elsevier, Amsterdam Lisle, R, J., Braham, P. and Barnes, J. W. Basic geological mapping, 2011. Basic Geological mapping. 5th Edn, Wiley and Sons,, New York	2011
3.	Bennison, G. H., Olver. P. A. and Mossley, K. 2013. An introduction to geological structures and maps. 8th Edn. Taylor and Francis, London.	2013

Note: The course is based on fieldwork that be carried out during the weekends and/or holidays without effecting regular teaching hours.

Appendix-B

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF THE DEPARTMENT:

EARTH SCIENCES

1. Subject Code: ESN- 583

Course Title: Petrophysics and Seismic Rock

Characterization

2. Contact Hours: L: 3

T:1

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS:0

MTE: 25

ETE: 50

PRE:0

Credits: 4 5.

6. Semester: Both

7. Subject Area: PEC

Pre-requisite: Knowledge of well-logging and seismic prospecting 8.

Objective: The main aim of the course is to give knowledge about the rocks and fluids 9. to various frequencies of measurements and a working knowledge of how to integrate a measured property in lab or using well logs to its practical applications in Exploration Geophysics and Reservoir Engineering.

10. Details of Course:

SI. No.	Contents	Contact Hours
1.	Scale, Units and Conversions: Core-log-to-seismic scales, common units and conversion	2
2.	Stress-Strain Behaviour: Principal stresses, priciples of elastic and viscoelastic behaviour, Elastic Tensor, Anisotropy (Intrinsic and stress)	5
. 3 .	Dynamic Properties of Reservoir Rocks: reservoir rock types, diagenesis, natural fractures, dissoution, porosity (dual and triple) systems, permeability, mobility, wettability, capillarity	5
4.	Dynamic properties of Reservoir Fluids: Hydrocarbon composition (SARA), classification of reservoir fluids, phase behaviour of single and multicomponent fluid systems, PVT properties	5
5.	Best Practices in Core Analysis: Routine Core Analysis, Special Core Analysis, Strength tests, GRI. Conventional versus unconventional, steady state and pseudo-steady state permeability	5
6.	Multiscale Core Data-Overlap/Redundancy; MICP, Adsorption, XRD, XRF, FTIR,CT SCAN, QEMSCAN	3
7.	Core-to-Log Calibration and Seismic Property Evaluation: impact of mineralogy. Vshale and salinity on grain density, porosity, sturation, elastic moduli 57 - Well-to-Seismic Ties: Basics of VSP, Checkshot surveys, time- depth cirve and velocity profile, Synthetic Seismograms,	5

8.	Rock-Fluid Interaction: matrix alteration due to dissolution/ cementation and resulting change in porosity and permeability; grain-to-grain bond weakening, surface energy reduction resulting in shear weakenning.	2
9.	Experimental Methods in Seismic Rock Property Measurement: Working principles of piezoelectric crystals, strain gages (semiconductor, foil), tilt meters. LVDT, ultrasonic measurement, static load measurement, low-frequency measurements.	5
10.	Fluid Substitution and Effective Medium Modeling: Fluid substitution in core and log data using idealistic Gassmann, and Generalized Gass, amm & Hashin-Shtrikman, introducing secondary porosity and other heterogeneity to realize equivalent porosity medium, modeling elastic properties using effective medium models	5
	Total	42

11. Suggested Books:

S.No.	Name of Authors/ Books/ Publishers	Year of
		Publications
1.	Petrophysics: Theory and Practice of Measuring	2010
	Author: Dkennar Tiab and Eric C Donaldson	
	Publishers: Gulf Professional Publishing	
2.	Petroleum Reservoir Rock and Fluid Properties	2013
	Author: Abhijit Y Dandekar	
	Publisher: CRC Press	
3.	Petrophysics :Fundamentals of Petrophysics of Oil and Gas	2012
	Reservoirs	
	Authors: Leonid Buryakovsky, George V. Chilingar, Herman H.	·
	Rieke, Sanghee Shin	
	Publisher: Scrivener (Wiley) Publishing I.I.C.	
4.	A Geoscientist's Guide to Petrophysics	2007
	Authors: Bernard Zinszner, Francois-Maric Pellerin	
	Publisher: IFP Publication	
5.	Fundamentals of Borehole Seismic Technology	2010
	Author: Pereira, A.M. and Jones, M.	
	Publisher: Schlumberger	



Appendix-D

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPT. /CENTRE: Humanities and Social Sciences

1. Subject Code: H S	N 513	Course Title	e: Public Po	olicy: Theory and Practice	
2. Contact Hours:	L: 3	T: 1	P: 0		
3. Examination Dura	ntion (Hrs):	Theory 3		Practical 0	
4. Relative Weighta	ge: CWS	25 PRS 0	MTE 25	ETE 50 PRE 0	

5. Credits: 4 6. Semester: Spring 7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart knowledge of public policy-making and understanding of the complexity of policy problems.

10. Details of Course

S. No.	Contents	Contact hours
1	Introduction:	5
	Concepts, Nature, Scope and Significance of Public Policy; Types of Public Policy: Regulatory, Welfare, Distributive and Re-distributive;	
	Evolution of Public Policy Studies; Attributes of Good Policy-Making Process	
2	Theories and Models of Policy Making:	7
	Policy Paradigms; Elite Theory and Pluralism; Institutionalism and	
	Rational Choice Theory; Systems Theory and the Process Model; Policy	
	Networks; Public Policy Cycle; Policy Instruments	
3	Policy Implementation:	7
	Actors and institutions; Policy making process; Aspects of policy	
	implementation: State and Central Governments; Public Opinion and	
	Interest Group Politics; Problems in Public Policy Implementation:	
	Conceptual, Political and Administrative Problems; Conditions for	
	Successful Implementation	
4	Policy Monitoring and Evaluation:	7
	Policy Monitoring: Approaches and Techniques; Constraints in Policy	
	Monitoring; Measures for Effective Policy Monitoring; Policy	
	Evaluation: Role, Process and Grenia, Types of Evaluation; Evaluating	
	Agencies; Problems in Policy Evaluation	



5	Domains of Public Policy, Policy Changes and Convergence: Development Policies: Population, Education, Health, Employment; Welfare Policies: Social Security and Social Insurance; Policy changes; Cross-national policy convergence; Causes and conditions of cross-national convergence; Future challenges for policy analysis	8
6	Policy Making in India: Constitutional Framework for Policy Making: Legislature, Executive, Judiciary, National Development Council; Role of NITI Aayog; Other Forces in Policy-making: NGOs, Political Parties, Interest Groups, Professional Lobbyists, Media, Professional and Corporate Bodies, International organizations and Agencies; Policy Interplays and Trade- offs; Deficiencies in the Existing Policy-Making System and Process; Reforms in Policy Making Process	- -

11. Suggested Books

11. S S. No.	Name of Books/Author/Publisher	Year of Publication/ Reprint
0.110.		2003
1	William N Dunn, Public Policy Analysis - An Introduction,	2003
1) ord value	2013
2	Pearson, 3 th Edition Thomas Dye, <i>Understanding Public Policy</i> , Prentice Hall, New	
2	1 All Didition	2003
3	James Anderson, Public Policy Making: An Introduction,	
i	1 - 1 XCCC 210 Edition	2002
4	1 Trill I Deter Unne Implementing Public Folicy, Sago	2009
5	RV Vaidyanatha Ayyar, Public Policy Making in Thura, I can be a	
•	Uing and the Massa Dollar	2012
6	Prabir Kumar De, Public Policy and Systems, Pearson Education	
"	India, New Delhi	



Appendix 'H' Item No. Senate/64.19

Appendix-C

ANSWER	BOOK	NO.:
---------------	-------------	------

No. o	f Supr	lementary	/ Sheets	
-------	--------	-----------	----------	--

24 pages

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE END TERM EXAMINATION

Name of	of the StudentClass/Bra	anch	<u>.</u>
Subjec	t CodeTitle	•	
Date of	Enrollment NoSignature o	of Invigilator	······································
	INSTRUCTION	Question Number	MARKS AWARDED
ľ	E: Read the following INSTRUCTIONS carefully.		
1.	Write on both sides of the paper of the answer book.	1	ti .
2	Give the number of question and its part, if any, at the beginning of the answer. Use the same number for the answer as is given in question paper.	2	
3.	No part of this book is to be torn off. Rough work should be scored out.	3	
4.	Supplementary sheets will be provided on request.		
5.	No one will be allowed to borrow instrument, pen pencil, eraser, scale, ink etc. in the Examination Room.	4	
6.	No candidate, who leaves the Examination Room while dong his paper except for a short period with the permission of the invigilator, will be permitted to return to the Examination	5	
	Room.	6	
7.	Supplementary sheet(s) and graph paper used, if any, must be tagged with the answer-book.	7	
8.	Please handover answer-book to the invigilator at the end to the exam.		
9.	Mobile Phone, Pager, Programmable Calculator, Laptop etc. are strictly prohibited inside the Examination Room.	8	
10.	Use of any unfair-means will render the students liable or punishment as per institute regulations.	9	
	I have read the above instructions.	10	
	Signature of Student	- · ·	,
	- 61 -	Total	



16 pages

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE MID TERM EXAMINATION

	of the StudentClass/Bra		
-	of Exam Enrollment No. Signature of		
NOT	INSTRUCTION To Dood the following INSTRUCTIONS perefully	Question Number	MARKS AWARDED
1.	E: Read the following INSTRUCTIONS carefully. Write on both sides of the paper of the answer book.	1	
2.	Give the number of question and its part, if any, at the		
۷.	beginning of the answer. Use the same number for the answer as is given in question paper.	2	
3.	No part of this book is to be torn off. Rough work should be scored out.	3	
4.	Supplementary sheets will be provided on request.		
5.	No one will be allowed to borrow instrument, pen pencil, eraser, scale, ink etc. in the Examination Room.	4	
6.	his paper except for a short period with the permission of the	5	
	invigilator, will be permitted to return to the Examination Room.	6	
7.	Supplementary sheet(s) and graph paper used, if any, must be tagged with the answer-book.	7	
8.	Please handover answer-book to the invigilator at the end to the exam.		
9.	Mobile Phone, Pager, Programmable Calculator, Laptop etc. are strictly prohibited inside the Examination Room.	8	
10.	Use of any unfair-means will render the students liable or punishment as per institute regulations.	9	
	I have read the above instructions.	10	
	Signature of Student	Total	



Α	NIC	A/E	: Г	DO	$\Delta \mathbf{L}$	NO	
А	IN 5	VVF	- K	BU	ハル	NU	١.

Supplementary	Sheet	No
---------------	-------	----

4 pages

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE MID/END TERM EXAMINATION

Name of the Student		Class/Branch	• • • • • • • • • • • • • • • • • • • •
Subject Code	Title		
Date of Exam	.Enrollment NoS	Signature of Invigilator	



DEPARTMENT OF CHEMISTRY INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

M.Sc. (Chemistry) Chemistry XX CY I Program Code: Department: Year:

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE DEPARTMENT OF CHEMISTRY

M.Sc. (Chemistry) Chemistry

ZX □ CX Program Code: Department: Year:

<u>~</u>	ŀ	E PRE		1	20	- 0	- 0	50		- 0	- 05		.			100	3	- 001	- 05	<u> </u>		
ght (%	-	ETE	-		'	- 20	20	1	7	2	<u>.</u>	_	_	,		1,	- -	<u> </u>	4	,		
Relative Weight (%)		MTE			1	25	25	3,0	C7	25	25	-							25	3		
Relati		PRS			50	1	.,		1	, :							,	t				
		CWS			ı	25	25	3	C7	25	25			ı	-				25	3		
Exam.	Duration	a			80	0	c	> (5	0	0					,	0	0		0		
Ex	Dur	Т			0	3	7	,	3	(7)	c	1					0	0	,	r		_
; t	/eek	A.			80	0	0	5	0	0	c	,	80			ļ	<u> </u>	0		O	0	
Contact	Hours/Week	T.		(tumn)	0	0		>	O	0	0	>	0	1	(2)	pring)	<u> </u>	0	, «	0	0	
-	Η̈́	1	,	II (Au	0	3	, ,	0	<u>ښ</u>	m	4	,	15		TX7 /C.	(C) AT	0	С	,	<u> </u>	'n	
		Credits		Semester-III (Autumn)	4	~	, ,	3.	m	6	2	0	19			Semester-1v (Spring)	8	2	1	m	13	
		Sub.	Area	S	PCC	DEC	ן נו נו	PEC	PEC	PEC		יייי					PR	CFM	JACIVI	PEC		_
Teaching Scheme		Courses Title			I aborotory.III		Program Elective-11	Program Elective-III	Propram Elective-IV	Description Flective-V	Flogram Liceave	Program Elective-VI	Total				Droject	i Iglaci	Seminar	Program Elective-VII	Total	
		Sub Code			CVN1 601	5	CYN-	CYN-	CVN	City	CYN-	CXN					CV2 ICV2		CYN- 604	-NAJ	+	
		S.No.			-	-	7	3		j .	Ċ.	S					-	<u>-</u>	7	7	;	_



	4	13	
	3	16	76
	2	23	
		21	
Summary	Semester	Semester-wise Total credits	Total Credits

Program Elective Courses in Second Semester (M.Sc. Chemistry)

		Teaching Scheme		; ;	Ü H	Contact Hours/Week		Exam. Duration	ı.	-	Relativ	Relative Weight (%)	(%)	
			.		,				\dagger		o digit	N ATTE	Line	תמת
S.No.	S.No. Sub Code	Courses Title	Sub.	Credits	7	<u></u>	ب	<u>.</u>		2 2 2 2	S	PKS MIE	1 1	T K E
• • •			Area											
	CYN-512	CYN-512 Nuclear and Radiochemistry	PEC	. 03	<i>κ</i> υ.	0	0	n	0	25	ı	25	20	ı
2.	CYN-514	CYN-514 Heterocyclic Chemistry	PEC	03	m	0	0	3	0	25	1	25	50	ı
3.	CYN-516	CYN-516 Chemistry of Main Group and Transition Elements	PEC	03	3	0	0	د .	0	25	ı	25	50	t.
							_	,	<	20	-	30	20	
4.	CYN-518	CYN-518 Structure, Bonding and Properties of Solids) 14.4 	03	3	D ·	n n	٠	>	C7		3	0	•

Program Elective Courses in Third Semester (M.Sc. Chemistry)

		•										_						_
			PRE		1		:		1		ı		ı		i		1	
()	(S)		ETE	C L	2	1	3)) 	Š	20	(2	Ş	0C		40	
1 1 1	Kelative Weignt (%)		MTE	20	C7	-	25		25		25		3	i	. 72	,	20	
	Kelativ		PRS		ı	-	•		4		ı		ı		1		20	
		į	CWS	i	£7		25		25		25		72		25		20	100
	ij	Duration	Ъ		<u> </u>		0.		0		0	,	-		0		0	
	Exam.	Dura	L	,		-	ω.		m ´		ر		m		ന		. 7	
	ಕ	/eek	ď		<u> </u>		0		0		0		o .		<u> </u>		2/2	
	Contact	Hours/Week	[ļ			0		0		0		<u> </u>		0		0	
		H	,,,		m		က်		m		~		m		m		7	
			Credits		03		03		03		03	į	. 03		03		03	
			Sub.	Area	PEC		PEC		PEC		PEC		PEC		PEC		PEC	
A CONTRACTOR OF THE CONTRACTOR	Teaching Scheme)	Courses Title		ŀ	Techniques	1	Microanalysis	Electroanalytical Chemistry		Inorganic Biochemistry and	Reaction Mechanism	Solid-State Chemistry and its	Applications	Frontiers in Bioinorganic	Chemistry	CYN-615 Crystal and Molecular Structure	
34			Sub Code		CYN-603		CYN-605		CYN-607		CYN-609		CYN-611		CYN-613		CYN-615	
			Ś	No	1.		2.	•	33.		4		5.		9.		7.	

- 66 **-**

					- -					_					.	· ·		
1	1	ľ		1		ı		ı		,		,					ı 	
20	20	20		20		20		20		Ç))	40	2	20	3	2	3	
25	25	25		25)	25		25	}	30	ì	20	9	50	3	30	3	
ı	ı	,		'		1		1			1	000	07		1		ī	
25	25	25		25	ì	25		25	3	30	C.7	00	07	300	77		C 7	
<u> </u>	0	 c			>	0	·	_	>	i	>			,	>	í	0	
3	3	Cr.)	64	า	~	ī. 1	7	J	,	Λ	,	7	1	<u>ن</u>	,	m	
0	0	c	>	c	>	0	>	c	> ·		<u> </u>	9	7/7	\ \ \	- -	1	0	
0	0		>	6	>,		>.	c	>	,	o 	ļ	<u> </u>	١	<u> </u>		0	
3	m	"	<u> </u>	,	າ	7	<u>, </u>	,	<u>ئ</u>		m	\ -	7	-	8		<u>~</u>	-
03	03	0.2	S		03	5	2		03		03		03		03		03	
PEC	PEC	0.00	7. 7.	9	PEC	CHA	THE C	()	PEC		PEC		PEC		PEC		PEC	
Supramolecular Chemistry	Modern Organic Synthetic	Methods	Organic Structure	Determination	Organic Semiconductors		Proteins and Polypeptides		Advanced Surface and Colloidal	Chemistry	Advanced Physical Chemistry		Materials Chemistry		Nanoscale Materials: Properties	and Applications	CYN-635 Advanced Magnetic Resonance	Spectroscopy
CYN-617	CYN-619		CYN-621		CYN-623		CYN-625		CYN-627		CYN-629		CYN-631	-	CYN-633		CYN-635	
8.	.6		10.		11.		12.		13.		14.		15.		16.		67	
l		l	_		1			_		•						-	67	-

		Program Elective	Cours	Elective Courses III rout on Semester (1715)	20 151	110011				1				
,					C	Contact		Exam.	<u>.</u>		Relativ	Relative Weight (%)	t (%)	<u> </u>
		I Cavining Schwing			Hon	Hours/Week	ek	Duration	ion					
S.N.O.	Sub Code	Courses Title	Sub.	Credits	Ţ	Τ	Q.	H	ы	CWS	PRS	MTE	ETE	PRE
orre.			Area					-					(
1	CYN-606	CYN-606 Total Synthesis	PEC	03	3	.0	0	<u>—</u>	0	. 25	1	25	05	ı
					,	(c	c		30	6	25	50	
2.	CYN-608	CYN-608 Chemical Biology	PEC	03	 	⊃. 	>	ر م	>	7	>	3)	
			1		,	6	5	C	c	5	20	2.0	40	1
3.	CYN-610	CYN-610 Molecular Modeling and	PEC	03	7	- ·	7/7	٧	>	3)			
		Simulations		-										
			\ ;		,	٥		,	c	25	•	25	50	1
4.	CYN-612	CYN-612 Carbon Nanomaterials and	PEC	03	γ)	>	>	<u></u>	>	j -	ı) 1) }	
	,	their Applications			(•	Ç	c	c	000	20	20	40	1
5.	CYN-614	Enantiomeric Separation	DEC	03 —	7))	7/7	7	>	07	3	1	-	

NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-501

Course Title: Quantum Chemistry, Symmetry and Group Theory

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weight: CWS: 25

PRS: 0

PRE: 0

5. Credits: 3

MTE: 25

T: 0

ETE: 50

8. Pre-requisite: Nil

6. Semester: Autumn

7. Subject Area: PCC

9. Objective: To provide basic concepts and mathematical treatment of atomic model, chemical bond, symmetry and group theory.

10. Details of Course:

S. No.	Contents	Contact Hours
1.	Quantum chemistry: Basic postulates, eigenvalues and eigenvectors, Hermitian operators, applications including translational, vibrational and rotational degrees of freedom - particle in 1D/2D/3D box, particle in a ring, rigid rotor, harmonic oscillator. Electronic, vibrational and rotational transitions. Solution of Schrödinger equation for the hydrogen atom; radial and angular functions, atomic orbitals and electron spin. Multi-electron systems, term symbols.	14
2.	Approximate techniques and chemical bonding: Born-Oppenmenter approximation, variation and perturbation methods with examples. Valence bond theory including mathematical treatment of sp, sp ² and sp ³ hybridized orbitals, molecular orbital theory with suitable examples, Huckel molecular orbital approach.	14
3.	Introduction to semi-empirical and <i>ab initio</i> methods. Molecular symmetry and group theory: The concept of groups, symmetry operations and symmetry elements in molecules, matrix representations of symmetry operations, point groups, representation of a group, reducible and irreducible representations, great orthogonality theorem and its consequences,	
4.	Applications of group theory: Group theory and quantum mechanics, applications of group theory to atomic orbitals in ligand fields, molecular orbitals, symmetry of normal modes of vibrations, prediction of infrared, Raman active vibrational modes, and electronic transitions.	1 6
	modes, and electronic transitions. Total	42

S. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
-	Levine, I. N. "Quantum Chemistry", 7th Ed., PHI Learning Pvt. Ltd., Delhi.	2013
	McQuarrie, D. A. "Quantum Chemistry" Reprint, Viva Books.	2007
2	Atkins, P. "Molecular Quantum Mechanics", 4 th Ed., Oxford University Press.	2010
3.	Atkins, P. "Molecular Quantum Mechanics", 4 Ed., Oxford States Cotton, F. A., "Chemical Applications of Group Theory", Reprint, Wiley Eastern.	1994
4.	Cotton, F. A., "Chemical Applications of Group Theory, Repling Willow	1

NAME OF DEPT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-503

Course Title: Thermodynamics, Interfaces and Solids

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weight: CWS: 25

MTE: 25 PRS: 0

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective of Course: To familiarize the students with thermodynamics aspects of chemical phase equilibria, surface processes and ionic systems, and solids.

S.No	Contents	Contact Hours
· • _		
1.	Classical thermodynamics: Thermodynamic treatment of phase equilibria, thermodynamic properties of solutions, chemical potential, chemical potential of real gases and fugacity, thermodynamic function of mixing, thermodynamic treatment of ideal and non-ideal solutions, concept of activity, excess thermodynamic functions. Thermodynamic equilibria in one and two component systems.	12
2.	Statistical thermodynamics: Concept of microstates and ensembles, microcanonical, canonical and grand canonical ensemble, average distribution, partition functions and its relation with thermodynamics properties, Maxwell-Boltzmann, Bose-Einstein, Fermi-Dirac statistics, Molecular partition functions, translational, vibrational, and rotational partition functions. Ideal monoatomic and diatomic gases and their thermodynamic properties.	10
3.	Thermodynamics of surfaces and interphases: Surface and interfacial phenomenon, macromolecules, adsorption of gases by solids, BET theorem, determination of surface area of solids, adsorption from solution, electrical phenomenon of interphases.	7
4.	Thermodynamics of ionic systems: Thermodynamics of reversible and irreversible electrochemical systems, thermodynamic foundation of theory of ionic interaction and calculation of energy of ionic interaction, interpretation of electrical conductance of electrolytes, thermodynamic treatment of diffusion potential. Thermodynamics of different types of chemical processes accounting in living systems, metabolic and biosynthetic reaction, thermodynamics of ionic polymers.	
5.	Solids: Structural classification of binary (AX, AX ₂ , etc.) and ternary (ABX, ABX ₂ , ABX ₃ , AB ₂ X ₄ , etc.) compositions, powder X-ray diffraction – Bragg's peak, absences, indexing of simple systems. Bonding in solids – introduction to metals, insulators and semiconductors, electronic structure of solids. Electrical conductivity, mobility, thermal conductivity, and specific heat of solids. Magnetic properties of solids, magnetization and susceptibility.	
	Total	42



S.	Name of Authors/ Books/ Publishers	Year of Publication/
N o.		Reprint
1.	Seddon, J. M. and Gale, J. D., "Thermodynamics and Statistical Mechanics", Royal Society of Chemistry.	2001
2.	McQuarrie, D. A. and Simon, J. D., "Physical Chemistry", Reprint, Viva Student Edition.	2013
3.	McQuarrie, D. A., "Statistical Mechanics", Reprint, Viva Books Pvt. Ltd.	2013
	Atkins, P.W., "Physical Chemistry", 7th Ed., ELBS, Oxford University Press.	2003
4.		
	Silbey, R.J. and Alberty, R.A., "Physical Chemistry", 4th Ed., John Wiley & Sons,	2003
5.	Inc., New York.	
6.	West, A. R., "Solid State Chemistry and its Applications" Reprint, Wiley, India	1987
7.	Wells, A. F., "Structural Inorganic Chemistry", 5th edn., Clarendon Press, Oxford	1984
8.	Spaldin, N. "Magnetic Materials: Fundamentals and Device Applications", Cambridge University Press	2003



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject code: CYN-505

Course Title: Basic Analytical Chemistry

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS: 0

MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart basic knowledge in different concepts of analytical chemistry.

10. Details of the Course:

S. No.	Contents	Contact Hours
1.	Measurement basics: Signal and noise, sensitivity and detection, hardware and software techniques for enhancing signal to noise, accuracy and instrument	4
	calibration, basic electronic components and circuits for instrumentation used in chemical analysis, optical components for instrumentation used in chemical analysis.	
2.	Statistical concepts for analytical chemistry: Error analysis, regression plots, criteria for rejection of data, Q-test, t-test, F - test, ANOVA, control chart, use of spreadsheet.	6
3.	Spectral methods: Spectrophotometry—Beer-Lambert law, its applications and limitations, single and double beam spectrophotometer, analysis of mixtures, fluorimetry, nephelometry, turbiditmetry. Atomic absorption spectrometry—principle and applications, flame emission spectrometry (flame photometry).	7
4.	Electroanalytical methods: Polarography, amperometric and bio-amperometric titrations.	5
5.	Nuclear methods: Fundamentals of radioactivity and decay, preparation of radioisotopes for tracers, applications with radiotracers, radiometric titration, radioactivity measurements by gas filled and scintillation detectors.	6
6.	Solvent extraction . Partition law and its limitations, distribution ratio, separation factor, factors influencing extraction, multiple extractions. Extraction of metal chelates.	5
7.	Basic chromatography: Introduction and classification, theory of column chromatography, retention time, retention volume, capacity factor, concept of plate and rate theory, resolution, column performance, normal and reverse phase chromatography, paper and thin layer chromatography, ion-exchangers.	9
	Total	42

S. No.	Name of Authors/Books/ Publishers	Year of Publication/ Reprint
1.	Ewing, G.W., "Instrumental Methods of Chemical Analysis", 5th Ed. McGraw Hill.	2004



2.	Mendham, J., Denny, R.C., Barnes, J.D. and Thomas, M.J.K., "Vogel's Text	2004
	Book of Quantitative Chemical Analysis", 6th Ed. Pearson Education.	
3.	Christian, G.D., "Analytical Chemistry" 6 th Ed. Wiley	2008
4.	Sood, D.D., Reddy, A.V.R. and Ramamoorthy, N., "Fundamentals of	2004
l	Radiochemistry", IANCAS, BARC, Mumbai.	



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-507

Course Title: Structure and Reactivity of Organic Molecules

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs)

Theory: 3

Practical: 0

4. Relative weightage:

CWS: 25 PRS: 0

MTE: 25 ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective of Course: To impart the knowledge of structure-reactivity and the reaction mechanism.

S.	Contents	Contact
No.		Hours
1.	Conformational analysis of cyclic systems: Cyclohexane and its derivatives (mono-, di- and tri-substituted), fused (decalins) and bridged bicyclic systems, dynamic stereochemistry, conformational rigidity and mobility, quantitative correlation between conformation and reactivity, effect of conformation on the reduction of cyclic ketones, nucleophilic addition to carbonyl group (Cram, Karabatsos, Felkin-Ahn models, Cieplak effect), nucleophilic substitution on cyclohexane substrates, cyclohexane epoxide formation and opening, elimination reactions of cyclohexyl halides, acetate esters and related compounds, deamination of 2-amino-cyclohexanols, elimination vs substitution competition and neighboring group participation reactions of acyclic and cyclic molecules.	12
2.	Physical organic chemistry: Basic concepts, thermodynamic and kinetic requirements, rate and equilibrium constants, reaction coordinate diagram, transition state (activated complex), nature of activated complex, Hammond postulate, reactivity vs selectivity principle, Curtin-Hammett principle, microscopic reversibility, kinetic	8
3.	Methods for elucidating mechanism: Kinetic analyses of simple and complex reactions, steady state and saturation kinetics, isotope effects – primary and secondary isotope effects, steric and equilibrium isotope effects, solvent isotope effects, heavy atom isotope effects, substituent effects – origin (inductive, field, resonance, steric, solvent and polarizability). Hammett linear free energy relationship, substituent parameter (sigma), reaction constant (rho), use of Hammett plot for mechanism determination, deviation from linearity, inductive vs resonance effects -Taft parameters, nucleophilicity and nucleofugality, factors affecting nucleophilicity (basicity/acidity, solvation, polarizability and shape), Swain-Scott parameters, Edwards and Ritchie correlations, solvent effects - bulk and specific solvent effects, Grunwald-Winstein plots, Bronsted relationships, experiments for identifying mechanism (example Cannizzaro reaction), product and intermediate identification, common intermediate detection (example Ritter reaction and Beckmann fragmentation), trapping and competition experiments, isotope labeling, crossover	
4.	experiments. Catalysis: Binding in transition state vs ground state, electrophilic catalysis, acid and Catalysis: Binding in transition state vs ground state, electrophilic catalysis, acid and Catalysis: (general and catalysis)	i 6
	base catalysis, nucleophilic, covalent, Bronsted acid base catalysis (general and specific, Bronsted catalysis law, Leffler law), Libido rule.	



Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication / Reprint
1.	Anslyn, E. V. and Dougherty, D. A., "Modern Physical Organic Chemistry", University Science Books.	2006
2.	Clayden, J., Greeves, N. and Warren, S., "Organic Chemistry", Oxford University Press.	2012
3.	Carey, F. A. and Sundberg, R. J., "Advanced Organic Chemistry", Part A: Structure and Mechanisms, 5 th Ed., Springer.	2007
4.	Nasipuri, D., "Stereochemistry of Organic Compounds: Principles and Applications", New Age International.	2014



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code:

CYN-509

Course Title: Coordination Chemistry

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS: 0 MTE: 25 ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective:

To impart basic and advanced concepts of coordination chemistry.

S. No.	Contents	Contact Hours
1.	Structure, bonding and properties of transition metal complexes: Different types of ligands and coordination geometry (symmetry considerations), coordination number, isomerism (recapitulation), HSAB concept, thermodynamic stability, successive and overall stability constants, determination of stoichiometry (Job's method) and stability constants by spectrophotometric, potentiometric and polarographic methods, Irving-William series, chelate and macrocyclic effect.	6
2.	Stereochemical aspects of coordination complexes: Stereoisomerism in inorganic complexes, isomerism arising out of ligand and ligand conformation, chirality and nomenclature of chiral complexes, optical rotatory dispersion (ORD) and circular dichroism (CD).	5
3.	Metal-ligand bonding: Overview of crystal field and ligand field theories of 4-, 5- and 6-coordinated complexes, d-orbitals splitting in linear, trigonal, octahedral, square planar, tetrahedral, square pyramidal, trigonal-bipyramidal and cubic complexes, measurement of CFSE (d¹ to d¹0) in weak and strong ligand fields, Jahn-Teller distortion, nephelauxetic series, variation of lattice energy, ionic radii and heat of hydration across 1 st row transition metal ions.	6
4.	Molecular orbital theory (MOT) of coordination compounds: Composition of ligand group orbitals, molecular orbital energy diagrams of octahedral, tetrahedral, square planar complexes including both σ and π bonding, angular overlap model.	7
5.	Electronic spectra of coordination compounds: Energy states from spectral terms of d ⁿ configurations, selection rules for ligand-field and charge transfer transitions in metal complexes, band intensities, factors influencing band widths, splitting of various terms, Orgel and Tanabe-Sugano diagrams of octahedral and tetrahedral d ⁿ complexes, calculation of ligand field parameters, luminescence, phosphorescent complexes.	7
6.	Magnetic properties of coordination compounds: Fundamental equations in molecular magnetism, magnetic susceptibility and magnetic moment, diamagnetic and paramagnetic behavior of transition metal complexes, spin-orbit coupling effects (L-S coupling and j-j coupling), orbital angular moment and its quenching in octahedral and tetrahedral complexes, temperature independent paramagnetism (TIP) of complexes, spin cross over phenomenon, spin admixed states, metal-metal direct spin interaction and super exchange spin-spin interaction through bridging	11

	ligands, ferromagnetic, anti-ferromagnetic, ferrimagnetic behaviour of transition	
	metal compounds, effect of temperature on their magnetic properties, single	
l.	molecule magnets.	
]	Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Cotton, F.A., Wilkinson, G., Murillo, C.A. and Bochmann, M., "Advanced Inorganic Chemistry", 6 th Ed., John Wiley & Sons.	1999
2.	Douglas, B.E., McDaniel, D.H. and Alexander, J.J., "Concepts and Models in Inorganic Chemistry", 3 rd Ed., John Wiley & Sons.	2001
3.	Figgis, B.N., and Hitchman, M.A "Ligand Field Theory and Its Applications", Wiley Eastern Ltd.	1999
4.	Huheey, J.E., Keiter, E.A. and Keiter, R.L., "Inorganic Chemistry Principle of Structure and Reactivity", 4 th Ed, Pearson Education, Inc.	2003
5.	Atkins, P., Overton, T., Rourke, J., Mark, W. and Armstrong, F., "Shriver and Atkins' Inorganic Chemistry", 4 th Ed, Oxford university press.	2009



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-511

Course Title: Laboratory-I

2. Contact Hours:

L: 0

T: 0

P: 12

3. Examination Duration (Hrs):

Theory

Practical 12

4. Relative Weightage: CWS:

0 PRS: 50

5. Credits: 6

MTE:

0 ETE:

0 PRE: 50

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart practical knowledge and skills in physical, inorganic and organic chemistry laboratories.

Contents	Contact Hours
List of Inorganic Chemistry Experiments	
 Semi-micro qualitative analysis involving 6 radicals including interfering radicals. Determination of metal ions by gravimetric-cum-volumetric analysis: Ag (I) gravimetrically and Cu(II) volumetrically. Determination of Cu(II) gravimetrically and Zn(II) volumetrically. Determination of Fe(III) gravimetrically and Ca(II) volumetrically. Gravimetric analysis of a mixture of two metal ions such as Cu and Zn. Synthesis of coordination compounds and metal content determination: (i) [Cu(NH₃)₄.H₂O]SO₄, (ii) [Fe(acac)₃], (iii) [Mn(acac)₃], (iv) [Mn(C₂O₄)₃]. 	4×14
Some experiments require two-three turns.	
List of Physical Chemistry Experiments	
 Determination of pK₁ and pK₂ of an acid using pH meter. Determination of cell constant and verification of Kohlrausch's law. Determination of specific rotation of lactic acid/sucrose by polarimeter. Determination of molar refraction equivalent to -CH₂, C, H, and O. Determination of composition of liquid mixture by refractive index measurements. Determination of dimerization constant of benzoic acid. Conductometric titration of different acids against bases. Verification of Ostwald's dilution law by conductometric measurements of acetic acid. Verification of Freundlich's adsorption isotherms and calculation of characteristic constants. Verification of Langmuir adsorption isotherms and determination of surface area. Determination of surface excess concentration and thickness of interfacial adsorbed layer by surface tension measurements of water-n-butanol mixture. Determination of the Parachor of binary mixture of miscible solute by surface tension measurements. 	4 × 14
13. Verification of Hardy-Schultze rule for positive/negatively charged colloids. 14. Determination of critical micelle concentration of sodium dodecylsulphate/cetyltri-	

	methylammonium bromide by surface tension method.	-
	List of Organic Chemistry Experiments	
1.	Separation of organic mixtures by TLC and PTLC.	
2.	Synthesis of derivatives for carbonyl, amino and active methylene compounds.	
3.	Diels-Alder reaction between anthracene and maleic anhydride.	
4.	Oxidation of hydroquinone to <i>p</i> -benzoquinone.	
5.	Oxidation of benzoin to benzil.	4 × 14
6.	Conversion of benzil to quinoxaline.	
7.	Reduction of camphor.	
8:	Synthesis of 2-iodobenzoic acid by Sandmeyer reaction	•
9.	Aldol condensation (benzaldehyde + acetone or cinnamaldehyde + acetone).	
10.	Synthesis of binaphthol by green reaction.	
11.	Knoevenagel condensation between aldehyde (4-diethylaminobenzaldehyde) and	Ì
	malonic acid, cyanoacetic acid or malononitrile.	
12.	Friedel-Crafts reaction: synthesis of 1,4-di-tert-butyl-2,5-dimethoxybenzene.	
13.	Preparation and purification of cis- and trans-stilbenes by Wittig reaction.	
14.	Bromination of acetanilide.	
	Total	168

S. No.	Authors/ Title/ Publisher	Year of Publication/Reprint
1	Mendham, J., Denney, R.C., Barnes J.D. and Thomas M.J., "Vogel's Text Book of Quantitative Chemical Analysis", 6 th Ed., ELBS Longman Group UK Ltd.	2004
2	Srivastava T.N. and Kamboj P.C., "Analytical Chemistry", Vishal Publications.	2000
3	Furniss B.S., Handford A.J., Smith P.W.G. and Tatchell A.R., "Vogel's Text Book of Practical Organic Chemistry", 5 th Ed., Longman.	1996
4	Leonard J., Lygo B. and Procter G., "Advanced Practical Organic Chemistry", Chapman & Hall.	1995
5	Levitt, B.P., "Findlay's Practical Physical Chemistry", 9th Ed., Longman	1973
6	Garland C.W., Nifler J.W. and Schoemaber D.P., "Experiments in Physical Chemistry", 7 th Ed., McGraw-Hill International.	2002



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-502

Course Title: Organometallics, Inorganic Chains and Clusters

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0 MTE: 25

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective: To impart advanced concepts in organometallic chemistry and structural aspects of inorganic chains, rings, cages and clusters.

S.	Contents	Contact
No.		Hours
1.	Structure and bonding in organometallics: 18 electron rule and its application to π -acceptor ligands, limitations of 18 electron rule, description of bonding models for π -acceptor ligands including CO, alkenes (Dewar-Chatt-Duncanson model) and tertiary phosphines, physical evidences and consequences of bonding. Main group organometallics: Introduction, review of comparative aspects of synthetic methods, reactivity and bonding in ionic, covalent, electron deficient and electron rich organometallic compounds. Kinetics and mechanism of ligand substitution (associative and dissociative), oxidative addition and reductive elimination, transmetallation, migratory insertions, reactivity at metal-bound ligands.	9
2.	Organotransition metal chemistry: σ -Bonded transition metal-alkyls, - aryls, - alkenyls(vinyls), -alkynyls(acetylides), reactions in σ -organyls: homolytic cleavage, reductive elimination, electrophilic cleavage, insertion, β -metal hydrogen elimination, α -abstraction or α -elimination and γ - and δ -remote C-H functionlization.	5.
3.	Organotransition compounds with multiple metal-carbon bonding: Transition metal-carbenes/-carbynes, -bridging carbenes/carbynes, reactions of carbene/carbyne complexes such as ligand substitution, nucleophilic, electrophilic attack, dismutation, and ligand coupling reactions.	4
4.	Organotransition compounds with multicenter bonds: Concept of hapticity, transition metal complexes of alkenes, Ziese salt, alkynes, allyls, butadienes; π -metal complexes of cyclobutadienes, cyclopentadienyls, arenes, cycloheptatrienyls and cyclooctatetraenes, reactions and bonding in ferrocene; stereochemical non-rigidity in organometallic compounds and fluxionalty, bimetallic and cluster complexes.	6
5.	Inorganic chains, rings, cages and clusters: Chains— catenation and heterocatenation, structural aspects of silicate minerals and silicones, one-dimensional conductors: (SN) _x chains, chalcogenide glasses, iso- and heteropolyanions. Rings—borazines, boron nitride, phosphazenes—structural models, phosphazene polymers, and other homocylic and heterocyclic inorganic ring systems. Cages—Boron cage compounds—structural aspects (boranes-styx number and Wade's rule) of higher boranes, carboranes, metallacarboranes, phosphorous cage compounds with P—P, P—O, P—S. Clusters—metal clusters, metal carbonyl clusters, di-, tri-, tetra- and	14

	hexanuclear clusters.	
6.	Applications of organometallics and clusters in catalysis: Alkene metathesis,	4
	Cativa and Monsanto processes for production of acetic acid, carbonylation and	
	decarbonylation reactions, Wacker process, cyclooligomerisation of acetylene using	
	Ni/Cr catalysts, Mobil and Fischer-Tropsch processes, polymer-bound catalysts,	
	metal carbonyl clusters in catalysis.	
	Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Huheey, J.E., Keiter, E.A. and Keiter, R.L., "Inorganic Chemistry Principle of Structure and Reactivity", 4 th Ed, Pearson Education Inc.	2003
2.	Cotton, F.A., Wilkinson, G., Murillo, C.A. and Bochmann, M., "Advanced Inorganic Chemistry", 6 th Ed., John Wiley & Sons.	1999
3.	Hill, A.F., "Organotransition Chemistry", The Royal Society of Chemistry, Cambridge.	2002
4.	Bochmann, M. (Ed.), "Oxford Premier Series on Organometallics", Vol. 1 and 2, Oxford Press.	2002
5.	Gupta, B.D. and Elias A.J., "Basic Organometallic Chemistry", 2 nd Ed., University Press (India) Pvt. Ltd.	2013



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-504

Course Title:

Kinetics and Photochemistry

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs) Theory: 3

Practical:

4. Relative weightage: CWS: 25

MTE: 25

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Spring

PRS: 0

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective of Course: To impart basic knowledge of the kinetics and photochemistry.

10. Details of Course:

S.No.	Contents	Contact Hours
1.	Theories: Theoretical calculation of energy of activation using potential energy surface diagram, absolute reaction rate theory, comparison between gas phase and solution reactions.	10
2.	Type of reactions: Kinetics of chain reactions, detections of radical and kinetics of HBr, H ₂ O ₂ reactions, explosion limits, elementary idea of unimolecular reactions, application of following to the reaction kinetics—solvent effect, kinetic isotope effect and salt effect, experimental technique for studying the fast reaction kinetics, kinetics of homogenous and heterogenous catalysis, kinetics of polymerization.	14
3.	Electron transfer dynamics: Electron transfer in homogeneous systems, theory of electron transfer processes, electron tunneling, experimental results, electron transfer in heterogeneous systems, electrode-solution interface, rate of charge transfer in electrode reactions, study of kinetics of electrode processes.	8 .
4.	Photochemistry: Quantum efficiencies of photochemical and photophysical processes, experimental techniques for continuous photolysis. Primary and secondary photochemical processes, Franck-Condon principle and its applications, rates of absorption and emission, lifetimes of electronically excited states and their fate, quenching of excited states species — dynamic and static quenching, radiationless transition and pre-dissociation, energy transfer processes, FRET analysis, mechanistic analysis and reaction dynamics. Radiation chemistry—Interaction with ionizing radiation with matter, dosimetry, generation of free radicals and intermediates, comparison between photo- and radiation chemistry.	10
	Total	42

S.No.	Name of Authors/ Books/ Publishers	Year of Publication/Reprint
1.	Laidler, K.J., "Reaction Kinetics", Anand Sons, New Delhi.	2005
2.	Amis, E.S., "Solvent Effect of Reaction Rates and Mechanism", Academic Press.	2005
3.	Mukherjee, K.K., "Fundamentals of Photochemistry", New Age	2004



Γ	International Pvt. Ltd., New Delhi.	
	International Fvt. Ltd., 1909 Dent.	2003
4	Lakowicz, J.R., "Principles of Fluorescence Spectroscopy", Plenum	2003
1	Press, New York.	
<u></u>	11035, 100 Tolk.	1998
5.	Wishart, J.F. and Nocera, D.G., "Photochemistry and Radiation	1990
	Chemistry", Oxford University Press, USA.	



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-506

Course Title: Organic Reaction Mechanisms

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical:

4. Relative weightage:

CWS: 25

PRS: 0 MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester:

Spring

7. Subject Area: PCC

8. Pre-requisite: Nil

9. Objective of Course: To familiarize the students with different types of organic reactions.

S. No.	Contents	Contact Hours
1.	Single bond [C-X (X = C, O, N)] formations: Nucleophilic additions to carbonyls and stereochemical aspects through various models (Cram, Cram chelation and Felkin-Anh models), chemistry of enolates (kinetic and themodynamic) and enamines, enolates, lithium and boron enolates in aldol and Michael reactions, alkylation and acylation of enolates, mechanism of aldol (Mukiyamaaldol), Knoevenagel, Claisen, Dieckmann, Perkin, Stobbe, Darzen, Acyloin condensations, organolithium, organomagnesium (Grignard), organozinc, organocopper (Gilman & Normant) reagents in synthesis, epoxidations (Prilezhaev, Sharpless, Jacobsen and Shi), Metal catalyzed C-C bond formations (Ullmann, Buchwald-Hartwig, Sonogashira, Heck, Suzuki, Stille, Nozaki-Hiyama and Kumada reactions).	14
2.	Multiple bond [C-X (X = C, N)] formations: Phospohorus, nitrogen and sulfur ylids, Wittig reaction, Wittig-Horner reaction, Tebbe olefination, Julia olefination, Robinson annulation, Mannich reaction, Peterson olefination, Ramberg-Backlund rearrangement, McMurry reaction, Shapiro reaction, β -eliminations (Hoffman & ester pyrolysis), Cope elimination, selenoxide elimination, dehydration of alcohols, Corey-Winter reaction, olefins from epoxides, reduction of acetylenes, olefin metathesis (Schrock's catalyst, Grubbs' catalyst), ring closing metathesis, enyne metathesis, Thorpe reaction, Corey-Fuchs reaction, Seyferth-Gilbert homologation, Ohira-Bestmann modification.	
3.	Pericyclic reactions: Classification, electrocyclic, sigmatropic, cycloaddition, chelotropic and ene reactions, conservation of orbital symmetry, state correlation diagrams, frontier molecular orbital (FMO) theory, aromatic transition state (ATS) theory, generalized orbital symmetry (GOS) rule, photochemical cycloaddition reactions, Diels-Alder reaction, Dipolar cycloadditions, retrocycloadditions, electrocyclic reactions, conrotation and disrotation, orbital correlation diagrams for four-, six- and eight-electron cases, torquoselectivity, sigmatropic rearrangements, [1,3], [1,5] and [3,3] shifts, Curtius and Schmidt rearrangements, Cope and Claisen rearrangements, di-π-methane rearrangement, ene reaction, cheletropic reactions, Norrish I and II reactions.	
	Norrish I and II reactions. Total	42





S. No.	Name of Authors/Books/Publishers	Year of Publication / Reprint
1.	Carey, F. A. and Sundberg, R. J., "Advanced Organic Chemistry, Part B: Reactions and Synthesis", 5 th Ed., Springer.	2007
2.	Anslyn, E. V. and Dougherty, D. A., "Modern Physical Organic Chemistry", University Science Books.	2006
3.	Clayden, J., Greeves, N. and Warren, S., "Organic Chemistry", Oxford University Press.	2012
4.	Smith, M.B., "Organic Synthesis", 3rd Ed., Academic Press.	2010
5.	Bruckner, R., "Organic Mechanisms: Reactions, Stereochemistry and Synthesis", Springer.	2010



NAME OF DEPTT/CENTRE: DEPARTMENT OF CHEMISTRY

CYN-508 1. Subject Code:

Course Title: Molecular Spectroscopy

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

ETE: 50

4. Relative Weightage: CWS: 25 PRS: 0

MTE: 25

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: CYN-501; Elementary knowledge of symmetry and group theory.

9. Objective: To provide advance knowledge of spectroscopic techniques for identification and elucidation of structures of molecules.

S.	Contents	Contact Hours
No. 1.	Overview of molecular spectroscopy: Different aspects of molecular spectroscopy, the Born-oppenheimer approximation, transition probability, oscillator strength, the integrated absorption coefficient. Microwave spectroscopy: Classification of the rotors, intensity of the rotational lines, population of energy levels, non-rigid rotation, anharmonicity and centrifugal distortion, effect of isotopic substitution. Rotation spectra of the linear, spherical top and asymmetric top polyatomic molecules, microwave technique.	6
2.	Infrared and Raman spectroscopy: Types of Violation Canada Combination bands, Fermi resonance phenomenon, the finger print region, FTIR spectroscopy and application. Rayleigh and Raman scattering, polarizabilities, rotational and vibrational Raman spectra, selection rules, polarization of the light and	6
3.	UV-visible spectroscopy: Electronic spectra, Frank-Condon Frinciple, piculasociation spectra, Fortrat diagram, conjugated polyene and enone systems, different types of charge transfer transitions and their basis. Charge transfer spectra in organic and	5
4.	Photoelectron spectroscopy: The photoionization processes, Auger and autoionization processes, deexcitaion by fluorescence, outlines of UPS, XPS and Auger techniques and their applications in interpretation of valence and core shell spectra of atoms and molecules.	4
5.	Magnetic resonance spectroscopy: Nuclear moments, nuclear spin states in a magnetic field and the resonance phenomenon, relaxation processes and their importance. Bloch equation, Larmor frequency, shielding constant and chemical shifts. Spin-spin coupling and quantum chemical description of spin systems—spectra of two (AX cases) and three (AMX) spin systems. Dynamic NMR and line shapes. General introduction to double resonance experiments and nuclear overhauser effect, chemical shift reagents multinuclear NMR. ESR Spectroscopy: Principle of ESR and interpretation of its applications for ESR	14
6.	spectra of -CH ₃ and -CH ₂ radicals. Mössbauer spectroscopy: The Mössbauer effect and chemical isomer shift. Magnetic and quadrupolar effects in the Mössbauer spectra of complexes of ⁵⁷ Fe and other nuclei. Tota	c .
	and quadrupolar effects in the Mossbauer spectra of complexes of 10 and Total	1 42



S. No.	Name of Authors/ Books/ Publishers	Year of Publication/Reprint
1.	Banwell, C.N. and McCash, E.L.M., "Fundamentals of Molecular Spectroscopy", 4 th Ed. McGraw-Hill N. Y.	1999
2.	Slichter, C.P., "Principles of Magnetic Resonance", Springer Verlag.	1981
3.	Graybeal, J.D., "Molecular Spectroscopy", McGraw-Hill.	1988
4.	Atkins, P. and Paula, J.de, "Physical Chemistry", 7th Ed., Oxford Univ. Press.	2003
5.	Drago, R.S., "Physical Methods in Inorganic Chemistry", Reinhold Publishing Corp., East West Press Ltd.	1986



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-510

Course Title: Laboratory-II

2. Contact Hours:

T: 0

P: 12

3. Examination Duration (Hrs):

Theory

Practical: 12

4. Relative Weightage: CWS:

0 PRS: 50

MTE:

0 ETE: 0 PRE: 50

5. Credits: 6

6. Semester: Spring

7. Subject Area: PCC

8. Pre-requisite: CYN-511

9. Objective: To impart practical knowledge and skills in physical, inorganic and organic chemistry laboratories.

	Contents	Contact
		Hours
	List of Inorganic Chemistry Experiments	
1.	Comparison of the electronic spectra of [Ni(H ₂ O) ₆] ²⁺ , [Ni(NH ₃) ₆) ²⁺ and [Ni(en) ₃] ²⁺ and qualitative verification of the spectrochemical series, and quantitative estimation of nickel by spectrophotometry.	
2.	Synthesis and spectrophotometric study of copper complexes: (i) synthesis of bis(salicylaldimine)copper(II) and cis-bis(glycinato)copper(II), (ii) record the spectra of bis(salicylaldimine)copper(II) and cis-bis(glycinato copper(II), and (iii) record spectra of Cu ²⁺ in water, NH ₃ , ethylene diamine and glycine, and arrange the ligands in order of increasing field strength and (iv) quantitative estimation of	4×14
3.	copper by spectrophotometry. (i) Study of the complex formation between Fe(III) and thiocyanate/salicylic acid/sulphosalicylic acid or between Ni(II) and o-phenanthroline, and (ii) spectrophotometric determination of formation constant of the complex (Job's method and molar ratio method).	
4.	Preparation of (i) [Ni(NH ₃) ₆]Cl ₂ /SO ₄ , (ii) [Ni(en) ₃]Cl ₂ /SO ₄ , (iii) bis(salicylaldimine)nickel(II), and analysis by different methods, viz. IR, UV-visible and ¹ H NMR spectroscopy.	
5. 6.	potassium tris(oxalato)ferrate, and their characterization by metal determination, various spectroscopic (I.R. and U.VVis) methods, magnetic moment determination, and photochemical behavior of iron complex. Synthesis and characterization of [Co(en) ₃]Cl ₃ . Separation of its optical isomers	
	and determination of their optical rotation by using polarimeter. Some experiments require two-three turns.	
•	List of Physical Chemistry Experiments	
	. To study the kinetics of H ⁺ catalyzed hydrolysis of an ester.	
2	. To study the kinetics of saponification of an ester.	
	. To study the kinetics of Fe ³⁺ catalyzed decomposition of hydrogen peroxide.	
4	To study the kinetics of inversion of sucrose using polarimeter.	4×14
6	 Determination of equilibrium constant of Kl₃ complex by distribution method. Determination of compound formation between liquids by viscosity variation with composition of mixtures of liquids using Ostwald viscometer. 	

 7. Determine the composition of KCl-KBr mixtures by potentiometric titration against silver nitrate solution. 8. Verification of Beer-Lambert's law using potassium permanganate solution. 9. To study the quenching of fluorescence of organic dye(s). 10. To study the variation in miscibility of phenol in water with temperature and to find out the critical solution temperature (CST) and also to investigate the effect of impurity on CST. 11. To determine the molecular weight of a volatile substance using Victor Meyer method. 12. To determine the cell potentials for different electrochemical cells and also to measure different thermodynamic parameters. Some experiments require two turns. 	
List of Organic Chemistry Experiments	
 Preparation of p-nitroaniline from acetanilide. Preparation of pyridinium dichromate and its use in oxidation of benzyl alcohol. Cannizzaro reaction of an aromatic aldehyde (p-nitrobenzaldehyde). Synthesis of ω-nitrostyrene from an aromatic aldehyde and nitromethane. Synthesis of chalcone from an aromatic aldehyde and acetophenone. Extraction of oils from ground nuts using soxhlet apparatus. Synthesis of α-bromo cinnamic acid or phenyl acetylene from benzaldehyde, (formation of cinnamic acid, bromination and elimination reactions). Preparation of meso-stilbene dibromide and its conversion to diphenylacetylene. Fisher indole synthesis. 	4×14
Some experiments require two-three turns.	4.60
Total	168

S. No.	Authors/ Title/ Publisher	Year of Publication/Reprint
1	Mendham, J., Denney, R.C., Barnes J.D. and Thomas M.J., "Vogel's Text Book of Quantitative Chemical Analysis", 6 th Ed., ELBS Longman Group UK Ltd.	2004
2	Srivastava T.N. and Kamboj P.C., "Analytical Chemistry", Vishal Publications.	2000
3	Furniss B.S., Handford A.J., Smith P.W.G. and Tatchell A.R., "Vogel's Text Book of Practical Organic Chemistry", 5 th Ed., Longman.	1996
4	Leonard J., Lygo B. and Procter G., "Advanced Practical Organic Chemistry", Chapman & Hall.	1995
5	Levitt, B.P., "Findlay's Practical Physical Chemistry", 9th Ed., Longman	1973
6	Garland C.W., Nifler J.W. and Schoemaber D.P., "Experiments in Physical Chemistry", 7 th Ed., McGraw-Hill International.	2002



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject code: CYN-5 12

Course Title: Nuclear and Radiochemistry

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0

MTE: 25

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

ETE: 50

8. Pre-requisite: Fundamental knowledge in physical chemistry.

9. Objective: To impart basic knowledge in nuclear chemistry and apply its concepts for characterization of materials and analysis.

S. No.	Contents	Contact Hours
1.	Properties of nucleons and nuclei: Nuclear stability and natural radioactivity, size and shape of a nucleus, nuclear spin, magnetic properties of a nucleus, nuclear magnetic resonance, nuclear resonance or recoil-less absorption, electric quadrupole moment, nuclear parity and nuclear statistics.	6
2.	Nuclear models: Shell model, liquid drop model, Fermi gas model, collective model and optical model.	4
3.	Radioactivity: Radioactive decay, decay kinetics, parent daughter decay growth relationship, concepts of transient and secular equilibrium, alpha, beta and gamma decay, artificial radioactivity.	6
4.	Nuclear reaction: Bethe's notation, types of nuclear reaction, reaction cross section, Q- value and threshold, compound nucleus theory, transuraniens, photo and thermonuclear reaction, fusion reactor, nuclear fission, fission fragments and mass distribution, fission energy, theory of nuclear fission, introduction to nuclear reactor and importance of four factor formula, reprocessing of spent fuels, nuclear waste management.	8
5.	Interaction of radiation with matter: Stopping power and range for charged particles, interaction with X-rays and gamma rays – photoelectric effect, Compton scattering, pair production, nuclear detectors using semiconductors, effect of radiation on polymers and inorganic compounds, concepts of positron and its use in materials science.	10
6.	Ion beam techniques: Activation analysis with case studies, radioisotopes for nuclear medicine, elemental mapping of trace and major elements, depth wise analysis by RBS, elastic recoil deflection analysis for hydrogen measurement, ion implantation, ion beam mixing.	8
	Total	42



S. No.	Name of Authors/Books/ Publishers	Year of Publication/ Reprint
1.	Loveland, W., Morrissey and D., Seaborg, G.T., "Modern Nuclear Chemistry", John Wiley and Sons, New Jersey.	2006
2.	Arnikar, H.J., "Essentials of Nuclear Chemistry", New Age International (P) Ltd., New Delhi.	2003
3.	Friedlander, G., Kennedy J.W., Miller, E.S. and Macais, J.M., "Nuclear and Radiochemistry", John Wiley and Sons, Inc. New York.	1981



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-514

Course Title: Heterocyclic Chemistry

2. Contact Hours:

L: 3

T: 0

3. Examination Duration (Hrs)

Theory: 3

P: 0 Practical: 0

4. Relative weightage: CWS: 25

PRS: 0

MTE: 25

ETE: 50 PRE:

5. Credits: 3

6. Semester:

Spring

7. Subject Area:

PEC

8. Pre-requisite: Basic organic chemistry and synthetic methods.

9. Objective of Course: To impart a broad understanding of the major classes of 5- and 6-membered ring heterocyclic compounds.

S. No.	Contents	Contact Hours
1.	Heterocycles: Systematic nomenclature of heterocyclic compounds (Hantzsch-Widman, replacement and fusion methods), biological importance of heterocyclic compounds.	2
2	Five-membered heterocycles with one heteroatom: Chemical structures of furan, pyrrole and thiophene, and degree of aromaticity. General syntheses methods for 5-member rings. Paal-Knorr, Feist-Benary, Hantzsch and Knorr syntheses. Electrophilic substitution, reactants employed and orientation of the substituent on the ring.	10
3	Benzo derivatives of five-membered heterocycles with one heteroatom: Preparation of indole and carbazole derivatives. Fisher, Bischler, Madelung and Reissert syntheses. Preparation and reactivity of benzofurans (coumarins), benzothiophenes, dibenzofurans and dibenzothiophenes.	8
4.	Pyridines, quinolines and isoquinoles: Influence of the imine group on the reactivity of the pyridine ring. Nucleophilic and electrophilic substitutions on pyridine, quinolines and isoquinolines. Comparison of reactivity with benzene and naphthalene. Preparation of pyridine salts and pyridine N-oxides and synthetic applications. Skraup, Friedlander, Pfintzinger Bischler-Napieralski and Pictet syntheses.	10
5.	Heterocycles with 5 or 6 members and two or three heteroatoms: Syntheses and reactivity of oxazoles, thiazoles, oxadiazoles, thiadiazoles, benzothiazoles, benzothiadiazoles, triazole, benzotriazole, pyrimidines, pyrazines, quinoxalines and triazines	6
6.	New materials derived from heterocycles: Syntheses of indigo, mauveine, cyanines, tetrathiafulvalenes and related dyes, organic sensitizers for DSSC, electron donors and acceptors for organic solar cells, optical chemosensors and organic semiconductors for thin-film transistors.	
	Total	42



S.No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Gilchrist, T. L., "Heterocyclic Chemistry", 3 rd Ed., Pearson Education, India. (ISBN: 978-0582278431).	2007
2.	Sainsbury, M., "Heterocyclic Chemistry", Wiley. (ISBN: 978-0-471-28164-1)	2002
3.	Katritzky, A. R., Ramsden, C. A., Joule, J. A. and Zhdankin, V. V., "Handbook of Heterocyclic Chemistry", 3 rd Ed., Elsevier. (ISBN: 978-0-08-095843-9)	2010
4.	Gupta, R. R., Kumar, M. and Gupta, V., "Heterocyclic Chemistry, Vol. I: Principles, Three- and Four-Membered Heterocycles", Springer. (ISBN: 978-3-642-72278-3)	1998
5.	Gupta, R. R., Kumar, M. and Gupta, V., "Heterocyclic chemistry, Vol. II: Five-Membered Heterocycles", Springer. (ISBN: 978-3-642-08460-7)	1998
6.	Joule, J. A. and Mills, K., "Heterocyclic Chemistry", Wiley-Blackwell. (ISBN: 978-1-4051-3300-5)	2010



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-516

Course Title: Chemistry of Main Group and Transition Elements

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS: 0

MTE: 25

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

8. Pre-requisite: Basic knowledge of inorganic chemistry.

9. Objective: To impart advance concepts in chemistry of main group and transition elements.

S.	Contents	Contact Hours
No.		14
1.	Chemistry of main group elements: (i) Chemistry of Gr. 1 and 2 metals—solutions of alkali metals in liquid ammonia (reactions, electrical and magnetic properties), metal anions, complexation of Gr. I/II metals with crown ethers, cryptands and calixarenes, anomalous behavior of Li and Be. (ii) Chemistry of p-block elements—borides, borates and boran halides, allotropes of carbon (diamond, graphite, fullerene, carbon nanotubes), carbides, chlorofluorocarbons, silicon halides, silanes, silanols. Hydrides, oxides and oxoacids of pnictogens (N, P), chalcogens (S, Se and Te) and halogens. Pseudohalogens, inter-halogens, polyhalide anions, synthesis, structures and reactivity of compounds of xenon, bonding in xenon fluorides. Overall structural and bonding aspects (VBT) of B, Al, Si, N, P and Cl compounds.	
2.	Types of non-covalent interactions: Inter- and intramolecular hydrogen bonding interactions and their effects, electrostatic interactions (ion-ion, ion-dipole, dipole-dipole, dipole-induced dipole) and other weak intermolecular forces. Principle of self-assembly, host-guest chemistry and molecular receptors, examples of supramolecular inorganic architectures, and supramolecular photochemistry. Metallic Bonding—band model, soft X-ray spectra and N(E) curves, binding energy in metals, conductors, semiconductors and insulators, effect of temperature and impurity on conductivity.	
3.	Concept of acid and bases: Bronsted and Lewis acids and bases, pH, pKa, acid base concept in non-aqueous media, buffer solution, protonic acids, proton affinities, differentiation and leveling solvents, Hammet Ho scale, acidic behavior of the binary hydrides, cosolvating agents, oxyacids. aquoacids, amphoteric oxides, nonprotonic concepts of Acid - Base reactions, Lux concept, solvent ion theory of acids and bases, non-aqueous solvents—liquid ammonia, acetic acid, BF ₃ . Hard and soft acids and bases, symbiosis, theoretical and electronegativity concepts of hardness and softness.	
4.	Lanthanides and actinides: Separation and isolation of lanthanides, separation of Np, Pu and Am from U, electronic spectra and magnetic properties of lanthanides and actinides, general comparison of lanthanides and actinides and their applications in technology, lanthanide shift reagents.	
5.	Introductory bioinorganic chemistry: Inorganic composition of cells, compartmentalization, classification of biomolecules, biological metal-coordination	1

sites including special ligands like porphyrins and pterin. Role of metallobiomolecules in transport, transfer and transcription (preliminary ideas selective transport and storage of iron. Chemistry of elements in medicine—chelation), n
therapy, cancer treatment, imaging agents, anti-arthritis agents, radioisotopes and contribution of individual elements. Electron transfer proteins - active site structure and functions of ferredoxin rubridoxin and cytochromes, and their comparisons. Mechanism of nitrogen	n,
fixation. Total	al 42

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Huheey, J. E., Keiter, E. A. and Keiter, R. L., "Inorganic Chemistry Principle of Structure and Reactivity", 4 th Ed, Pearson Education, Inc.	2003
2.	Douglas, B.E., McDaniel, D.H. and Alexander, J.J., "Concepts and Models in Inorganic Chemistry", 3 rd Ed., John Wiley & Sons.	2001
3.	Cotton, F.A., Wilkinson, G., Murillo, C.A. and Bochmann, M., "Advanced Inorganic Chemistry", 6 th Ed., John Wiley & Sons.	1999
. 4.	Atkins, P., Overton, T., Rourke, J., Mark, W. and Armstrong, F. "Shriver and Atkins' Inorganic Chemistry", 4 th Ed, Oxford University Press.	2009



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-518

Course Title: Structure, Bonding and Properties of Solids

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory 3 Practical

4. Relative Weightage: CWS:

25 PRS:

25 ETE:

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart fundamental understanding of crystal structure, bonding and physical properties of solids.

MTE:

SI. No.	Contents	Contact Hours
1.	Symmetry in the crystalline state: Crystal symmetry, elements of translation- screw axis and glide planes, symmetry in a cube, crystal classes, stereographic projection of crystal systems, space symmetry and space groups, Hermann- Mauguin and Schoenflies notations, Wyckoff positions, representation of monoclinic and orthorhombic space groups.	
2.	X-Ray diffraction: Crystal planes and directions, Bragg's law in reciprocal space and Ewald sphere, structure factor, integrated intensity and systematic absences/presences, indexing and simulation of powder X-ray diffraction patterns for simple systems.	8
3.	Crystal chemistry: Hard sphere model, structures derived from HCP and CCP packing, crystal structures of various compositions, derived structures and polytypes, non-stoichiometry in solids, atomic order/disorder in solids, single crystals, polycrystals, quasicrystals, amorphous / glassy solids.	8
4.	Bonding in solids: Bonding in molecular solids – polymorphism, bonding in extended solids - ionic, covalent and metallic. Band theory of solids - classification of semiconductors, metals and insulators, free electron theory, Bloch's theorem, concept of density of state and elementary band theory, band structures of one-, two- and three- dimensional solids, selected metals and insulators.	8
5.	Properties of solids: Thermal, electrical, magnetic and dielectric properties of solids.	10
	Total	42



Sl. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	West, A. R., "Solid State Chemistry and its Applications", Reprint, Wiley India.	2013
2.	Rao, C.N.R. and Gopalakrishnan, J., "New Directions in Solid State Chemistry", 2 nd Ed., Cambridge University Press.	1997
3.	Stout, G.H. and Jensen, L.H., "X-Ray Structure Determination: A Practical Guide", 2 nd Ed., Wiley-Interscience.	1989
4.	Giacovazzo, C., Artioli, G. and Monaco, H. L., "Fundamentals of Crystallography", Oxford University Press.	2006
5.	S. Nicola, "Magnetic Materials: Fundamentals and Device Applications", Cambridge University Press.	2003
6.	Cox, P. A., "The Electronic Structure and Chemistry of Solids", Oxford University Press.	1987



T: 0

MTE:

NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-601

Course Title: Laboratory-III

2. Contact Hours:

L: 0

P: 8

3. Examination Duration (Hrs):

Theory

Practical 8

4. Relative Weightage: CWS: 0 PRS: 50

0 ETE: 0 PRE: 50

5. Credits: 4

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: CYN-511 and CYN-510

9. Objective: To impart practical knowledge and skills in handling instruments in physical, inorganic and organic chemistry laboratories.

10. Details of Course:

	Contents	Contact Hours
7. 8. 9. 10 1	 K₂Cr₂O₇ in a given mixture. Determination of Na, K in a soil sample by flame photometry. Determination of metal in alloy samples by AAS. Radiation measurement by GM counter. Determination of metal ions by polarography and voltammetry methods. Extraction of Fe³⁺ using 8-hydroxyquinoline. Catalytic oxidation of organic substrates and analysis by GC. 	8×14
	Total	112

S.	Authors/ Title/ Publisher	Year of Publication/Reprint
No.	Mendham, J., Denney, R.C., Barnes J.D. and Thomas M.J., "Vogel's	2004
1	Mendham, J., Denney, R.C., Barnes J.D. and Thomas Wist,	
	Text Book of Quantitative Chemical Analysis", 6th Ed., ELBS Longman	
	Crown IIV I td	i
2	Ewing G.W., "Instrumental Methods of Chemical Analysis", 5th Ed.,	2004
	McCuary Uill	
3	Furniss B.S., Handford A.J., Smith P.W.G. and Tatchell A.R., "Vogel's	1770



	Text Book of Practical Organic Chemistry", 5th Ed., Longman.	
4	Garland C.W., Nifler J.W. and Schoemaber D.P., "Experiments in	2002
	Physical Chemistry", 7th Ed., McGraw-Hill International.	
5	Levitt, B.P., "Findlay's Practical Physical Chemistry", 9th Ed., Longman	1973



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject code: CYN-603

Course Title: Advanced Analytical Techniques

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25 PRS: 0

MTE: 25

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart knowledge of specialized topics in analytical chemistry.

10. Details of the Course:

S. No.	Contents	Contact Hours
1.	Electroanalytical methods: Instrumentation, theory and applications of polarography—normal DC, pulse and differential pulse polarography, AC polarography. Linear and cyclic voltammetry, square wave voltammetry, coulometry at controlled potential, chronopotentiometry, anodic stripping voltammetry. Sensors—different types of solid and liquid sensors, nano material and chemically modified sensors, applications in environmental and biological sample analysis.	14
2.	Nuclear and X-ray methods: Radiotracers— choice and synthesis of radiotracers, isotope dilution methods, neutron activation analysis. Material science studies using positron emitters and Mössbauer source. Principles of X-ray spectra, X-ray absorption, emission fluorescence and diffraction methods and applications. Ion beam analysis, proton induced X-ray emission, Rutherford backscattering spectrometry, elastic recoil detection analysis for hydrogen measurement, nuclear microprobe.	14
3.	Mass spectrometry. Different types of ion sources, mass analyzers and detectors, resolution and resolving power, interpretation of mass spectra, hyphenated systems like LC-MS, GC-MS, MS-MS.	8
4.	Atomic spectroscopy and hyphenation Atomic absorption spectroscopy with electro thermal atomizers, cold vapor technique, inductively coupled plasma atomic emission methods – principle and instrumentation, hyphenation with mass spectrometry (ICP-MS), laser ablation and HPLC.	6
	Total	42

S.	Name of Authors/Books/ Publishers	Year of
No.		Publication/
		Reprint

1.		1990
2.	N.Y Sane, R.T. and Joshi, A.P., "Electroanalytical Chemistry: Theory and	1999
	Applications", Quest Publications. Loveland, W., Morrissey, D. and Seaborg, G.T., "Modern Nuclear	2006
	Chemistry", John Wiley and Sons, New Jersey.	
4.	Sood, D.D., Reddy, A.V.R. and Ramamoorthy N., "Fundamentals of Radiochemistry", IANCAS, BARC, India.	2004
5.	Skoog, D.A., Holler, F.J. and Crouch, S.R., "Principles of Instrumental Analysis", Thomson Brooks/Cole, Canada.	2007



NAME OF DEPTT/CENTRE: DEPARTMENT OF CHEMISTRY

Course Title: Separation Techniques and Microanalysis 1. Subject code: CYN-605

P: 0 T: 0 2. Contact Hours: L: 3

3. Examination Duration (Hrs): Theory: 3 ETE: 50 PRE: 0 MTE: 25 4. Relative Weightage: CWS: 25 PRS:

Practical: 0

7. Subject Area: PEC 6. Semester: Autumn 5. Credits: 3

8. Pre-requisite: Nil

9. Objective: To impart knowledge of specialized topics in analytical chemistry.

S. No.	Contents	Contact Hours
1.	Chromatographic techniques: Processes leading to non-ideal chromatography, extended form of Van Deemter equation, gas chromatography— instrumentation, types of column, stationary and mobile phases, detectors, Kovat's index, high pressure liquid chromatography—instrumentation, stationary and bonded-stationary phases, detectors, ion chromatography, size exclusion chromatography, supercritical fluid chromatography, affinity chromatography, electrophoresis.	12
2.	Liquid – liquid extraction: Principle, significance of various terms, batch and counter current extraction, classification of extraction systems, extraction equilibria of metal chelates, ion association extraction systems, extraction with high molecular weight amines, synergism, stripping, backwashing, salting out agents, masking agents, emulsion formation, identification of extracting species, analysis of organic phase, analysis of raffinate, environmental considerations, solid phase extraction.	12
3.	Membrane and allied methods of separation: Fundamentals and various terms, Liquid membranes, Cloud point extraction, Micellar enhanced separation processes, external field induced membrane separation processes.	4
4.	Sample preparation techniques for analysis: Extraction of semi-volatile organics from liquid, solid matrices, sample preparation for metal analysis, isoloation of nucleic acid for analysis, methods for preparing thin films for spectroscopic studies	
5.	Microfabrication, microanalysis and automation: Photolithography, electron and ion beam lithography, microfluidics for analytical techniques used in micro-total analytical System (μTAS), automation in analytical methods, instrumental parameters for automated devices, principles and techniques of automation for microanalysis with emphasis on sequences of operational modes in segmented and continuous flow, non destructive autoanalysers, applications in environmental and clinical cases.	

		Total	42
	•		

S. No.	Name of Authors/Books/ Publishers	Year of Publication/ Reprint
1.	Rousseau, R.W., "Handbook of Separation Process Technology", John Wiley & Sons.	1989
2.	Fifield, F.W. and Kealey, D., "Principles and Practice of Analytical Chemistry", 5 th Ed., Blackwell Science.	2000
3.	Li, N.N., Fane, A.G., Winston, W.S. and Matsuura, H. T., (Eds.), "Advanced Membrane Technology and Applications", Wiley.	2008
4.	Mitra, S. (Ed.), "Sample preparation techniques in Analytical Chemistry", John Wiley and Sons, New Jersey.	2003
5.	Madou, M.J.," Fundamentals of Microfabrications and Nanotechnology", 3 rd Ed., CRC Press.	2011



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-607

Course Title:

Electroanalytical Chemistry

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs)

Theory:

3

Practical: 0

4. Relative weightage: CWS: 25

MTE: 25 PRS: 0

PRE:

5. Credits:

6. Semester: Autumn

7. Subject Area:

PEC

8. Pre-requisite: Nil

9. Objective of Course: To introduce theoretical and practical knowledge of various electroanalytical systems.

10. Details of Course:

S.No.	Contents	Contact
		Hours
1.	Polarographic techniques: Operational amplifiers concept and design of polarographic circuit using op-amps. Ilkovic equation, theory of diffusion, kinetic, adsorption and catalytic currents. Controlled potential electrolysis and coulometry. 2 and 3 electrodes systems. Polarography versus voltammetry, determination of number of electrons. Theory of reversible, quasi-reversible and irreversible electrode processes. Pulse and differential pulse polarography and their superiority over DC polarography. A.C. polarography.	14
2.	Voltammetric techniques: Linear and cyclic sweep voltammetry, Randles Seveik equation, effect of sweep rate and evaluation of adsorption characteristics of reactant or product using CV. Coupled chemical reactions and their characterization. Characteristics of commonly used working electrodes such as glassy carbon, platinum, pyrolytic graphite and reference electrodes SCE and Ag/AgCl. Enzyme catalysed oxidations of biomoleules viz., uric acid, guanine, adenine and their comparison with electrochemical reactions. Anodic and cathodic stripping and determination of metal ions, pollutants and biomolecules using stripping	14
3.	voltammetry. Sensors: Amperometric and voltammetric sensors. Modified electrodes and their advantages over conventional electrodes in sensing variety of metals and biomolecules. Nanomaterials in electrode modification—C ₆₀ , single wall and multi wall carbon nanotubes. Preparation and characterization of modified surfaces. Applications of sensors in determining cases of doping.	
4.	Polarographic and cyclic voltametric studies of coordination compounds: Compounds containing one or more redox centers, coupled chemical reactions— EE and EEE mechanisms, stability constant of complexes.	7
	Total	42

	S.	Name of Authors/ Books/ Publishers	Year of Publication/	
ļ	No.		Reprint	



1.	Meites, L., "Polarographic Techniques", 3 rd Ed., Interscience publishers, N.Y.	1990
2.	Lund and Baizer, "Organic Electrochemistry", Marcel Dekker, New York.	2000
3.	Bard, A.J. and Faulkner, L.R., "Electrochemical Methods-Fundamentals and	2000
	Applications", John Wiley.	1999
4.	Sane, R.T. and Joshi, A.P., "Electroanalytical Chemistry: Theory and	1777
	Applications", Quest Publications.	



T: 0

NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-609

Course Title: Inorganic Biochemistry and Reaction Mechanism

2. Contact Hours:

L: 3

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative weightage:

CWS: 25 PRS: 0

MTE: 25 ETE: 50

PRE: (

11 11010111 1 1 1 1 2 2 2

. .

2. 20

5. Credits: 3

6. Semester: Autumn

7. Subject Area:

PEC

8. Pre-requisite: Knowledge of coordination chemistry.

9. Objective of Course: To familiarize the students with mechanisms of inorganic reactions and inorganic biochemistry.

S.No.	Contents	Contact Hours
1.	Inorganic reaction mechanism: Substitution reactions in octahedral complexes—exchange reactions, acid- and base-hydrolysis, annation reaction, solvolytic and catalysed reactions. Substitution reactions in square-planar complexes—effect of non-participation of ligands on reactivity, cis and trans effects.	6
2.	Electron transfer reactions: Outer- and inner-sphere mechanisms, factors affecting electron transfer reaction rates, theories of electron transfer reactions, solvated electron.	5
3.	Photochemistry of metal complexes: Introduction to inorganic photochemistry, photochemically excited states and excited state processes for transition metal complexes, photochemical reactions of coordination compounds (Cr and Ru complexes), types of photochemical reactions in transition metal complexes—substitution, decomposition, fragmentation, rearrangement and redox reactions. Mechanism of charge transfer (CT) to mass (CTTM) photoreduction. Applications of photochemical inorganic reactions in synthesis, catalysis, biological processes and in lasers.	6
4.	Inorganic biochemistry: Metalloproteins and enzymes—Role of metal ions in the active sites, structure and functions of metalloproteins and enzymes containing Mg, Ca, V, Mn, Fe, Co, Ni, Cu and Zn ions. Detailed structure and mechanistic studies of the following—Mn-photosystem-II, catalase, pseudocatalase, oxygen carriers, haemoglobin, myoglobin, non-porphyrin oxygen carriers, hemerythrin, hemocyanin, Fe-ribonucleotide reductase, cytochrome c oxidases, cytochrome P-450s, Ni-urease, hydrogenase, nitrogen fixation, Cu-blue copper protein, tyrosinase, galactose oxidase, superoxide dismutases, Zn-carbonicanhydrase, carboxypeptidase, alcohol dehydrogenase. Biological importance of Vitamin B ₁₂ and coenzymes, and their biomimetic studies.	
5.	Chemical toxicity and metallotherapy: Toxic chemicals in the environment, toxic effects of arsenic, cadmium, lead, mercury, carbon monoxide, cyanide and other carcinogens, metal containing drugs in therapy, interaction of heavy metal ions with DNA, DNA cleavage, structure-activity relationship and mode of action. Organometallic compounds as therapeutic drugs and enzyme inhibitors.	8
-	Total	42

S.No.	Name of Authors/Books/ Publishers	Year of Publication/ Reprint
1.	Huheey, J.E., Keiter, E. and Keiter, R., "Inorganic Chemistry: Principles of Structure and Reactivity", 4 th Ed., Pearson Education Asia, 3 rd Indian reprint.	2001
2.	Wilkins, R.G., "Kinetics and Reaction Mechanism of Transition Metal Complexes". 2 nd Revised Ed., VCH, New York.	1991
3.	Mukherjee, G.N. and Das, A., "Elements of Bioinorganic Chemistry", I'm Ed.,	1993
4.	Gillman, G., "Pharmacological, Basis of Therapeutic", 9th Ed., McGraw Hill.	1996
5.	Bertini, I., Gray, H.B., Lippard, S.J. and Valentine, J.S., "Bioinorganic Chemistry", University Science Books, U.S.A.	1994 -
6.	Lippard, S.J. and Berg, J., "Principles of Bioinorganic Chemistry", University Science Books, U.S.A.	1994



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-611

Course Title: Solid-State Chemistry and its Applications

2. Contact Hours:

L: 3

T: 0

р:

3. Examination Duration (Hrs):

Theory: 3

PRS: 0

Autumn

Practical: 0

4. Relative weightage: CWS: 25

MTE: 25

ETE: **50**

PRE: 0

5. Credits: 3

6. Semester:

7. Subject Area:

PEC

8. Pre-requisite: Knowledge of inorganic chemistry.

9. Objective of Course: To familiarize the students with crystal structures of common inorganic

compounds and their characterization methods.

S.No.	Contents	Contact
		Hours
1.	Crystal structure of inorganic compounds: Overview of close packing, packing efficiency, interstitial sites, limiting radius ratios, method of determination of ionic radii. Ionic crystals containing two or three different elements— FeO, ZnO, CdS, fluorite, antifluorite, nickel-arsenide, CaC ₂ , CdI ₂ and TiO ₂ , FeTiO ₃ , MgAl ₂ O ₄ , Fe ₂ NiO ₄ , garnets, BaTiO ₃ and KNiF ₃ . Non-ionic crystals— SiC, (BN)x, giant molecules, layer structures, crystals composed of discrete molecules.	11
2.	Defect structures: Thermodynamic defects and their consequences, solid electrolytes, non-stoichiometric compounds, F-centers and applications of defects in non-stoichiometric compounds.	3
3.	Methods to synthesize solid-state materials: Ceramic method, solid-state reaction and its kinetics, hydrothermal, sol-gel, co-precipitation (precursor), vapour phase transport methods. Different methods to grow single crystals.	6
4.	Amorphous inorganic materials: Glasses, refractories, materials obtained from organometallic chemical vapour deposition (MOCVD). New materials: Conducting polymers, carbon nanotubes, carbon nanorods and fullerenes. Electronic materials: Insulating, semiconducting and superconducting materials, ferroelectrics and dielectrics.	6
5.	Intercalation chemistry: Introduction, intercalation reactions in graphite, layered double hydroxides, layered sulfides, applications of intercalation chemistry.	3
6.	Mesoporous materials and their catalytic applications: Various types of mesoporous materials (oxides, sulphides), tailoring of pore size, applications of mesoporous materials in heterogeneous catalysis.	3
7.	Structural characterization of metal complexes by physical methods: Extended X-ray absorption spectroscopic (EXAFS), X-ray photoelectron spectroscopic (XPS), X-ray absorption near edge spectroscopic (XANES), electron spin spectrometric (ESR), electron spectroscopy for chemical analysis (ESCA) studies, soild state NMR, HMBC, HMQC, Mössbauer spectroscopic studies of metal complexes, thermal methods (TG, DTA and DSC), atomic force microscopy (AFM) and transmission electron microscopy (TEM).	. 10
	Total	42

S.No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Douglas, B.E., McDaniel, D.H. and Alexander, J.J., "Concepts and Models of Inorganic Chemistry", 3 rd Ed., John Wiley & Sons, Inc., New York.	2001
2.	West, A.R., "Solid State Chemistry and its Applications", Reprint, Wiley India.	2013
3.	Smart, L. and Moore, E., "Solid State Chemistry: An Introduction", Nelson Thornes Ltd.	2001
4.	Rao, C.N.R. and Gopalakrishnan, J. "New Directions in Solid State Chemistry", 2 nd Ed., Cambridge University Press, Cambridge.	1997
5.	Whittingham, M.S. and Jacobson, A.J. (Ed.), "Intercalation Chemistry", Academic Press, New York.	1982



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-613

Course Title: Frontiers in Bioinorganic Chemistry

2. Contact Hours:

L: 3

T: 0

MTE: 25

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative weightage: CWS: 25

PRS: 0

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective of Course: To impart knowledge about metalloproteins from cellular and molecular

biological point of view.

S.No.	Contents .	Contact Hours
1.	Homeostatic mechanism: Cellular components and pathways in the context of metal ions, homeostatic mechanism in cell – prokaryotes to eukaryotes to human. Evolutionary pathway metals, metallocofactors and prosthetic groups.	3
2.	Metal ion transport and assembly of metalloproteins: Details of the metal transport in Yeast and in higher organisms, Proteins involved in uptake and efflux, metallochaperones, transcription factors (Acel and Macl, copper sensor). Assembly of metals in protein, photoactivation. Heme synthesis, covalent and non-covalent interactions of heme with protein, assembly of heme in heme proteins-cytochrome c vs cytochrome b5, heme chaperoning and role of CCME. Identification of a protein as heme protein, heme oxygenase, reconstitution of hemeproteins with modified heme/other cofactors and their application in biocatalysis and electron transfer.	6
3.	Molybdenum and tungsten in biology: Hyperthermophilic and thermophilic bacteria. Mo and W containing enzymes, mechanism of catalytic activity-nitrogenase, sulfite oxidase, nitrate reductase, acetylene hydratase, xanthine oxidase, DMSO reductase. Structural and functional modeling of Mo and W sites and their applications as biocatalysis.	6
4.	Iron in biosystem: Non-Heme— iron-sulphur proteins, other non-heme iron proteins—lipoxygenase and its implication in cancer research, nitrile hydratase and its application to industry. Structural and functional modeling of heme and non-heme metal-sites and their applications as biochemistry and biocatalysis with examples such as nitrile hydratase, lipoxygenase, acetyl coenzyme synthetase (ACS), DAP1. Heme— catalytic mechanism of nitric oxide synthese and heme oxygenase.	
5.	Metal ions and diseases: Role in Alzheimer's disease—Aggregation of proteins, role of copper, zinc and iron. Application of radiochemistry for the identification of metal ions. Metal binding in prion protein—binding of copper and manganese. Manganism— occupational exposure, manganese toxicity, effect on calcium channel, proteomics of manganese toxicity. Inorganic NO-donor and their applications	8
6.	Bioinformatics and postgenomic era: Search of metalloprotein and metal binding	<u> </u>

	motif (eg Dap1). De novo design of proteins, artificial heme binding protein, target protein. Modeling with protein structure from protein data bank. DNA intercalation and electron transfer through DNA, RNA metal interactions.	5
7.	Biomineralization: Biomineralization in the context of bone, teeth and mollusk cells, application into materials science and biomimetic engineering. Bioorganometallic chemistry—introduction and applications.	4
8.	NMR structural biology and structure solution of metalloproteins: Selection of a target protein, plasmid preparation and overexpression, reparation of sample for NMR. Overexpression of heme protein—cytochrome c vs cytochrome b. Labeling of protein by ¹⁵ N and ¹³ C, standardization of overexpression and purification (heme as well as nonheme). Details of the NMR Experiments for spectral analysis, paramagnetic NMR, structure solution.	5
	Total	42

S.No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Cotton, F.A. and Wilkinson, G. "Advanced Inorganic Chemistry", 4 th Ed. John Wiley & Sons, New York.	1980
2.	Huheey, J.E., Keiter, E.A. and Keiter, R.L. "Inorganic Chemistry: Principles of Structures and Reactivity", 4 th Ed., Low Print Edition, Pearson Education Ltd, Asia, Reprint in India.	2001
3.	Bertini, I., Gray, H.B. Lippard, S.J. and Valentine, J.S. "Bioinorganic Chemistry", University Science Book, South Asian Edition Reprint.	2004
4.	Pecoraro, V.L. "Manganese Redox Enzymes", VCH: New York.	1992
5.	Bertini, I., Sigel, A. and Sigel, H. "Handbook on Metalloproteins", Marcel Dekker.	2001



NAME OD DEPTT./CENTRE: DEPARTMENT OF CHEMISTRY

1. Subject: CYN-615

Course Title: Crystal and Molecular Structure

2. Contact Hours:

L: 2 T: P: 2/2

3. Examination Duration (Hrs): Theory: 2

Practical: 0

4. Relative weightage: CWS:

20

MTE: 20

PRE: 0

5. Credits: 3

6. Semester: Autumn

PRS: 20

8. Pre-requisite: Nil

7. Subject Area: PEC

9. Objective of Course: To highlight relationship between crystal symmetry and their molecular structure determination process.

10. Details of Course:

S. No.	Contents	Contact Hours
1.	Symmetry and space group: Concept of crystal morphology and crystal symmetry, introduction of screw axis and glide plane, crystal systems and space groups, relation between systematic absence and space group.	6
2.	Single crystal X-ray diffraction: X-ray source, generation of monochromatic wavelength, X-ray diffraction, Bragg equation, concept of reciprocal lattice, Bragg's law in reciprocal lattice, atomic scattering factor, structure factor equation and limiting conditions, intensity and intensity statistics, polarization and Lorentz correction, temperature factor and anisotropic effect, optical classification of crystals, phase problem. Small angle X-ray scattering/diffraction.	10
3.	Single crystal X-ray data handling: Fourier techniques for electron density calculation in X-ray structure, electron density map, heavy atom method, Patterson function, direct method, normalized structure factor, Σ_2 relationship, least square refinement, anomalous dispersion, twinning in crystal.	8
4.	Neutron diffraction: Introduction to neutron diffraction, neutron sources and detection of neutrons, scattering cross sections, application to the studies of molecular structure especially for the detection of neighbouring and light atoms.	4
	Total	28

List of Experiments:

1.	Introduction to the WinGX software to handling single crystal data.
2.	Determination of space group from the hkl data set by systematic absences.
3.	Structure solving with isotropic displacement parameter.
4.	Structure solving with anisotropic displacement parameter and temperature effect.
5.	Treatment of hydrogen for complete structure determination.
6.	Generation of cif, ortep and checking of cif

* This experiment requires two terns.



S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
1.	Stout, G.H. and Jensen, L.H., "X-Ray Structure Determination: A Practical Guide", 2 nd Ed., John Wiley and Sons, New York.	1989
2.	Clegg, W., Blake, A.J., Gould, R.O. and Main, P. "Crystal Structure Analysis: Principles and Practice", Oxford Science Publications.	2006
3.	Ladd, M. and Palmer, R. "Structure Determination by X-ray Crystallography", 4 th Ed., Kluwer Academic/Plenium Publishers.	2003
4.	Massa, W. "Crystal Structure Determination" Springer-Verlag, Berlin, Heidelberg.	2004
5.	Buerjer, M.J. "Crystal Structure Analysis", John Wiley and Sons, New York.	1987



NAME OF DEPTT./CENTRE: DEPARTMENT OF CHEMISTRY

1. Subject: CYN-617

Course Title: Supramolecular Chemistry

2. Contact Hours:

P: 0

3. Examination Duration (Hrs): Theory: 3

Practical:

T: 0

4. Relative weightage:

CWS: 25

MTE: 25 PRS: 0

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Basic chemistry and spectroscopy.

9. Objective of Course: To impart in depth knowledge of non-covalent interactions in supramolecular systems and their applications.

10. Details of Course:

S.	Contents	Contact
No.		Hours
1.	Fundamentals of supramolecular chemistry: Definitions, brief overview and examples; types of non-covalent interactions (H-bonding, electrostatic (ion-ion, ion-dipole, dipole-dipole), hydrophobic and steric, π - π , van der Waals), concepts of host-guest complexation with examples from ionophore chemistry, complexation of ions, molecular baskets, chalices and cages—podands, crown ethers, cryptands, calixarenes, macrocyclic effect, complexation of neutral molecules, self-assembly, molecular boxes and capsules, self-complementary species and self-replication.	8
2	Supramolecular chemistry and biological processes: Cation binding (biological relevance, affinity and selectivity, artificial ionophores, natural and artificial cation channels). Anion and neutral molecule binding—relevance factors affecting affinity and selectivity, anion and neutral molecule binding in biology, artificial hosts for anions, katapinands, guanidinium receptors, receptors based upon Lewis acid-base concepts, enantio-selective anion recognition, cyclodextrins, anion binding based upon ion-dipole interactions, simultaneous anion-cation binding, neutral molecule recognition and binding.	8
3	Synthesis of supramolecules: Synthesis of macrocycles, synthesis of receptors for cations anions, and neutral molecules, non-covalent synthesis, metal directed self-assembly of complex supramolecular architecture—rotaxanes, catenanes.	6
4.	Physical methods in supramolecular chemistry: Spectroscopy in supramolecular chemistry, determination of stoichiometry, stability constants, and geometry of complexes, binding constant determination, dynamics of supramolecular systems (solid state vs solution behavior).	
5.	Application of supramolecular chemistry: Supramolecules in catalysis, as membrane transport, sensors, phase-transfer catalysts, supramolecular devices and switches, memories, logic gates and related systems, molecular scale machines (mechanical rotors, gears and brakes), conversion of light into fuels and light into electricity.	
ļ	Total	42

		17	
	Name of Authors/ Bokks/ Publishers	Year of	
S.No.	lable of Antifold Doving I appoint a		

		Publication/ Reprint
1.	Steed, J.W. and Aswood, J.L., "Supramolecular Chemistry", Wiley.	2000
2.	Dodziuk,,H., "Introduction to Supramolecular Chemistry", Springer, ISBN 1402002149.	2001
3.	Beer, P.D., Gale, P.A. and Smith, D.K., "Supramolecular Chemistry", Oxford Chemistry Printers, ISBN-10: 0-19-850447-0.	1999
4.	Cragg, P., "A Practical Guide to Supramolecular Chemistry", Wiley-VCH, ISBN: 0-470-86654-3.	2005
5.	Schneider, H.J. and Yatsimirsky, A., "Principles and Methods in Supramolecular Chemistry", Wiley-VCH, ISBN: 0-471-97253-3.	2000



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-619

Course Title: Modern Organic Synthetic Methods

2. Contact Hours:

P: 0 L: 3 T: 0

3. Examination Duration (Hrs):

Theory: 3

Practical:

4. Relative weightage: CWS: 25

PRS: 0 MTE: 25

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area:

PEC

50

8. Pre-requisite: Basic organic chemistry.

9. Objective of Course: To impart the knowledge of modern synthetic methods used in functional group transformations.

S. No.	Contents	Contact Hours
1.	Oxidations: Oxidations of hydrocarbons (alkanes, alkenes and aromatic), alkenes to epoxides (peroxides/per acids based), Sharpless asymmetric epoxidation, Jacobsen epoxidation, Shi epoxidation, alkenes to diols (Manganese, Osmiumbased), Sharpless asymmetric dihydroxylation, Prevost reaction and Woodward modification, alkenes to carbonyls with bond cleavage (manganese, osmium, ruthenium and lead based-ozonolysis), alkenes to alcohols/carbonyls without bond cleavage (hydroboration-oxidation, Wacker oxidation, selenium, chromium based allylic oxidation), ketones to α-hydroxy ketones, α,β-unsaturated ketones, ester/lactones (Baeyer-Villiger), alcohols to carbonyls (chromium, manganese, aluminum, silver, ruthenium, DMSO, hypervalent iodine and TEMPO based reagents), alcohols to acids or esters, phenols (Fremy's salt, silver carbonate).	12
3.	Named reactions: Baylis-Hillman reaction, Henry reaction, Nef reaction, Kulinkovich reaction, Ritter reaction, Sakurai reaction, Tishchenko reaction, Ugi reaction, Brook rearrangement and Tebbe olefination.	6
4.	Protection and deprotection of functional groups: Protection and deprotection of hydroxy, carboxyl, carboxyl, carboxy amino groups and carbon-carbon multiple bonds, chemo- and regioselective protection and deprotection, illustration of protection and deprotection in multi-step synthesis.	6
5.	Retrosynthetic analysis: Basic principles and terminology of retrosynthesis, guidelines, synthesis of aromatic compounds, one group and two group C-X disconnections, one group C-C and two group C-C disconnections, amine and alkene synthesis, important strategies of retrosynthesis, functional group transposition, important functional group interconversions, reversal of polarity (umpolung).	8
	Total	42



S. No.	Name of Authors/Books/Publishers	Year of Publication / Reprint
1.	Carey, F. A. and Sundberg, R. J., "Advanced Organic Chemistry, Part B: Reactions and Synthesis", 5 th Ed., Springer.	2007
2.	Carruthers, W. and Coldham, I., "Modern Methods of Organic Synthesis", 4 th Ed., Oxford University Press.	2004
3.	Smith, M.B., "Organic Synthesis", 3rd Ed., Academic Press.	2010



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-621

Course Title: Organic Structure Determination

2. Contact Hours:

L: 3

T: 0 P: 0

3. Examination Duration (Hrs):

Theory: 3

4. Relative weightage:

Practical:

CWS: 25

PRS: 0 MTE: 25 ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Rasic organic chemistry.

9. Objective of Course: To familiarize the students with the organic structure determination methods involving spectroscopy.

S. No.	Contents	Contact Hours
1.	Electronic spectroscopy: Electronic transitions in organic molecules, Woodward-Fieser rules for alkenes, Woodward rules for enones, aromatic compounds.	4
2.	Infrared and Raman spectroscopy: For simple organic molecules, predicting number of active modes of vibrations, analysis of representative spectra of compounds with various functional groups, application of isotopic substitution.	4
3.	Mass spectrometry: Basic principles, hard and soft ionization techniques, mass analyzer in ESI-MS and MALDI-MS, high resolution MS, isotope abundance, molecular ion, fragmentation processes (McL) of organic molecules, deduction of structure through mass spectral fragmentation.	6
5.	Nuclear magnetic resonance: Effect of magnetic field strength on sensitivity and resolution, chemical shift δ, inductive and anisotropic effects on δ, chemical structure correlations of δ, chemical and magnetic equivalence of spins, spin-spin coupling, structural correlation to coupling constant <i>J</i> , first order and second order spectra, examples of AB, AX, ABX, AMX and AA'BB' systems, simplification of second order spectrum, selective decoupling, double resonance, use of chemical shift reagents for stereochemical assignments, ¹³ C NMR, T ₁ relaxation, NOE effects, DEPT, determination of number of attached hydrogens, ¹ H and ¹³ C chemical shifts to structure correlations, study of dynamic processes by VT NMR, restricted rotation (DMF, DMA, biphenyls, annulenes), cyclohexane ring inversion, degenerate rearrangements (fullvalene and related systems). Multinuclear NMR, COSY, DQF-COSY, HETCOR, HMQC, HMBC, TOCSY, ROESY, VGSE. Spectroscopic application: Structure elucidation of organic compounds using	20
	spectroscopic methods. Total	42
1	* Star	TD.



S. No.	Name of Authors/Books/Publishers	Year of Publication / Reprint
1.	Silverstein, R. M., Webseter, F. X. and Kiemle, D., "Spectrometric Identification of Organic Compounds", 7 th Ed., John Wiley & Sons.	2005
2.	Kemp, W. L., "Organic Spectroscopy", Palgrave.	2008
3.	Pavia, D. L., "Spectroscopy", 4th Ed., Cengage.	2012
4.	Williams, D. and Fleming, I., "Spectroscopic Methods in Organic Chemistry", 6 th Ed., McGraw Hill Education (India) Private Limited.	2011



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-623

Course Title: Organic Semiconductors

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative weightage: CWS: 25

PRS: 0

MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Basic organic chemistry and spectroscopy.

9. Objective of Course: The course will deal with rapidly emerging areas in organic electronic materials.

10. Details of Course:

S.No.	Contents	Contact Hours
. 1.	Introduction: General description of conjugated organic oligomers, dendrimers and polymers. Conjugated polymer structural types (polyacetylenes, polyphenylene-vinylenes, polyphenylenes, polyfluorenes, polythiophenes, polyphenylenes, polyphenylenes, polymers, phosphorescent polymers). Carbon-rich compounds, Cross-conjugation.	6
2.	Synthesis: Useful synthetic methods for the construction of conjugated organic oligomers and polymers. C-C and C-heteroatom coupling reactions — historical context and latest developments. Representative examples. Mechanistic consideration. All-benzenoid polycyclic aromatic hydrocarbons— synthesis, self-assembly and applications in organic electronics. Solid state strategy for the preparation of carbon-rich polymers.	10
3	Properties: Electronic structure of organic semiconductors — relationship between two view points: solid state physics and molecular picture of conjugated organics. Electrochemistry, electrochromism and energy level measurements. Charge transport (electronic conduction in photoactive molecular-wires). Luminescence. Energy transfer and electron transfer. Excitation dynamics in organic semiconductors. Fluorescence sensing. Non-linear optical properties.	14
4.	Applications: Field-effect transistors, light-emitting diodes, photovoltaics and solar cells-device architectures, materials, characterization and theory of operation. Biosensors – electrochemical detection, fluorescence optical amplification (protein, DNA and RNA sensing), solid state applications (DNA chips and micro arrays).	12
	Total	42

S.No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Haley, M.M. and Tykwinski, R.R. (Ed.), "Carbon-Rich Compounds: From Molecules to Materials" Wiley.	2006
2.	Singh, J., "Smart Electronic Materials: Fundamentals and Applications", Cambridge University Press.	2005
3.	Fraxedas, J., "Molecular Organic Materials: From Molecules to Crystalline	2006

	Solids", Cambridge University Press.	
4.	Schubert, E.F., "Light-Emitting Diodes", 2 nd Ed., Cambridge University Press.	2006
5.	Brabec, C., Dyakonov, V., Parisi, J. and Sariciftei, N.S. (Ed.), "Organic Photovoltaics", Springer.	2003
6.	Agranovich, V.M. (Ed.), "Organic Nanostructures", IOS press.	2002
7.	Norio, M. (Ed.), "Cross-Coupling Reactions: A Practical Guide", Springer.	2002



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN -625

Course Title: Proteins and Polypeptides

2. Contact Hours:

L: 3

T: 0 P: 0

MTE: 25

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative weightage: CWS: 25

PRS: 0

1 Machelli

ETE: 50 PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

nea.

8. Pre-requisite: Basic knowledge of analytical techniques.

9. Objective of Course: To provide advanced knowledge in protein chemistry.

10. Details of Course:

S.No.	Contents	Contact Hours
1.	Proteins and peptides: General nature, characteristics, introduction to primary, secondary and tertiary structures.	4
2.	Separation and purification methods: Electrophoresis, isoelectric focussion, gel filtration, affinity chromatography, and ion exchange – choice of gel support materials (agarose, cellulose, polyacrylamide, glass beads, DEAE-cellulose, CM-cellulose etc.) HPLC.	8
3.	Fragmentation of polypeptides: Chemical methods – cleavage of di-sulfide bonds, oxidation, partial acid hydrolysis, cleavages at methionine, tryptophan, tyrosine, cysteine. Enzymic methods – protein modification reactions disulphide bond cleavage, alkylaton of sulphydryl groups, modification of lysine and arginine residues. Specificity and conditions for trypsin, thrombin, chymotrypsin, thermolysin, pepsin papaine.	12
4.	Determination of peptide sequences: Manual sequencing, solid phase sequence analysis, automated liquid phase sequence analysis, microsequence analysis using a gas-liquid solid-phase sequenator, C-terminal sequence analysis.	6
5.	Applications of electron impact mass spectrometry, X-ray crystallography and electron microscopy: In the structural analysis of peptides and proteins.	6
6.	Peptide mapping and peptide synthesis: Prediction of peptide and protein structure. Peptide synthesis including solid-phase and automated synthesizers.	6
	Total	42

S.No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Lehninger A.L., "Biochemistry", North Publishers.	1980
2.	Stryer L., "Biochemistry", CBS Publications and distributors.	1981



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

Course Title: Advanced Surface and Colloidal Chemistry 1. Subject: CYN-627

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative weightage: CWS: 25

PRS: 0

MTE: 25

PRE: 0 ETE: 50

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Knowledge of physical chemistry.

9. Objective of Course: To impart advanced knowledge of surface and interfacial phenomenon.

S.	Contents	Contact Hours
No. 1.	Surfactants and Interfacial Phenomena: Classification, micellization, c.m.c. and	9
	its determination. Shape and structure of micelles, effect of additives on	
	micellization, thermodynamics of micellization, solubilization and applications, effect of electrolytes on solubilization. Macro and micro emulsions, dispersion and aggregation of solids by surfactants.	
2.	Membranes and their Applications: Artificial and natural membranes, Donnan membrane equilibrium, transport of electrolytes, membrane potential and ion selective electrodes.	6
3.	Adsorption on solids and porous materials: Model for multilayer adsorption, BET isotherm and application to different types of adsorbents, adsorption by porous, non-porous and microporous solids. Estimation of specific surface area and pore size distribution. Special problems encountered with very narrow pore size material and adsorption from liquid phase.	7
4.	Colloid systems and their properties: Origin of the charges, electro-kinetic phenomena, electrophoresis, electroosmosis, sedimentation and streaming potential. The concept of electrical double layer and various models to explain its structure and properties, DLVO theory and stability of colloids. Smoluchowski theory of kinetics of coagulation and distribution of colloids aggregates. Organic and inorganic gels and clay colloids.	8
5.	Macromolecules: Concepts of mass and number average molecular weights, methods of determining molecular weights (osmometry, viscometry, diffusion and light scattering method), sedimentation, fractional properties of macromolecules, statistical distribution of end to end dimension, calculation of average dimension of various chain structures.	
	Total	42

S. N	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Hunter, R. J., "Foundation of Colloid Science", Oxford Univ. Press.	2009
2.	Lyklema, J., "Fundamentals of Interface and Colloid Science", Academic Press San Diego.	2000



3.	Adamson, A.W., "Physical Chemistry of Surfaces", 5th Ed., John Wiley and	1990
1	Sons, New York.	1991
4.	Kruyt, H.R., "Colloid Chemistry" Vol. I & II. Elsevier Press.	1771
5.	Greg, S.J. and Singh, K.S.W., "Adsorption, Surface Area and Porosity", 2 nd Ed.,	1982
	Academic Press. U K.	10.52
6.	Flory P. J., "Principles of Polymer Chemistry", 1st Ed., Cornell University Press.	1953
	Rubinstein M. and Colby R. C., "Polymer Physics", 1st Ed., Oxford University	2003
7.	Press	
8.	Billmeyer, F. W., "Textbook of Polymer Science", 3 rd Ed., Wiley India Private	2007
	Limited	



NAME OF DEPTT /CENTRE.

DEPARTMENT OF CHEMISTRY

1. Subject:

CYN-629

Course Title: Advanced Physical Chemistry

2. Contact Hours:

L: 3 T: 0

P: 0

3. Examination Duration (Hrs) Theory

03

Practical

4. Relative weightage: CWS: 25

25 PRS: 0

MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester:

Autumn

7. Subject Area: PEC

8. Pre-requisite:

Knowledge of physical chemistry

9. Objective of Course: To familiarize the students with advanced kinetics, thermodynamics and quantum chemistry.

10. Details of Course:

S.No.	Contents	Contact Hours
1.	Advanced chemical kinetics: Theories of unimolecular reactions, kinetics-proton transfer and electron transfer reactions, fast reactions—rapid flow, stopped-flow and relaxation techniques, molecular beam method, diffusion controlled reactions, oscillatory reactions, linear free energy relationship, elucidation of mechanism from kinetic data.	14
2.	Statistical mechanics and irreversible thermodynamics: Phase space, Liouville's theorem, Maxwell-Boltzmann, Bose-Einstein, Fermi-Dirac statistics. Affinities and fluxes, reversible and irreversible processes, entropy production for some important irreversible processes, entropy flow due to exchange of mater and energy, entropy changes due to chemical reaction, affinity and coupling of chemical reaction, the phenomenological laws and equations and their applications in chemistry, fluctuations, response functions, time correlation function, distribution function.	14
3.	Advanced quantum chemistry: Dirac Bra-ket notation, addition of angular momentum, use of ladder operators—rigid rotor and harmonic oscillator, variation method—treatment of He atom, perturbation method—examples of anharmonic oscillator, He atom, Stark and Zeeman splitting, Hartree-Fock method, introduction to post Hartree-Fock methods.	14
	Total	42

S. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Laidler, K.J., "Reaction Kinetics", Anand Sons, New Delhi.	2005
2.	Kondepudi, D. and Prigogine, I., "Modern Thermodynamics: From Heat Engines to Dissipative Structures", John Wiley & Sons.	1998
3.	Callen, H. B., "Thermodynamics and an Introduction to Thermostatistics", John Wiley and Sons.	1985
4.	Bransden, B.H. and Joachain, C.J., "Quantum Mechanics", Addison-Wesley.	2000
	Sakurai, J. J., "Modern Quantum Mechanics", Pearson Education.	1994



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN- 631

Course Title: Materials Chemistry

2. Contact Hours:

L: 2

T: 0

P: 2/2

3. Examination Duration (Hrs):

Theory: 2

PRS: 20

Practical: 0

4. Relative Weightage:

CWS: 20

MTE: 20

ETE: 40

PRE: 0

5, Credits: 3

7. Subject Area: PEC

6. Semester: Autumn

8. Pre-requisite: Structure, Bonding and Properties of Solids.

9. Objective: To introduce the students in the area of materials chemistry.

10. Details of Course:

S. No.	Contents	Contact Hours
1.	Reactions in the solid state and phase transitions: General principles, kinetics of solid-state reactions-nucleation and diffusion rate, Wagner mechanism, Kirkendall effect, Vegard's law, phase transition-Ehrenfest classification and Lanbda transition, structural and magnetic phase transitions order-disorder, martensitic and atensitic, ferro-para.	4
2.	Structure determination: X-ray powder diffraction - lattice refinement, pattern simulation, refinement of crystal structure—Rietveld method, profile fitting, refinement of occupancy and thermal factors, isotropic and anisotropic displacements, reliability factors. Neutron scattering—elastic and inelastic, diffraction experiments with neutron, chemical and magnetic structure, magnetic scattering, elucidation of magnetically ordered structures.	8
3.	Advanced materials and devices: Magnetic materials, dilute magnetic semiconductors, magnetic shape-memory alloys, magnetoresistive, semiconducting, superconducting, multiferroic and spintronic materials, graphene and its nanocomposites, Li-ion batteries, solid oxide fuel cells, perovskite solar cells, magnetic tunnel junctions and spin-valves.	` 10
4.	Preparation methods and processing of materials: Revisit to general synthetic routes- gas-solid, liquid-solid and solid-solid methods. Sputtering, pulsed laser deposition (PLD), molecular beam epitaxy (MBE), liquid quenching, arc melting, induction, zone melting, chimie-douce, high-pressure, mechanochemical, ball-milling methods.	6
	Total	28

List of experiments:

MISCOL EVDEN	ist of experiments.	
1.	Drawing and visualization of crystal structures and structural parameter analysis.	
2.	Indexing of powder X-ray diffraction (PXD) patterns.	
3.	Lattice parameter refinement and doping analysis.	
4.	Simulation of PXD patterns and occupancy modelling in doped systems.	
i	Whole pattern fitting-Le Bail method.	
5.	Rietveld refinement of PXD data for crystal structure determination.	
6.	Rietveid refinement of FAD data for crystal structure determination.	



7.	Density functional theory softwares.
8.	Electronic structure calculation of metals/insulators.
9.	Electron localization function and Fermi surface calculation.
]	

S. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	West, A. R., "Solid State Chemistry and its Applications", Reprint, Wiley India.	2013
2.	Rao, C.N.R. and Gopalakrishnan, J., "New Directions in Solid State Chemistry", 2 nd edn., Cambridge University Press.	1997
3.	Stout, G.H. and Jensen, L.H., "X-Ray Structure Determination: A Practical Guide", 2 nd edn., Wiley-Interscience.	1989
4.	Giacovazzo, C., Artioli, G., Monaco, H. L., "Fundamentals of Crystallography", Oxford University Press.	2006
5.	"Magnetic Materials: Fundamentals and Device Applications", Nicola Spaldin, Cambridge University Press.	2003
6.	"Electronic Structure of Materials"; A. P. Sutton, A. D. Sutton, Oxford University Press.	1993

Sl. No.	Softwares and Databases		Source
1.	Diamond	(commercial)	www.crystalimpact.com
2.	XCrysden	(freeware)	http://www.xcrysden.org/
3.	ICSD	(commercial)	www.fiz-karlsruhe.com
4.	Powder Cell	(freeware)	www.ccp14.ac.uk
5.	FullProf	(freeware)	www.ill.eu
6.	Full Potential Localized Orbital (FPLO) Code	(commercial)	www.fplo.de



NAME OF DEPTT/CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-633 Course Title: Nanoscale Materials: Properties and Applications

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs):

Theory: 3

Practical: 0

4. Relative Weightage: CWS:

PRS: 0

MTE: 25

ETE: 50

PRE: 0

5. Credits: 3

C C---

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart fundamental understanding on nanoscale materials, their properties and applications.

S. No.	Contents	Contac t Hours
1.	Introduction to nanoscale materials: The nano-length scale, quantum confinement effect, consequences of quantum confinement of electrons, conceptual development of band theory — from molecules to clusters/quantum dots to macroscopic crystals, material dependence of nanoscale and quantum	6
	size-effect, consequences of carrier confinement in semiconductors and metals.	:
2.	Structure of nanomaterials: Crystalline and amorphous nanomaterials, nanocrystals, surface energy and crystal facets, equilibrium shape of nanocrystals, Wulff construction, Roughening temperature, surface energy as a function of surface curvature, chemical potential and solubility as a function of	6
	surface curvature and particle size, Ostwald ripening, nucleation and growth of nanoparticles.	
3.	Physical and chemical characteristics of nanomaterials: Surface area measurement, determination of size and textural studies, composition and elemental analysis, high chemical reactivity of nanoscale materials, effect of size and shape on nanocrystal reactivity, agglomeration and sintering of nanomaterials, dispersibility and chemical stability of nanoparticles in solution, surface modification of metallic and semiconductor nanoparticles, nanofabrication and nanomanipulation	12
4.	Synthesis and applications of nanomaterials: Concepts of top-down and bottom-up approaches, chemical, aerogel, aerosol, spray-pyrolysis, microemulsion, solvothermal, sonochemical, and microwave methods of synthesis. Reactivity studies by adsorption—SO ₂ , CO ₂ , H ₂ S, CCl ₄ and chemical warfare agents, destructive adsorption, detoxification by adsorption, air purification, desulphurization, biocidal applications, modification of nanocrystalline metal oxides and their applications.	14
5.	Toxicology of nanomaterials: Health concerns of using nanomaterials, inhalation toxicity, oral toxicity, environmental toxicity, cyto- and bio-toxicity of	4



nanomaterials, environmental protection, precautions and case studies.		
	Total	42

Sl. No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Klabunde, K.J. (Ed.), "Nanoscale Materials in Chemistry", Wiley-Interscience, New York.	2001
2.	Schmid, G. (Ed.), Nanoparticles: From Theory to Application", Wiley-VCH, Weinheim.	2004
3.	Cao, G. and Wang, Y., "Nanostructures and Nanomaterials: Synthesis, Properties and Applications", 2 nd Revised Ed., World Scientific.	2011
4.	Rodriguez, J. A. And Fernandez-Garcia, M. (Ed.), "Synthesis, Properties and Applications of Oxide Nanomaterials", John Wiley, New York.	2006
5.	Rao, C.N.R., Müller, A. and Cheetham, A.K., "The Chemistry of Nanomaterials: Synthesis, Properties and Applications", Vol. 1 and 2, Wiley-VCH Verlag, Weinheim.	2004
6.	West, A. R. "Solid State Chemistry and its Applications", Reprint, Wiley India.	2013



NAME OF DEPTT/CENTRE: DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-635

Course Title:

Advanced Magnetic Resonance Spectroscopy

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS: 0

MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PEC

8. Pre-requisite: Basic knowledge of magnetic resonance spectroscopy.

9. Objective:

To provide advance knowledge of NMR and ESR techniques for structure elucidation of molecules and radicals.

S. No.	Contents	Contact Hours
1.	Magnetic resonance: Nuclear spin, Zeeman effect, Larmour precession, Boltzmann distribution of spin population, magnetic field and resolution, folded signals, peak broadening and solid state effect.	3
2.	Relaxation processes: Longitudinal and cross sectional relaxation mechanism of magnetic vector, Bloch equation, pulse strength and magnetization vector, spin-saturation, spin-inversion and spin-echo methods for relaxation times. Effect of chemical shift anisotropy, scalar coupling, dipole-dipole coupling and electron interactions on relaxation time.	4.
3.	Chemical shift: Shielding coefficient and chemical shift, low field and high field chemical shift, solvent effect and factors influencing the chemical shift. anisotropic effect in benzene and carbonyl compounds, chemical and magnetic equivalence.	3
4.	Scalar and dipolar couplings: Coupling of nuclear spins, coupled spins energy diagrams and sign of coupling constant in AX and AB systems. Geminal and vicinal coupling and Karplus relation for tetrahedral angle. Dipolar couplings in solids and bond length.	3
5.	Double resonance: Double resonance and Simplification spectra, pulse sequence, spin polarization transfer and Nuclear Overhauser effect, progressive and retrogressive internuclear relationship, selective decoupling, J-modulated spin-echo and attached proton test (APT), insensitive nuclei enhancement by polarization transfer (INEPT) and distortion less enhancement by polarization transfer (DEPT). Fourier transform and ¹³ C NMR spectroscopy.	6
6.	Dynamic NMR: Chemical and dynamic processes by NMR, temperature dependant rotamers and colescence temperature. Dynamic equilibrium, line shape and Gibb's free energy.	4
7.	2-Dimensional NMR spectroscopy: Basics of two dimensional NMR spectroscopy, 2D correlation (COSEY) and 2D Nuclear Overhauser enhancements Spectroscopy (NOESY) experiments and their applications in structural elucidation of organic molecules.	8 .
8.	ESR spectroscopy: Electron spin and magnetic moment, g-factor, triplet state and spin transitions, hyperfine structures, Zerofield splitting, Kramer degeneracy and	5

	double resonance.	
9.	NQR spectroscopy: Nuclear quadrupole moment and electrical field gradient, quardrupole coupling constants (NQCC) and asymmetry parameter (□), pure NQR and Zeeman spectra of spin 1 and spin 3/2 systems, the Towners-Dailey theory and interpretation of NQCC in terms of characteristic of the bonds and effects of other parameters.	6
	Total	42

S., No.	Name of Authors/ Books/ Publishers	Year of Publication/ Reprint
1.	Gunther, H., "NMR Spectroscopy. Basic Principles, Concepts and Applications in Chemistry", Wiley India.	2010
2.	Banwell, C.N. and McCash, M., "Fundamentals of Molecular Spectroscopy", McGraw Hill Education Pvt. Ltd, India.	2013
3.	Graybeal, J.D., "Molecular Spectroscopy", McGraw-Hill Book Company, India.	1988
4.	Drago, R.S., "Physical Methods in Inorganic Chemistry", Affiliated East West Press, Pvt. Ltd. India.	2010
5.	Weil, J.A., Bolton, J.R. and Wertz, J.E., "Electron Paramagnetic Resonace", Wiley-Interscience, New York.	1994



NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-606

Course Title: Total Synthesis

2. Contact Hours:

L: 3

T: 0 P: 0

3. Examination Duration (Hrs):

Theory: 03

Practical: 0

4. Relative weightage:

CWS: 25

MTE: 25 PRS: 0

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

8. Pre-requisite: CYN-506

9. Objective of Course: To familiarize the students with the tactics and strategies in the total synthesis of targeted compounds.

10. Details of Course:

S.	Contents	Contact Hours
No.	c and a significant of complex molecular architectures	2
1.	Introduction to strategies for synthesis of complex molecular architectures.	77
2.	Synthesis of antibiotics – penicillin V and tetracycline.	- '-
3.	Synthesis of alkaloids – reserpine and camptothecin.	7
	Synthesis of terpenoids – β-pinene, camphor, abietic acid and β-amirine.	8
4.	Synthesis of terpenoids – p-pinetic, campion, action and cortisone	8
5.	Synthesis of steroids and hormones – cholesterol, progesterone and cortisone.	10
6.	Synthesis of prostaglandins PGE_2 and $PGF_{2\alpha}$; glycosidic pigments anthocyanins and quercetin; macrocyclic lactam fluvirucin- B_1 -aglycone; and vitamin biotin.	10
	Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication / Reprint
1.	Finar, I.L., "Organic Chemistry", Vol. 2, 5th Ed., ELBS.	1975
2.	Corey, E.J. and Cheng, XM., "The Logic of Chemical Synthesis", Wiley-VCH,	1995
3.	Weinheim. Nicolaou, K.C. and Sorensen, E.J., "Classics in Total Synthesis", Wiley-VCH,	1996
4.	Weinheim. Gewert, J.A., Gorlitzer, J., Götze, S., Looft, J, Menningen, P., Nobel, T., Schirock, H. and Wulff, C., "Organic synthesis workbook", Wiley-VCH; Weinheim.	2000



NAME OD DEPTT./CENTRE: DEPARTMENT OF CHEMISTRY

1. Subject: CYN-608 Course Title: Chemical Biology

2. Contact Hours: L: 3 T: 0 P: 0

3. Examination Duration (Hrs): Theory: 3 Practical: 0

4. Relative weightage: CWS: 25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

8. Pre-requisite: Knowledge of inorganic, organic synthesis and basic of fluorescence.

6. Semester: Spring

9. Objective of Course: To develop the knowledge of applying chemical tools in recognition process of biological systems.

7. Subject Area: PEC

10. Details of Course:

Credits: 3

5.

S.	Contents	Contact
No.		Hours
1.	Application of spectroscopy in chemical biology: Brief introduction to chemical biology, biological structures and its constituents, spectroscopic tools to rationalize chemical reactions in biosystems, design of fluorescent and fluorogenic probes and optical biosensor molecules.	9
2.	Chemical reactions suitable for biological applications: Solid phase peptide synthesis, bioorthogonal reactions, functional group specific ligation techniques, strategies for attachment of synthetic molecules to biomolecules, Staudinger ligation, native chemical ligation, click chemistry, site selective protein modification.	12
3.	Chemical interpretation of protein labeling technologies: Green fluorescence protein (GFP), fluorescein arsenical hairpin (FlAsH), SNAP tag, CLIP tag, mutant β-lactamase (BL) tag, halo tag, and their selective significant applications.	12
4.	Chemistry behind artificial cellular components: Synthetic membranes, vesicles – ion transport, unnatural amino acids and their incorporation, DNA chemistry and its uses, nucleic acid templated chemistry, chemistry of morpholino and locked nucleic acid (LNA), siRNA.	9
	Total	42

S. No.	Authors/Title/Publishers	Year of Publication/ Reprint
1.	Waldmann, H. and Janning, P. "Chemical Biology: Learning Through Case Studies", Wiley-VCH, Weinheim.	2009
2.	Dobson, C.M., Gerrard, J.A. and Pratt, A.J. "Foundations of Chemical Biology", Oxford University Press.	2002
3.	Miller, A. and Tanner, J. "Essentials Of Chemical Biology: Structure and Dynamics of Biological Macromolecules", Wiley.	2002
4.	Waldmann, H. and Janning, P. "Chemical Biology: A Practical Course", Wiley-VCH. Weinheim.	2004
5.	Hermanson, G.T. "Bioconjugate Techniques", 3 rd Ed., Academic Press.	2013

6.	Lackowicz, J.R. "Principles of Fluorescence Spectroscopy", 3 rd Ed., Springer.	2006
7.	Blackburn, G.M., Gait, M.J., Loakes, D. and Williams, D.M. "Nucleic Acids in Chemistry and Biology", 3 rd Ed., RSC Publishing.	2006
8.	Schreiber, S.L., Kapoor, T. and Wess, G. "Chemical Biology: From Small Molecules to Systems Biology and Drug Design", Vol. 1-3, Wiley-VCH, Verlag GmbH & Co. KGaA.	2007



NAME OD DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject Code: CYN-610

Course Title: Molecular Modeling and Simulations

2. Contact Hours:

L: 2

P: 2/2 T: 0

3. Examination Duration (Hrs.):

Theory: 2

Practical: 0

4. Relative Weightage:

PRS: 20

MTE: 20

CWS: 20

ETE: 40

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

→8. Pre-requisites: CYN-501 and CYN-503.

9. Objective: To provide knowledge on computational methods in chemistry

10. Details of Course:

S. No.	Contents	Contact Hours
1.	Fundamentals of electronic structure: Basis function—hydrogen-like, Slater-type and Gaussian type orbitals, classification of basis sets — minimal, double zeta, triple zeta, split-valence, polarization and diffuse basis sets, correlation consistent basis sets, basis set super position error, energy minimization methods—derivative and non-derivative methods-simplex method, steepest	8
2.	descents method, Newton-Raphson method, minima, maxima and saddle points. Semi empirical and Ab intio methods: Approximation methods, self consistent field treatment of polyatomic molecules, closed shell systems—restricted Hartree Fock calculations, open shell systems—ROHF and UHF calculations, The Roothan—Hall equations, Koopman's theorem, HF limit and electron correlation, introduction to post Hartree-Fock and density functional methods.	8
3.	Electronic properties: Dipole moment, electrostatic potential, frequencies, population analysis, Mulliken and Lowdin analysis, solvent effects, polarizable and nonpolarizable models.	6
4.	Introduction to simulation methods: Molecular mechanics, Monte carlo and molecular dynamics simulations, periodic boundary conditions, radial distribution function, calculation of thermodynamic properties.	6
	Total	28

List of practicals:

- 1. Potential energy curves for covalently bonded molecules and van der Waals complexes.
- 2. Geometry optimization of molecules at different levels of theory.
- 3. Frequency calculation of molecules.
- 4. Excited states of small molecules.
- 5. Determination of energy barrier for simple reactions.
- 6. Constructing the distribution function for liquid water using molecular dynamics method.
- 7. Calculation of specific heat capacities of liquid water at different conditions.

	Name of Authors/ Books/ Publishers	Year of
5.	Marke of Authors/ books/ 1 ublishers	
No.		Publication/

		Reprint
1.	Jensen, F., "Introduction to Computational Chemistry", John Wiley & Sons Ltd.	1999
2.	Leach, A., "Molecular Modeling: Principles and Applications", Prentice Hall.	2001
3.	Cramer, C. J., "Essentials of computational chemistry: Theories and models",	2002
	John Wiley & Sons.	·
4.	Levine, I. N. "Quantum Chemistry", 7th Ed., PHI Learning Pvt. Ltd., Delhi.	2013



NAME OD DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. 1. Subject Code: CYN-612

Course Title: Carbon Nanomaterials and their Applications

2. Contact Hours:

P: 0 T: 0

3. Examination Duration (Hrs.):

Theory: 3

Practical: 0

4. Relative Weightage:

PRS: 0 CWS: 25

MTE: 25

ETE: 50

5. Credits: 3

6. Semester: Spring

7. Subject Area: PEC

8. Pre requisite: nil

9. Objective of course: To familiarize the student with applications of carbon nanomaterials.

10. Details of Course:

S. No.	Contents	Contact Hours
1.	Introduction and synthetic approaches: Uncatalysed and catalyzed synthesis of carbon nanomaterials, supported nanomaterials, nanodiamonds, fullerenes, single and multi wall carbon nanotubes, carbon dots, carbon fibre.	6
2.	Biological applications: Biomedical application of graphene, biosensors based on carbon materials, tissue engineering, fullerenes in photodynamic therapy, interaction of fullerenes with DNA, an overview of application in cancer therapy, environmental impact of carbon materials, toxicological effect on fish, invertibrates, bacteria, soil microbes.	10
3.	Carbon nanomaterials for sensing applications: Carbon nanotubes and graphene for sensing applications, graphene transistor, single molecule memory devices, organic transistor odour sensor and spintronics.	8
4.	Carbon nanomaterials for electronics, optoelectronics and photovoltaics: Carbon nanotubes—electronic and optoelectronic applications, transistors for digital electronics, graphene—electronics and optoelectronics, digital electronics, photovoltaics—fullerenes, carbon nanotubes, graphene.	14
5.	Space applications: Polymer and carbon nanotube composites for space applications, meteoroids, micrometeoroids, and space-debris, conductive coatings for electrostatic discharge, thermal conductivity, space elevator, solar sails.	4
	Total	42

S. No.	Name of Authors/Books/Publishers	Year of Publication/ Reprint
	Philip, W. HS and Akinwande, D., "Carbon Nanotube and Graphene Device	2011
	Physics", Cambridge University Press.	
	Reich, S., Thomsen, C. and Maultzsch, J., "Carbon nanotubes: Basic concepts and	2004
ļ	physical properties", Wiley-VCH.	
	Saito, R., Dresselhaus, G. and Dresselhaus, M. S., "Physical Properties of Carbon	1998
	Nanotubes", Imperial College Press, London.	
4.	Challa, K.(Ed.), "Carbon Nanomaterials", Vol. 9., Wiley-VCH.	2011

NAME OF DEPTT./CENTRE:

DEPARTMENT OF CHEMISTRY

1. Subject: CYN-614

Course Title: Enantiomeric Separation

2. Contact Hours:

L: 2

T: 0

P: 2/2

3. Examination Duration (Hrs):

Theory: 2

Practical:

4. Relative weightage: CWS: 20

PRS: 20

MTE: 20

ETE: 40

PRE: 0

5. Credits: 3

6. Semester: Spring

7. Subject Are: PEC

8. Pre-requisite: Knowledge of stereochemistry of organic compounds.

9. Objective of Course: To provide knowledge of modern chromatographic separation methods.

10. Details of Course:

S.	Contents	Contact Hours
No.	Y . I . M. I . Manual and a concepts abjustity and molecular	2
1.	Introduction: Modern stereochemical concepts— chirality and molecular structure, definitions and nomenclature.	<u></u>
2.	Techniques used for studies of optically active compounds: Methods not involving separation—polarimetry, NMR, isotope dilution, calorimetry, enzyme techniques. Determination of absolute configuration—X-ray, ORD, CD and chromatography based on comparison.	
3.	Modern chromatographic separation methods: Basic chromatographic theory, instrumentation – gas and liquid chromatography.	4
4.	Direct optical resolution: Theory, general aspects of chiral recognition models—coordination to transition metals, charge transfer interaction, inclusion phenomena. Thermodynamic and kinetic considerations.	4
5.	Chiral gas chromatography: Phases based on chiral metal complexes, inclusion effects-relative merits.	4 .
6.	Chiral liquid chromatography: CSPs based on naturally occurring and synthetic polymers, bonded synthetic chiral selectors, CMPAs.	6
7.	Analytical applications: Amino acids, natural products, pharmaceuticals, microbial and enzymatic reactions.	4
	Total	28

List of Experiments:

- 1. Isolation of L-asparagine from Asparagus or Soybean, its characterization, determination of specific rotation, change of specific rotation from levo to dextro in acidic solution.
- 2. Isomerization of 2-aryl propionic acid or other simple less expensive optically pure compound(s).
- 3. Resolution of (\pm) -mandelic acid.
- 4. Separation of (-)-ephedrine and (-)-mandelate.
- 5. TLC enantioseparation of DL-amino acids on plates impregnated with an optically pure base.
- 6. Ligand exchange TLC enantioseparation of DL-amino acids (or commercial racemic pharmaceuticals) using Cu(II)-L-Pro complex.
- 7. Enantiomeric separation of (±)-2-butanol via its reaction with (+)-tartaric acid.
- 8. Racemization studies using (R)-3-phenyl-2-butanone or (S)-3-methyl cyclohexanone.
- 9. Synthesis of chiral variant of Sanger's reagent,



S.No.	Name of Authors/ Books/ Publishers	Year of Publication/
		Reprint
	L "Descriptive Layer Chromatography" CRC-Taylor	2006
1	Kowalska, T. and Sherma, J., "Preparative Layer Chromatography", CRC-Taylor	
	& Francis, New York.	2003
2	Ahuja, S., "Chromatography and Separation Science", Academic Press,	
	1 1	<u> </u>
3 .	Amsterdam. Snyder, L.R., Glajch, J.L., and Kirkland, J.J., "Practical HPLC Method	1700
~	Development", Wiley, New York.	



Appendix-F

INDIAN INSTITUTE OF TECHNOLOGY ROORKEE

NAME OF DEPTT/CENTRE:

MATHEMATICS DEPARTMENT

1. Subject code: MAN-901 Course Title: SELECTED TOPICS IN ANALYSIS

2. Contact Hours:

T: 0 P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

4. Relative Weightage: CWS: 25

PRS: 0 MTE: 25 ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn/Spring 7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart the knowledge of some advanced topics in Analysis.

S. No.	Contents	Contact Hours
1.	Abstract Integration: The concept of measurability, simple functions, Properties of measure, Integration of positive functions and complex functions, Sets of measure zero.	7
2.	Positive Borel Measures: Vector spaces, Review of topological preliminaries leading to locally compact Hausdorff spaces, Riesz representation theorem, Regularity properties of Borel measures, Lebesgue measures, Continuity property of measureable functions.	7
3.	L ^p spaces: Convex functions and inequalities, L ^p spaces, Approximation by continuous functions.	5
4.	Banach Space Techniques: Banach spaces, Consequences of Baire's theorem, Fourier coefficients of L ¹ functions, Hahn Banach theorem.	6
- 5.	Integration on product spaces: Measurability on Cartesian products, Product measure and its completion, Fubini's theorem, Convolution, Distribution functions.	6
6.	Harmonic Functions: Laplacian of a harmonic function, Poisson integral of L ¹ functions, Mean value property, Boundary behavior of Poisson Integrals, Representation theorems.	6
7.	Analytic Continuation: Regular and Singular Points, Continuation along curves natural boundaries, Monodromy theorem.	5
	Total	42



S. No.	Name of Books / Authors/ Publishers	Year of Publication/ Reprint
1.	W. Rudin, Real and Complex Analysis, Tata McGraw Hill, Third edition	1987
2.	H.L. Royden, Real Analysis, Collier Macmillan	1988
3.	P.R. Halmos, Measure theory, Graduate Text in Mathematics, Springer Verlag, New York	1974
4.	M. Thamban Nair, Functional Analysis, Prentice Hall, India	2003
5.	E. Kreyszig, Introductory Functional Analysis with Applications, John Wiley and Sons	1989
6.	L.V. Ahlfors, Complex Analysis, McGraw Hill	1988
7.	J.B. Conway, Functions of one complex Variables I, Narosa Publishing House	2000
8.	S. Lang, Complex Analysis, Springer - Verlag.	2003



NAME OF DEPTT/CENTRE:

MATHEMATICS DEPARTMENT

1. Subject code: MAN-902 Course Title: ADVANCED NUMERICAL ANALYSIS

2. Contact Hours:

L: 3 T: 0 P: 0

3. Examination Duration (Hrs): Theory: 3

Practical: 0

6. Relative Weightage: **CWS: 25**

PRS: 0

MTE: 25

ETE: 50 PRE: 0

7. Credits: 3

6. Semester: Autumn/Spring

7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart the knowledge of some advanced topics in Numerical Analysis.

S. No.	Contents	Contact Hours
1.	LU and QR decomposition methods for solving a system of linear equations, Conjugate Gradient iterative method and its convergence. Norms of a matrix. III-conditioned system.	6
2.	Jacobi, Given and Householder method for finding eigenvalues of a symmetric matrix. LR and QR methods for finding eigenvalues of non-symmetric matrices.	8
3.	Elliptic PDE- Finite difference analysis to solve Poisson's equation in Cartesian and Cylindrical Coordinates in axial symmetry; Hockney Method or similar method for solving resulting block tri-diagonal systems.	6
4.	Parabolic PDE - Explicit and implicit finite difference schemes for heat conduction equation. Stability, consistency and convergence of finite difference schemes; ADI methods to solve tow dimension heat conduction equation.	6
5.	Hyperbolic PDE: Propagation of errors in finite difference methods for hyperbolic equations. Method of Characteristics for Hyperbolic Equation.	6
6.	Different weighted residual methods - Galerkin, Least Square and Collocation methods; Ritz method to solve a boundary value problem. Solving BVP by taking linear elements.	10
	Total	42



S. No.	Name of Books / Authors/ Publishers	Year of Publication/ Reprint
1.	G.D. Smith, Numerical Solution of Partial Differential Equations: Finite Difference Methods 3rd Edition, Clarendon Press; 3 edition.	1985
2.	M.K. Jain, Numerical Solution of Differential Equations, Wiley Eastern Ltd.	2006
3.	L. Collatz, Numerical Treatment of Differential Equations, Springer	1962
4.	M K Jain, S. R. K. Iyengar, and R K Jain, Numerical Methods for Scientific and Engineering Computation, New Age International Publisher	2006
5.	R S Gupta, Elements of Numerical Analysis, Mcmillan Indian Ltd.	2009



NAME OF DEPTT/CENTRE: MATHEMATICS DEPARTMENT

1 Subject code: MAN-903 Course Title: THEORY OF DIFFERENTIAL EQUATIONS

2. Contact Hours: L: 3 T: 1 P: 0

3. Examination Duration (Hrs): Theory: 3 Practical: 0

4. Relative Weightage: CWS:25 PRS: 0 MTE: 25 ETE: 50 PRE: 0

5. Credits: 4 6. Semester: Autumn/Spring 7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart the knowledge of some advanced topics in Differential Equations.

S.No.	Contents	Contact Hours
1.	System of differential equations: System of first order equations, existence and uniqueness of solution, Gronwall's inequality, continuous dependence on initial conditions and parameters.	3
2.	Linear systems: Autonomous systems, Transition matrix, Phase- space of two dimensional systems, time varying systems, fundamental matrix and its properties, linear systems with periodic coefficients.	4
3.	Stability of differential systems; Stability of linear systems, almost linear systems, stability of periodic solutions, Lyapunov stability theorems for nonlinear system, limit cycles, Poincare-Bendixon theorem, Lienard Systems, Construction of Lyapunov function, Bifurcations (Transcritical, Saddle-node, Pitchfork, Hopf, Sotomayor theorem)	8
4.	Review of first order PDE: classification, solution method for quasi-linear and nonlinear pde, discontinuous solutions, conservation laws and shocks	3
5.	Four important linear PDE's (transport, Lapace, heat and wave equations): fundamental solution, meanvalue formulae, properties of harmonic functions, Green's function and energy method.	6
6.	Sobolev spaces: Definition, approximations, sobolev inequalities, extensions, traces, compactness, dual spaces	8
7.	Elliptic Equations: Definitions, Existence Of Weak Solutions, Regularity, Maximum principles, Eignevalues and eigen-function. Linear evolution equations: Parabolic equations, hyperbolic equations, semigroup theory	10
	Total	42



S.No.	Name of Books / Authors/ Publishers	Year of Publication/ Reprint
1.	G.F. Simmons, Differential Equations with Applications and Historical Notes, 2nd Ed., McGraw-Hill.	1991
2.	R.P. Agarwal, D.O'Regan, An Introduction to Ordinary Differential Equations, Springer.	2008
3.	K.S. Bhamra, Ordinary Differential Equations, Narosa Publications.	2015
4.	I.N. Sneddon, Elements of Partial Differential Equations, McGraw-Hill.	1957
5.	L.C. Evans, Partial Differential Equations, 2nd Ed., American Mathematical Society.	2015
6.	M. Renardy, R.C. Rogers, An Introduction to Partial Differential Equations, 2nd Ed., Springer.	2010
7.	S. Kesavan, Topics In Functional Analysis And Its Applications, New Age International(P) Ltd.	2012



NAME OF DEPTT/CENTRE:

MATHEMATICS DEPARTMENT

Subject code: MAN-904 Course Title: SELECTED TOPICS IN ALGEBRA

2. Contact Hours:

L: 3 T: 0 P: 0

3. Examination Duration (Hrs): Theory: 3 Practical: 0

Relative Weightage: CWS: 25 PRS: 0

MTE: 25

ETE: 50

PRE: 0

5. Credits: 3

6. Semester: Autumn/Spring 7. Subject Area: PEC

8. Pre-requisite: Nil

9. Objective: To impart the knowledge of some advanced topics in Algebra.

S. No.	Contents	Contact Hours
1.	Review of group actions and Sylow's theorems. Free groups and relations, normal series, nilpotent and solvable groups.	8
2.	Review of rings and ideals, PID, Euclidean domains, and UFD. Modules, direct sums of modules, free modules, exact sequences, finitely generated modules over a PID, structure of finitely generated abelian groups, rational and Jordan canonical forms.	10
3.	Review of algebraic extensions of fields, algebraic closure and splitting fields. Normal extensions and separable extensions, finite fields, Galois theory - The Fundamental Theorem of Galois Theory, roots of unity, cyclotomic extensions, cyclic extensions, Galois group of a polynomial, solvable and radical extensions, insolvability of the quintic.	11
4.	Artinian and Noetherian modules and rings, modules of finite lengths, simple and semisimple modules and rings. Wedderburn-Artin theorem, nil radical and Jacobson radical, radical of an Artinian ring.	10
5.	Commutative rings: Primary decompositions of ideals and modules.	3
	Total	42



S. No.	Name of Books / Authors/ Publishers	Year of Publication/ Reprint
1.	Dummit D.S. and Foote R.M., "Abstract Algebra", John Wiley & Sons (3rd Edition)	2003
2.	Hungerford T.W., "Algebra", Springer	1980
3.	Bhattacharya P.B., Jain S.K. and Nagpaul S.R., "Basic Abstract Algebra", Cambridge University Press (2nd Edition)	1995
4.	Lang S., "Algebra", Springer (3rd Edition)	2005
5.	Jacobson N., "Basic Algebra Vol. I & Vol. II" Dover Publications (2nd Edition)	2009
6.	Musuli C., "Introduction to Rings and Modules", Narosa Publishing House (2nd Edition)	1997



NAME OF DEPT/CENTRE:

ELECTRONICS AND COMMUNICATION ENGINEERING

1. Subject Code: ECN - 578

Course Title: Digital System Design

2. Contact Hours:

L: 3

T: 0

P: 0

3. Examination Duration (Hrs.): Theory: 3

Practical: 0

4. Relative Weight:

CWS: 25

PRS: 0

MTE: 25

ETE: 50 PRE: 0

5. Credits: 3

6. Semester: Autumn

7. Subject Area: PCC

8. Pre-requisite: Digital Logic Design or Equivalent

9. Objective: To acquaint with the hardware description languages such as VHDL/Verilog for understanding the principles for designing digital and embedded systems.

SI.No.	Contents	Contact Hours
1.	Introduction to Digital and Embedded systems design: Digital Design Using ROMs, PLAs and PLAs, BCD Adder, 32 - bit adder, A shift and add multiplier, Array multiplier, and Binary divider. Introduction to Embedded system, Design cycle in the development phase for an embedded system, Use_ of target system or its emulator nd In-circuit emulator, Use of software tools for development of an ES.	4
2.	Hardware Description Languages (HDL): Digital system Design Process, Hardware Description Languages, Hardware Simulation, Hardware Synthesis, Levels of Abstraction, Characterizing Hardware Languages, Objects and Classes, Signal Assignments, Concurrent and Sequential Assignments.	6
3.	Design Organization and Parameterization: Definition and usage of Subprograms, Packaging Parts and Utilities, Design Parameterization, Design Configuration, Design Libraries. Type Declarations and Usage, Operators, Subprogram Parameter Types and Overloading, Other Types and Types Related Issues, Predefined Attributes, User Defined Attributes. Dataflow Description: Multiplexing and Data Selection, State Machine Description, Three State Bussing. Behavioral Description of Hardware: Process Statement, Assertion Statement, Sequential Wait Statements, Formatted ASCII I/O Operations, IC Design Flow.	8
4.	Practical Designs EPGA Architecture: Designing and Implementation of Finite State Machines for FPGA; Synthesis Techniques nd Timinig Analysis; Placement and Routing; Embedded Hardware and Software Design with FPGA.	8

5.	DSP Processor Architecture: Architecture; Functional Units; Fetch and Execute Packets; Pipelining; Registers; Linear and circular Addressing Modes; Instruction Set Assembler Directives for TMS320C6x or ADSP21xx; Linear Assembly; ASM statement within C; C-Callable Assembly Function; Timers; interrupts; Multichannel Buffered Serial Ports; Direct Memory Access; Memory Considerations; Fixed and Floating Point Format Code Improvement; Constraints	
	Programming Examples Using : C, Assembly, and Linear Assembly.	
6.	ARM Architecture and Organization: ARM Assembly Programming; THUMB Assembly Programming; ARM-THUMB Interworking; Assembly and C Mixed Programming; Exception Handling; ARM Tool chain (Assemblers, Compilers, Linkers & Debuggers); Firmware Programming; Cache & MMU; Peripheral Programming; ARM Cortex family of Processors and architecture; Operating modes, Registers and Memory Map of Cortex-M3; Embedded OS; Porting of Embedded OS on ARM.	8

SI.No.	Name of Books/Authors	Year of Publication
1.	Embedded System Design: Embedded System Foundations of Cyber-Physical Systems by Peter Marwedel, Springer	2010
2.	Embedded System Design: A Unified Hardware/Software introduction by Frank Vahid, Tony Givargis, John Wiley & Sons, Inc.	2001
3.	Fundamental of Logic Design - Charles H. Roth, and Larry L. Kinney, Brooks/Cole Inc.	2014
4.	Digital Logic and Microprocessor Design with VHDL, Enoch O. Hwang, Publisher- Thomson/Nelson	2006
5.	Digital Design and Computer Architecture, David Money Harris and Sarah L. Harris, Elsevier.	2012
6.	VHDL for Programming Logic, Kevin Skahill, Person Education	2004
7.	ARM System-on-Chip Architecture, Furber, S., 2nd ed. Pearson Education.	2000
8.	DSP Applications Using C and the TMS320C6x DSK, Rulph Chassaing, John Wiley & Sons, Inc.	2002





F. No. 24-2/2016 TS 1
Government of India
Ministry of Human Resource Development
Department of Higher Education

New Delhi, dated 8th April 2016

To The Directors of all IITs

Subject: Revision of tuition fee for the undergraduate programmes

Sir,

With a view to further strengthen the IITs, Chairperson of IIT Council, after examining the report of the Committee on funding of IITs and the recommendations of the Standing Committee of the IIT Council (SCIC) for revision of tuition fee in IITs, has approved revision of tuition fee in the IITs to Rs. 2 lakh a year from the academic year 2016-17, subject to the following for protecting the interests of the socially and economically backward students:

a) The SC/ST/PH students shall get complete fee waiver.

b) The most economically backward students (whose family income is less than Rs. 1 lakh per annum) shall get full remission of the fee.

c) The other economically backward students (whose family income is between Rs. 1 lakh to Rs. 5 lakh per annum) shall get remission of 2/3rd of the fee.

d) All students shall have access to interest free loan under the Vidyalaxmi scheme for the total portion of the tuition fee payable.

- 2. Further, all IITs are requested to utilize funds accrued through the increased student fee for the purpose of building good quality infrastructure with the assistance of the Higher Education Funding Agency (HEFA).
- The Directors of all IITs are requested to take action accordingly.

Yours faithfully,

(R. Subrahmanyam) Additional Secretary (TE)